

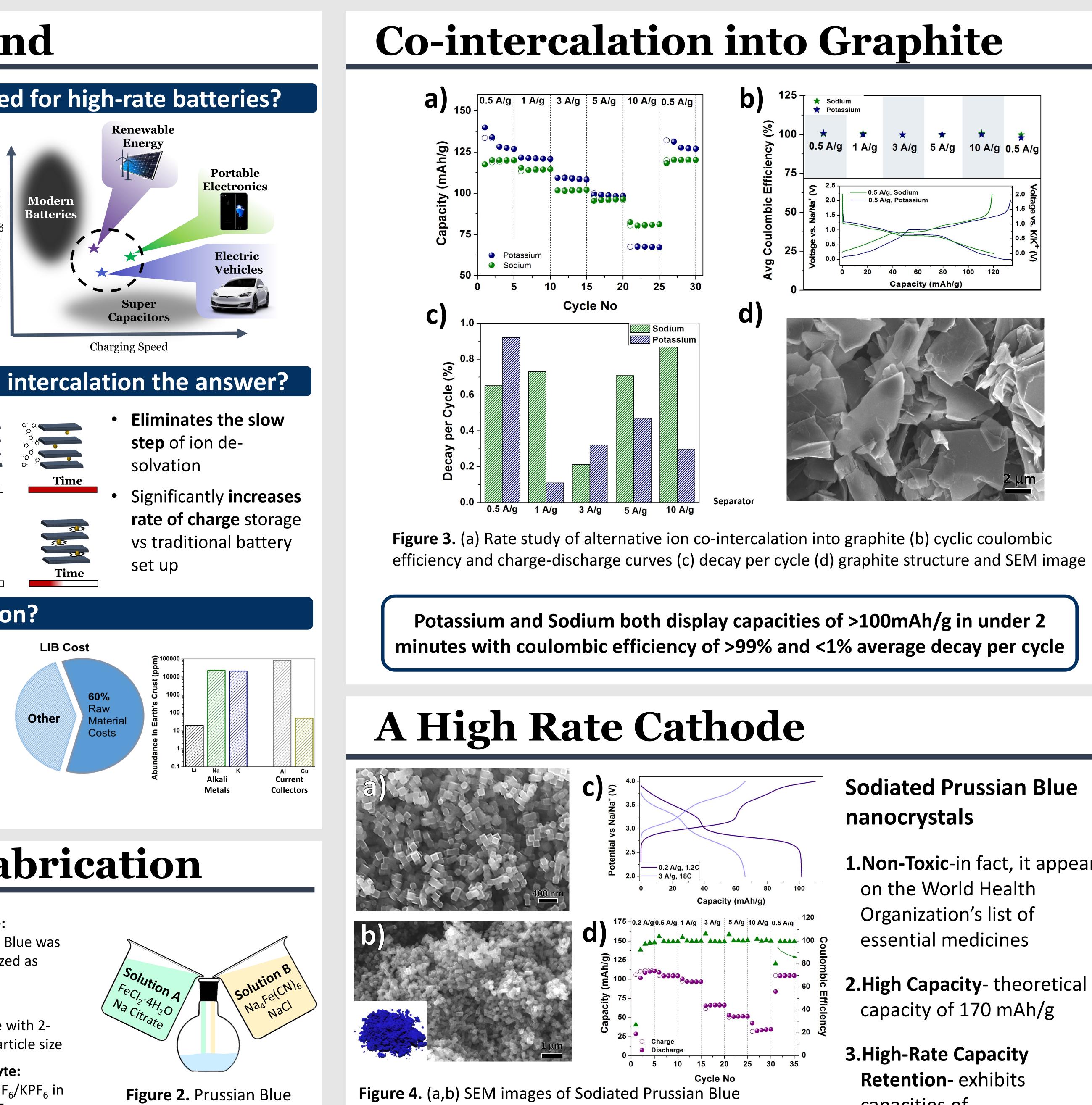
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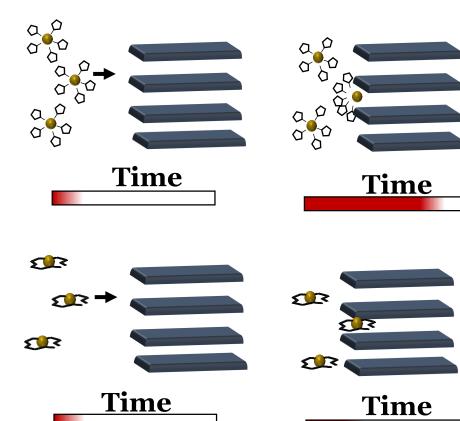
Background

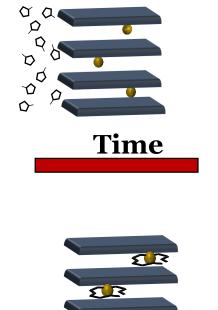
Why is there a need for high-rate batteries?

 Modern energy storage cannot meet demand for affordable technology with high power & ultrafast charging



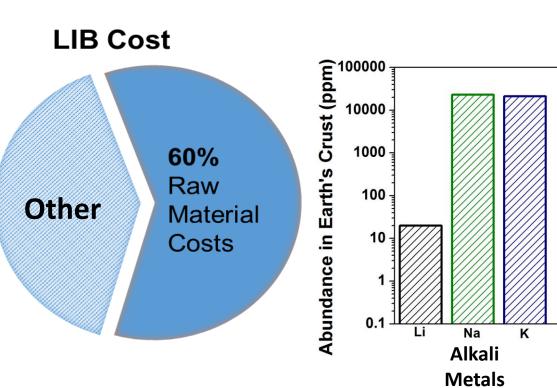
Why is co-solvate intercalation the answer?





Why alternative-ion?

- Na and K are 1000 times more abundant than Li in the earth's crust
- Na and K ion batteries utilize much cheaper current collectors than Li



Battery Fabrication

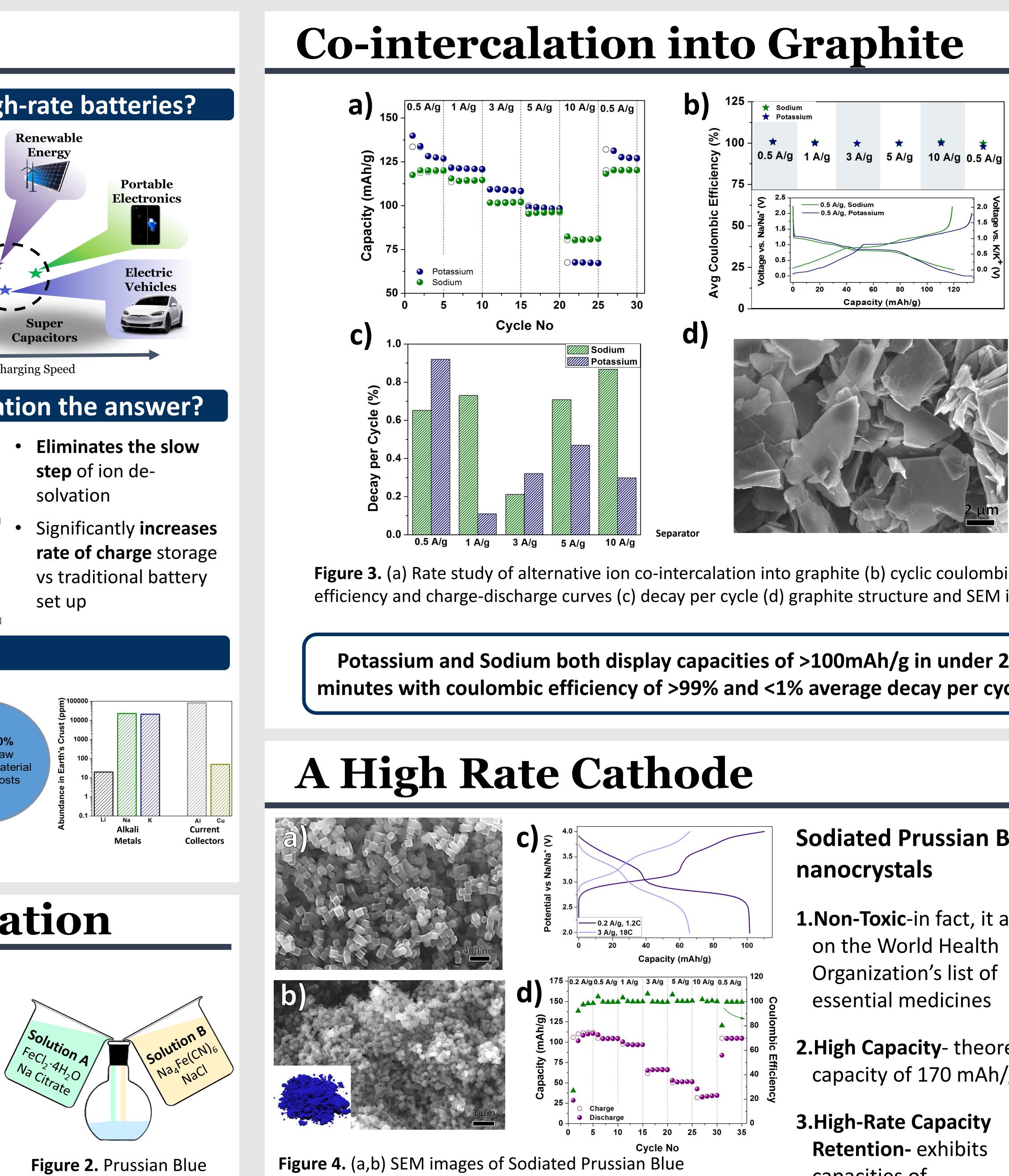
Anode **Separator** Cathode Spacer Spring Cap J. Mater. Chem. A, 2014,2, 4627-4632

Figure 1. Coin cell configuration

Cathode: Prussian Blue was synthesized as shown

Anode: Graphite with 2-15µm particle size

Electrolyte: $1 \text{ M NaPF}_6/\text{KPF}_6$ in DEGDME



synthesis schematic



High-power alternative-ion batteries via co-intercalation

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nanocrystals, (c) charge-discharge curves of cathode at varied rates (d) rate study of Prussian Blue

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- **1.Non-Toxic**-in fact, it appears
- **2.High Capacity** theoretical capacity of 170 mAh/g
- capacities of ~100 mAh/g at 7C

Conclusions

- than 5 minutes
- electric vehicles

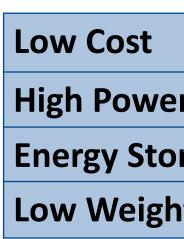


Figure 5. Qualitative comparison of most common energy storage devices versus designed performance of full cell alternative ion cointercalation devices

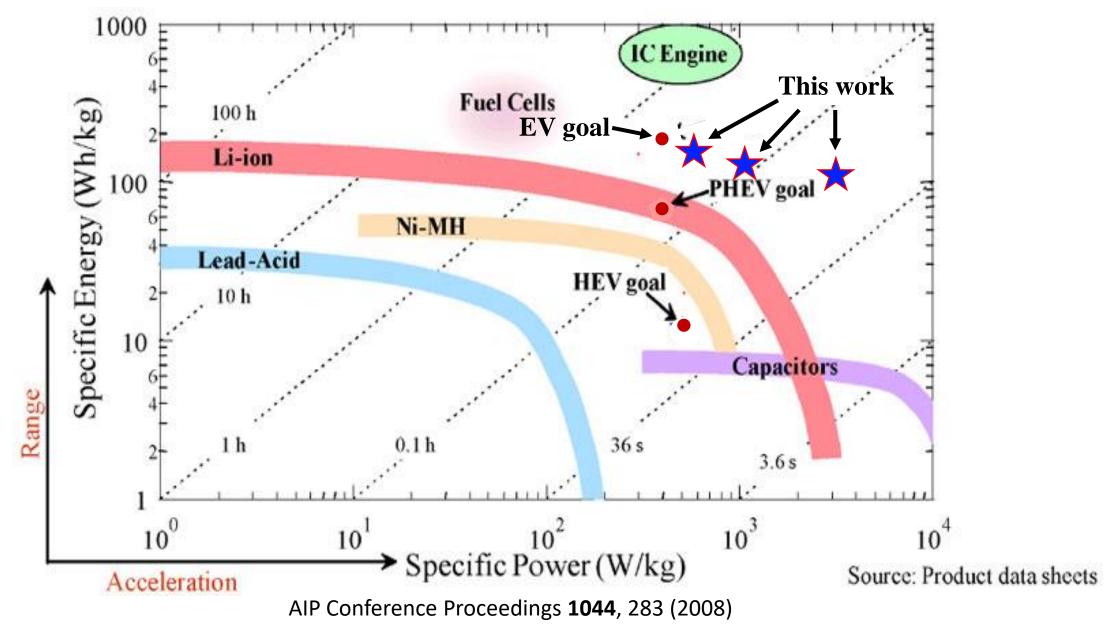
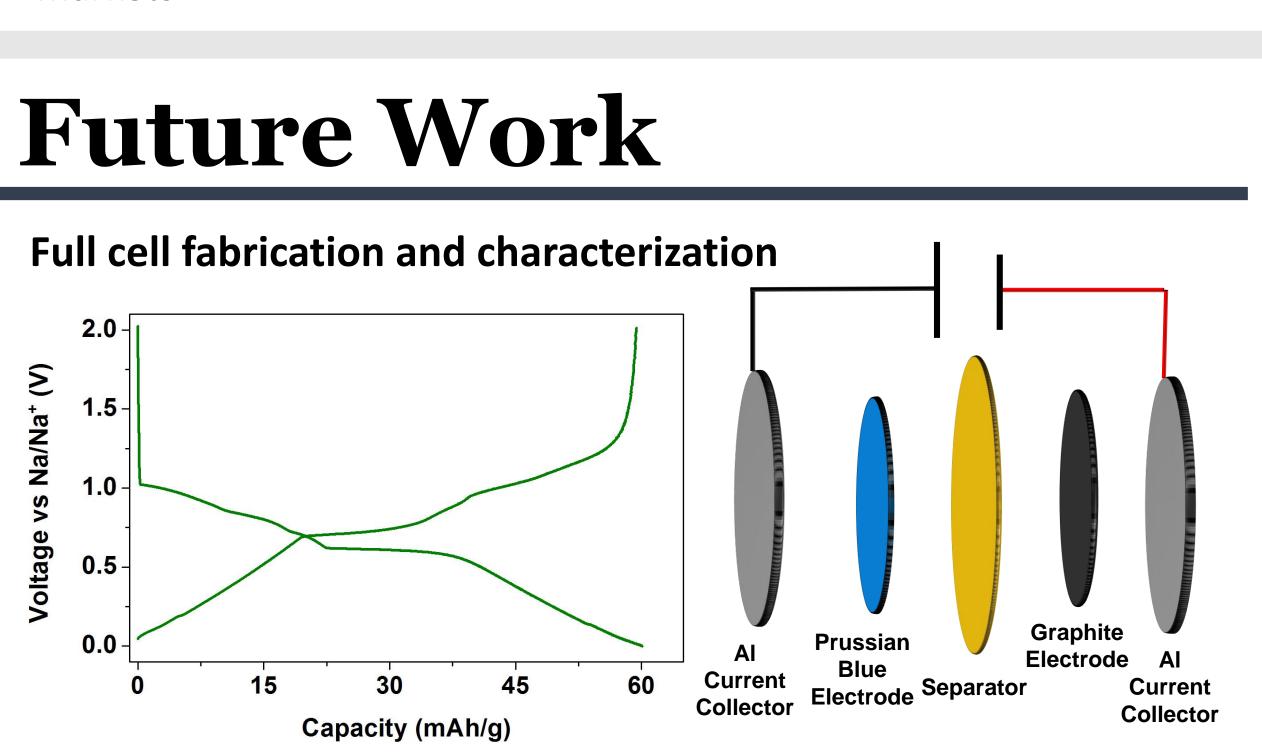
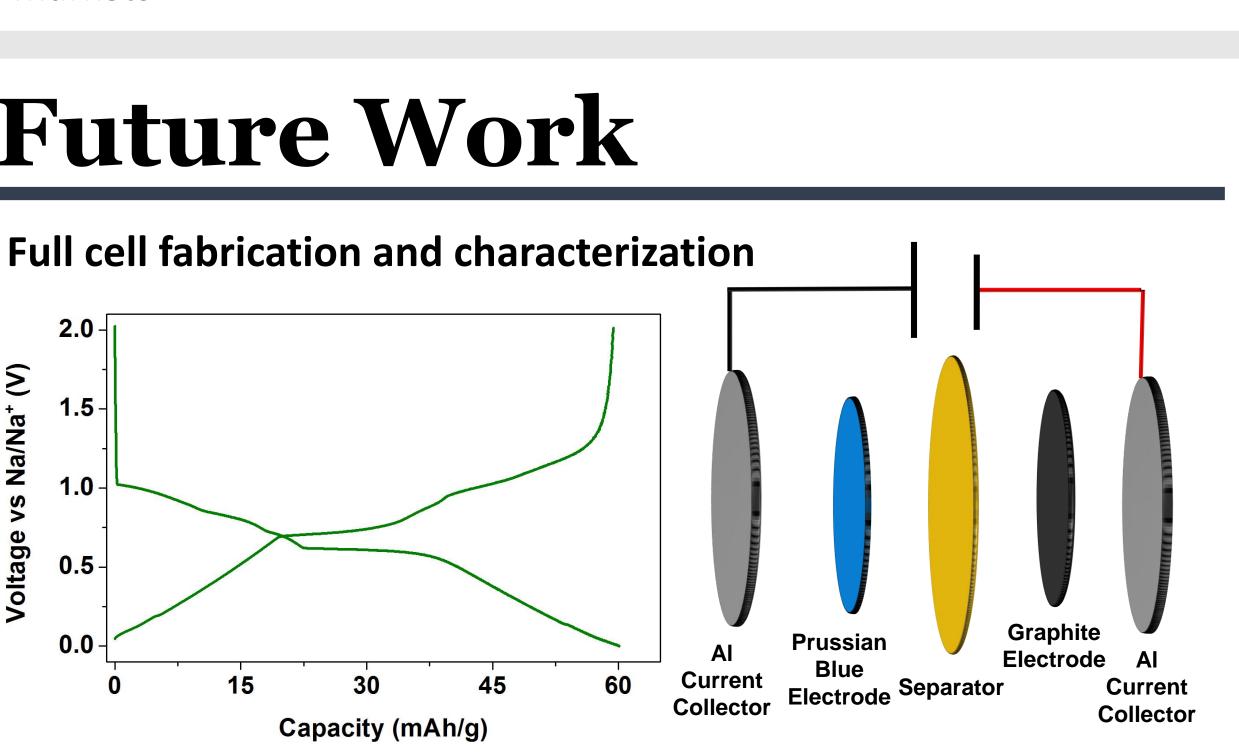


Figure 6. Ragone plot of modern energy storage methods, designed performance of this work, and the demands of emerging vehicle markets





Alternative-ion graphite co-intercalation enables charge >100 mAh/g in less than 2 minutes

Sodiated Prussian Blue exhibits **discharge >95 mAh/g in less**

A full device based on the co-intercalation phenomenon is designed to meet requirements for implementation in

	Super- Capacitors	Modern Batteries	Alternative-ion Co-Intercalation
			Devices
er			
rage			
nt			

