

Imaging Hidden Corrosion with SQUID Magnetometry

John P. Wikswo and Yu Pei Ma

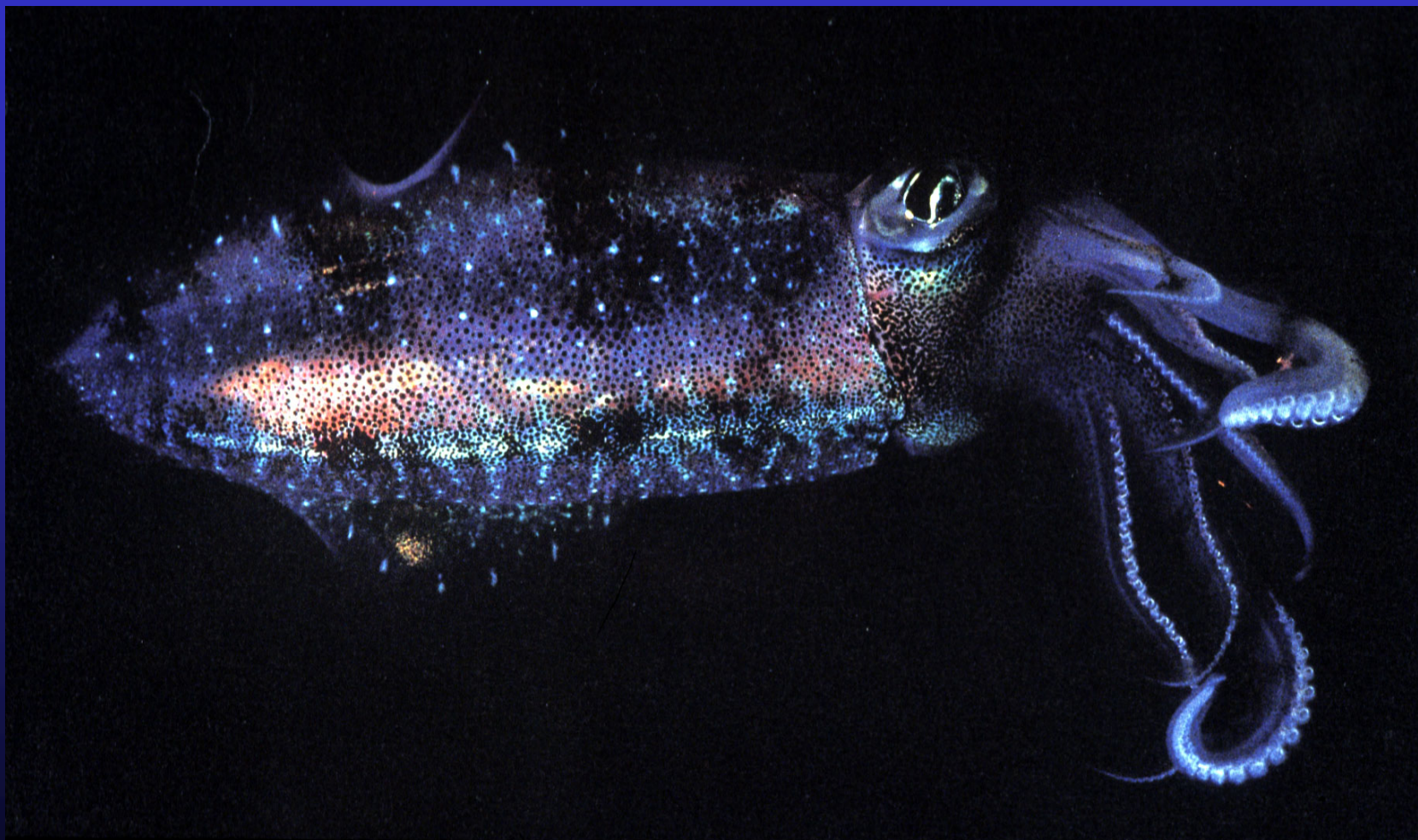
Vanderbilt University

Aqueous Corrosion Gordon Conference

New London, NH, 17 July 2002



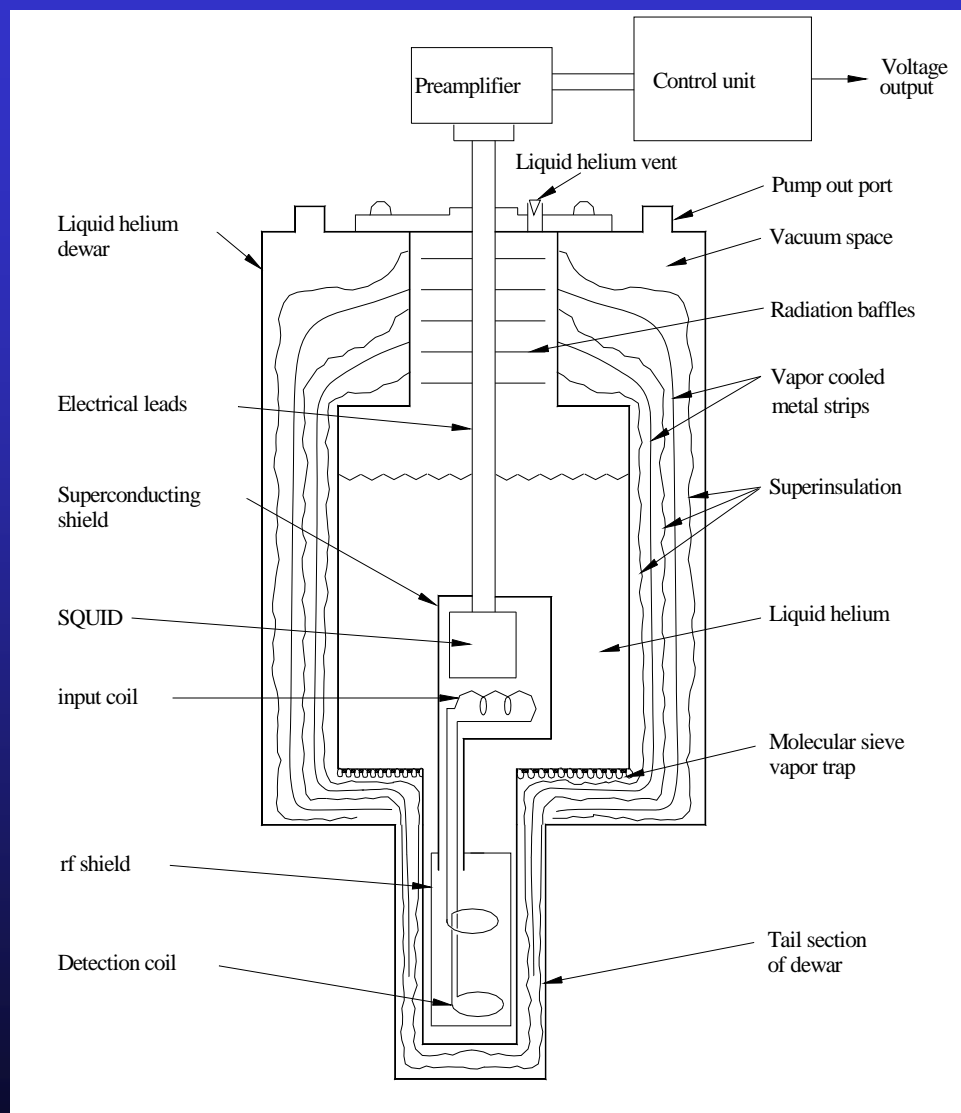
What is a SQUID magnetometer?





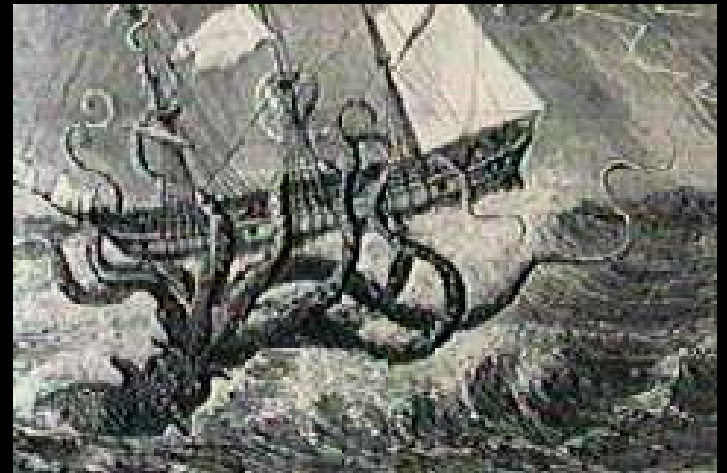
Superconducting QUantum Interference Device (SQUID) Magnetometer

- Pickup coil coupled to a SQUID that measures the current induced in the pickup coil.
- A flux-to-voltage converter with unrivaled **sensitivity** ($5\text{-}20 \text{ f T/Hz}^{1/2}$)
- **Spatial resolution:** 1 to 3 mm (20 μm max)
- **Bandwidth** of DC to 10's kHz.





SQUIDS for Naval NDE and DE



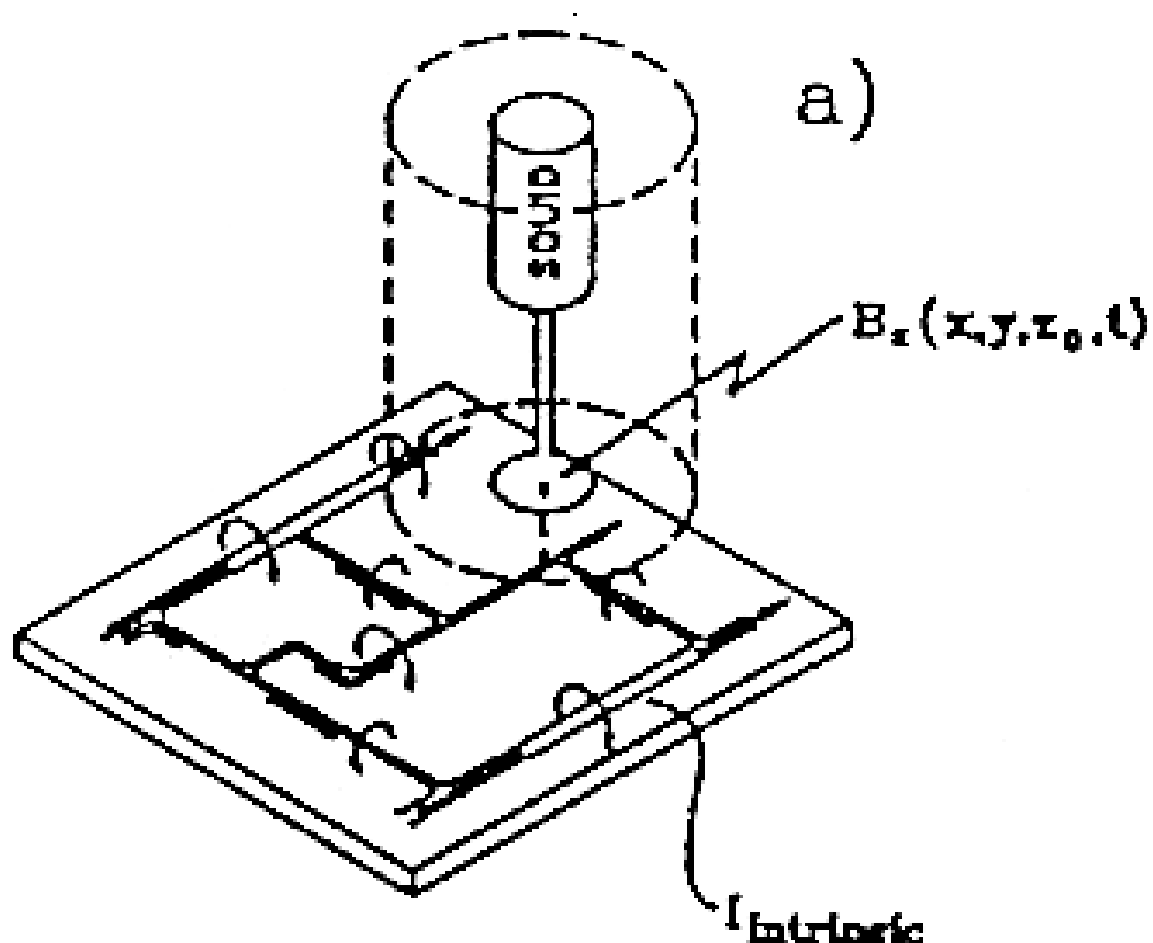
<http://unmuseum.mus.pa.us/kraken.htm>

<http://www.angelfire.com/anime3/kraken/>



SQUIDs for NDE, etc.

- Intrinsic currents
- Remanant fields
- Applied currents
- Thermal currents
- Eddy currents
- Ferromagnetic and stress-related magnetization
- Diamagnetic and paramagnetic susceptibility



This airplane is supposed to fly when it is older than anyone in this room....



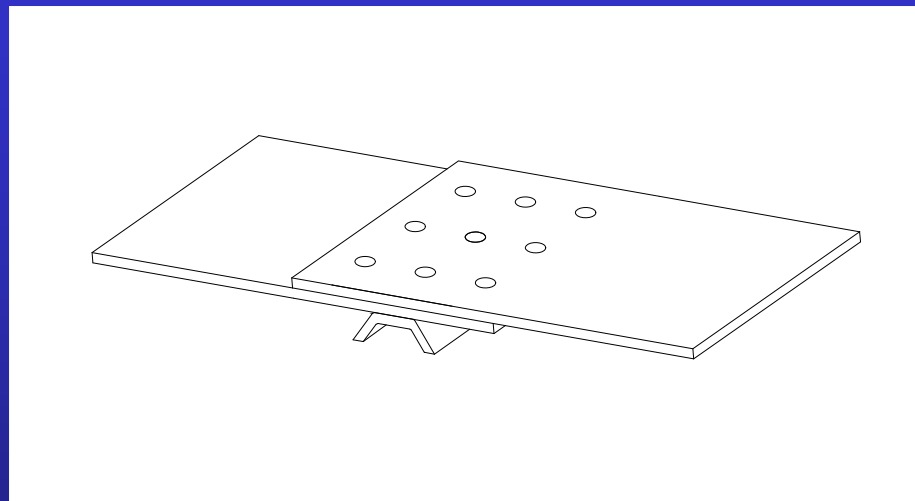
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<http://sill-www.army.mil/Graphics/aircraft/Page0002.html> kc1351.jpg

http://www.emeraldpacificcorp.com/emeraldevents/airshow/images/aircraft/pages/kc135-1_jpg.htm kc135_good.jpg



Aging Aircraft Corrosion Measurements Using SQUIDs



John Wikswo, Afshin Abedi, Grant Skennerton,
Yu Pei Ma, Delin Li

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How do you quantify hidden corrosion?

- NDI – detects corrosion damage and missing metal
 - Measurable material loss may take months
- Mass Loss – detects metal loss by weighing
 - Well-suited for determining the average rate over intervals as short as several weeks
 - Cannot be used on old lap joints or exfoliation/IGA
- Potentiometric measurements
 - Limited to exposed surfaces



Practical Corrosion Questions

What is the instantaneous rate of hidden corrosion?
How does it depend upon?

- Humidity
- Environment (salt, *etc.*)
- Corrosion abatement technology
- Maintenance
- Metallurgy

SQUIDs can help answer these questions.



1 cm² KC-135
4 mils of metal lost
10⁻² cm³

2.7 gm/cm³
27 gm/mole

10⁻³ mole $\Rightarrow I =$
 $\frac{3 + 6 \times 10^{23} \times 10^{-3} + 1.6 \times 10^{-19}}{80 \text{ years}} = 100 \text{ nA}$

3 × 10⁷ s/yr

$B = \frac{\mu_0 I}{2\pi r}$
= 20 pT
200:1 SNR ← !



Why use a SQUID magnetometer?

- There are no established techniques that can measure the rate of hidden corrosion
- There is little knowledge of how corrosion rates are affected by environment, structural condition, flight history, or maintenance procedures.
- Standard electrochemical techniques cannot study the instantaneous rate or distribution of hidden or exfoliation corrosion.
- SQUIDs are ideally suited to map the distribution of **hidden** corrosion ACTIVITY in an aircraft lap joint or wing plank
- Caution: The mechanisms by which corrosion activity produces the observed magnetic fields are not fully understood

How do you quantify hidden corrosion?



<http://www.tenhand.com/squid/bluesquid.jpg>

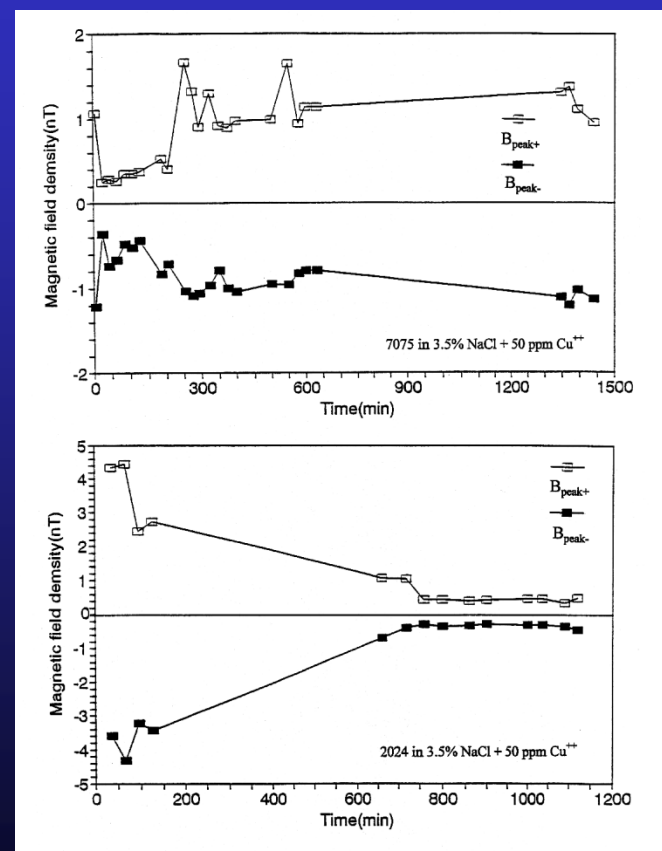
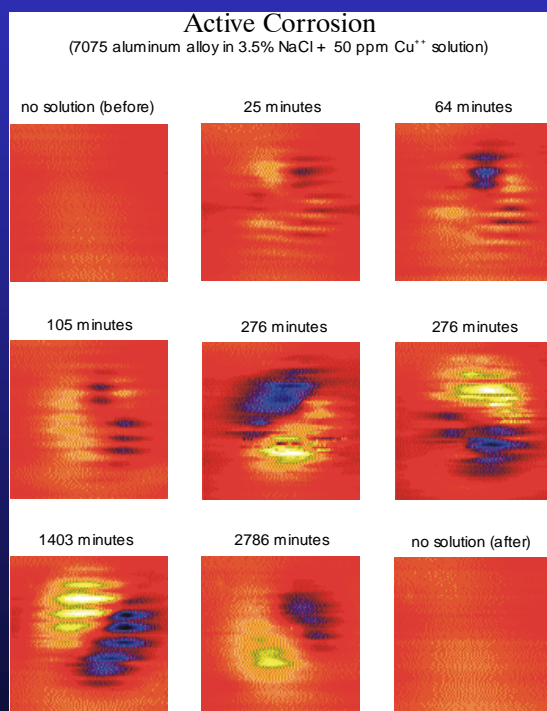
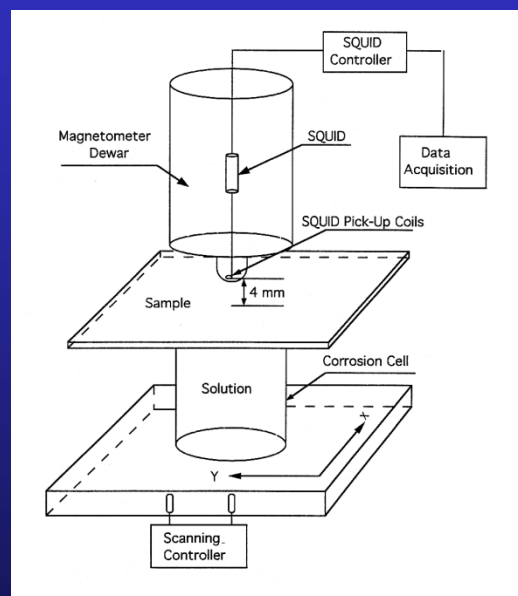


How do you quantify hidden corrosion?

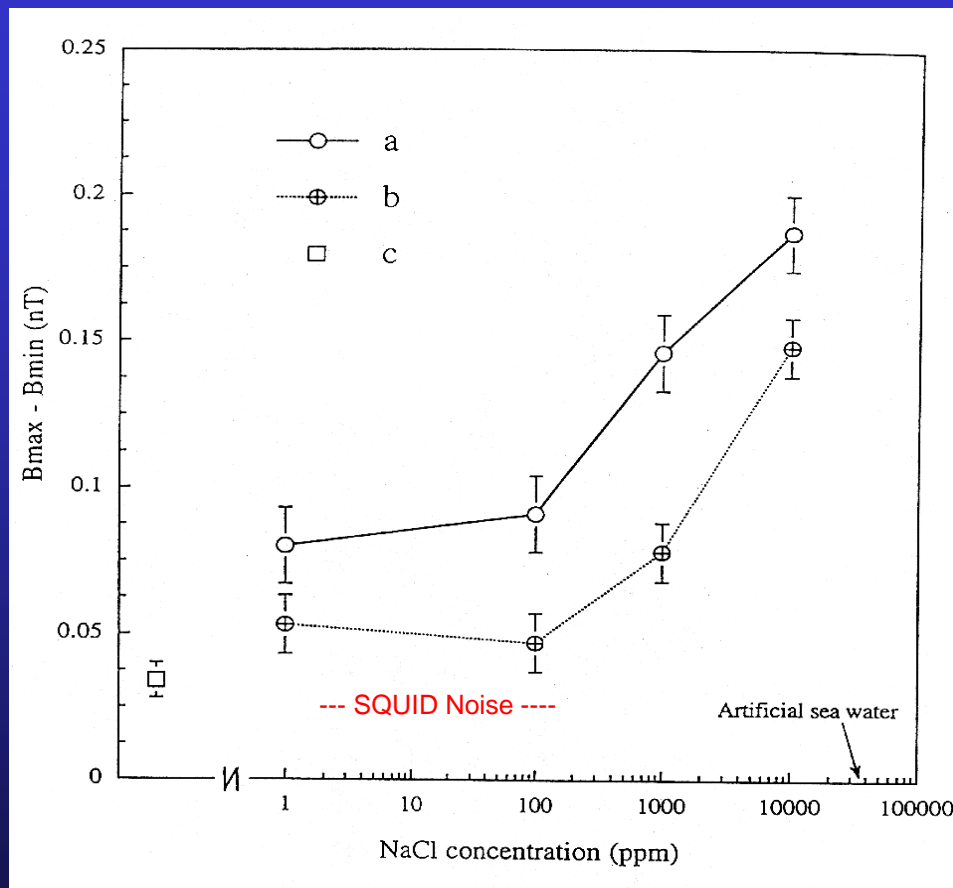
- NDI – detects corrosion damage and missing metal
 - Measurable material loss may take months
- Mass Loss – detects metal loss by weighing
 - Well-suited for determining the average rate over intervals as short as several weeks
 - Cannot be used on old lap joints or exfoliation/IGA
- Potentiometric measurements
 - Limited to exposed surfaces
- SQUIDs – detects magnetic field of corrosion currents
 - Can detect *instantaneous* corrosion
 - Difficult to obtain absolute calibration



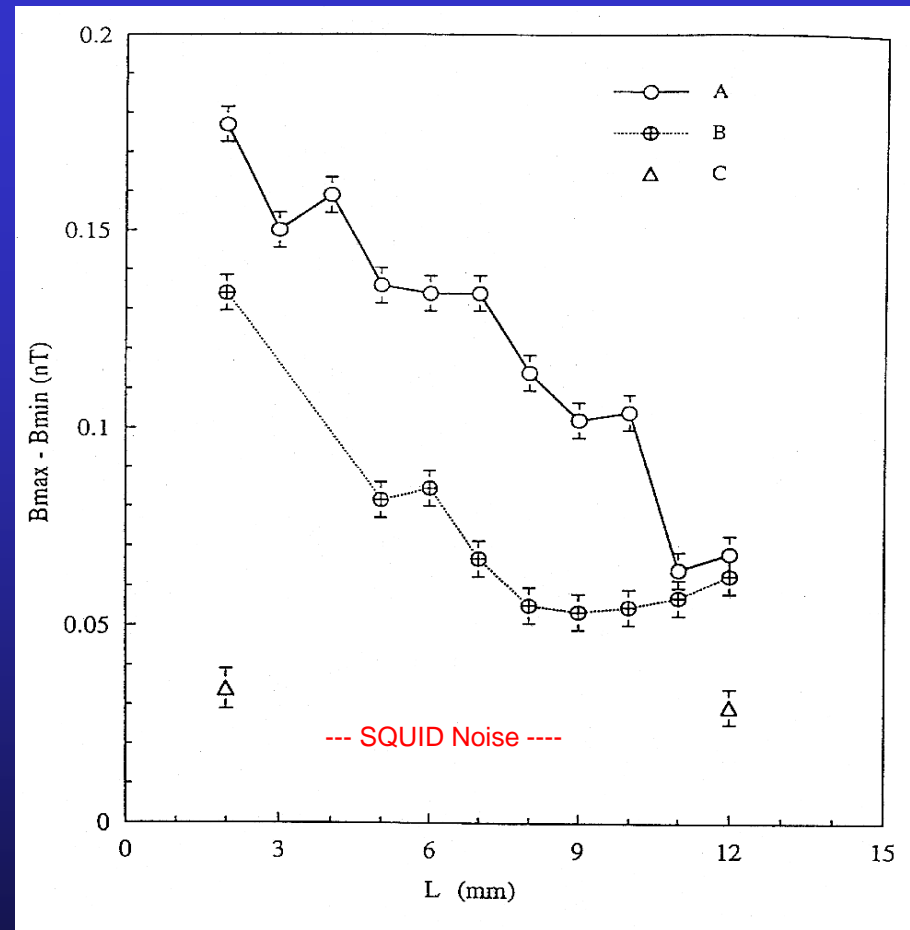
SQUID Corrosion Measurements: 1990-1995



SQUID Sensitivity for Weak or Deep Corrosion



SQUIDs can detect corrosion of aluminum in 1 ppm NaCl



SQUIDs can detect corrosion of aluminum through 1 cm of metal or air

Detection of Hidden Corrosion of Aircraft Aluminum Alloys by Magnetometry Using a Superconducting Quantum Interference Device," D. Li, Y.P. Ma, W.F. Flanagan, B.D. Lichter, and J.P. Wikswo, Jr., Corrosion, 53(2): 93-98 (1997).



Conclusions from AFOSR-URI studies

- SQUIDs are suited for the periodic, non-destructive analysis of corrosion test specimens where the corrosion activity is not directly accessible to a potentiostat, *e.g.*, corrosion that is hidden under a thick coating or one or more layers of metal.
- SQUIDs may be the only technique to detect these hidden currents non-destructively and instantaneously.
- **The external magnetic field does not reflect all of the internal corrosion activity, *i.e.*, there are field cancellation effects.**



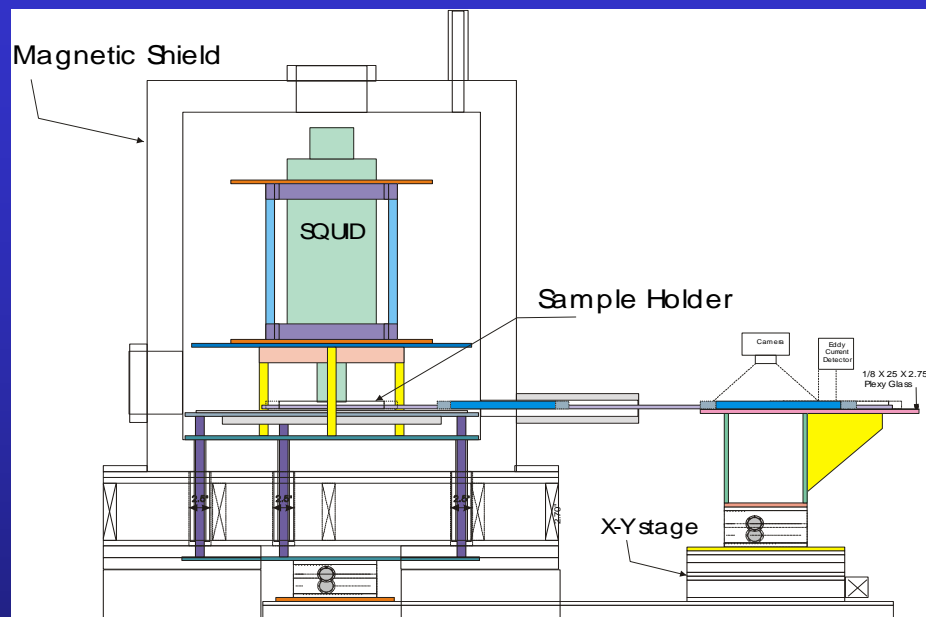
SQUID Corrosion Measurements in the Laboratory

- This is a laboratory technique for determining the rates of hidden corrosion under different conditions.
- This is **NOT** an NDI tool!
- It is highly unlikely that this technique can be applied to intact aircraft on the flight line!





The AFCO Corrosion SQUID System





SQUID-Specific Questions

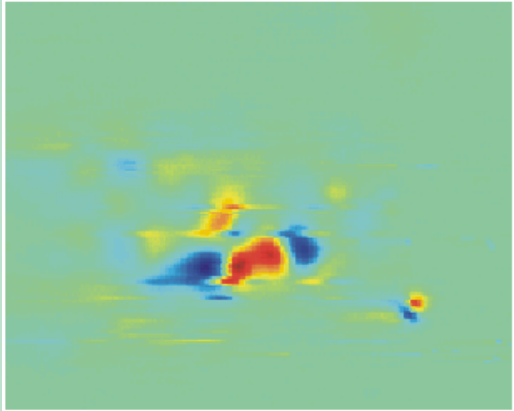
How do the SQUID data correlate with the instantaneous rate of corrosion?

How do the SQUID data correlate with total mass loss?

How do the SQUID images correlate with corrosion damage images?

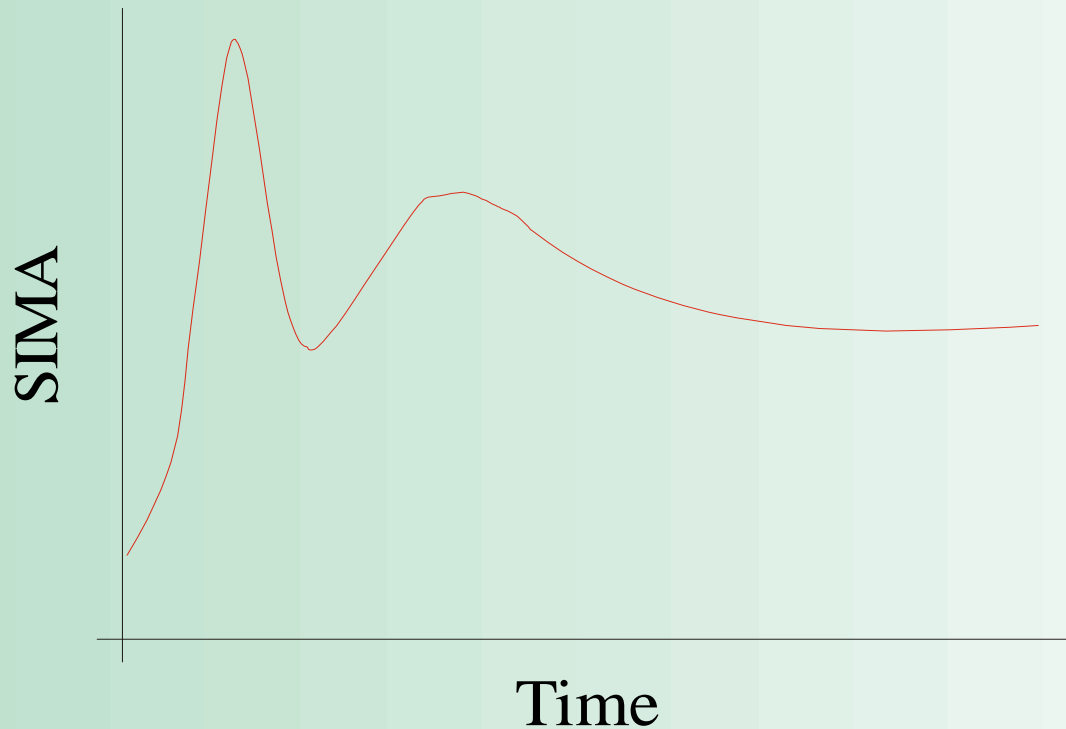
How do the SQUID data correlate with the instantaneous rate of corrosion?

Start with the spatially-integrated magnetic activity (SIMA)

$$\int_{XY} \left| \text{Magnetic Activity} \right| \Delta X \Delta Y = \text{SIMA}$$


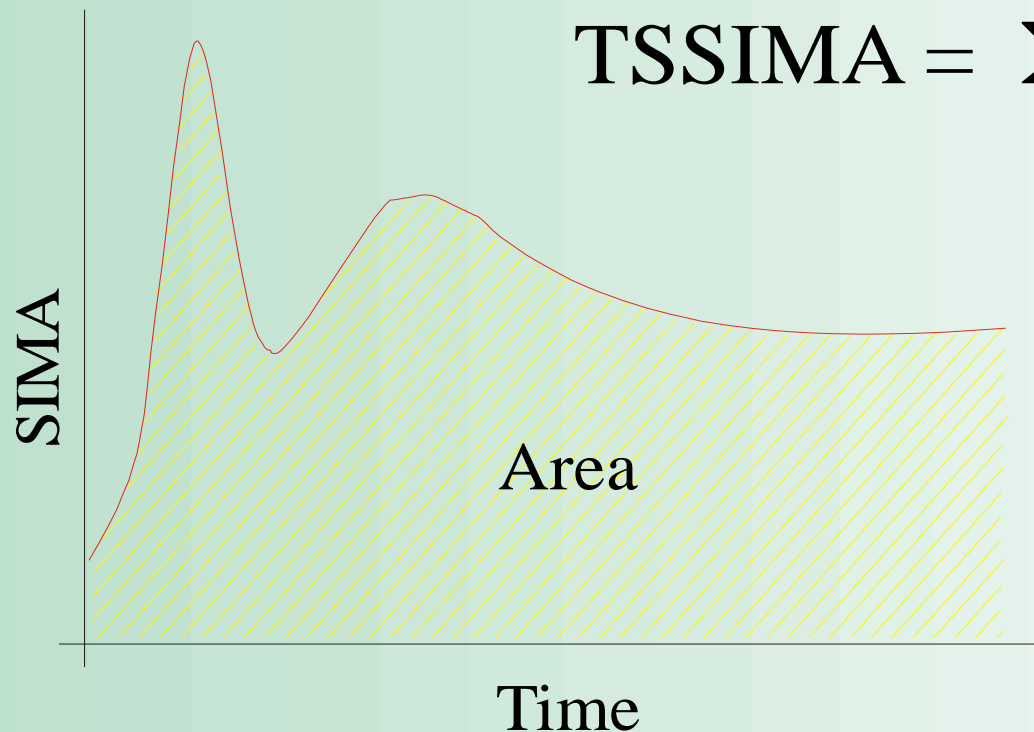
$$\text{SIMA}(t_j) = \sum_{XY} |B(x, y, t_j)| \Delta X \Delta Y$$

Ideally, SIMA is proportional to the instantaneous corrosion activity, i.e. corrosion rate



How do the SQUID data correlate with mass loss?

Use the temporally-summed spatially-integrated magnetic activity (TSSIMA)



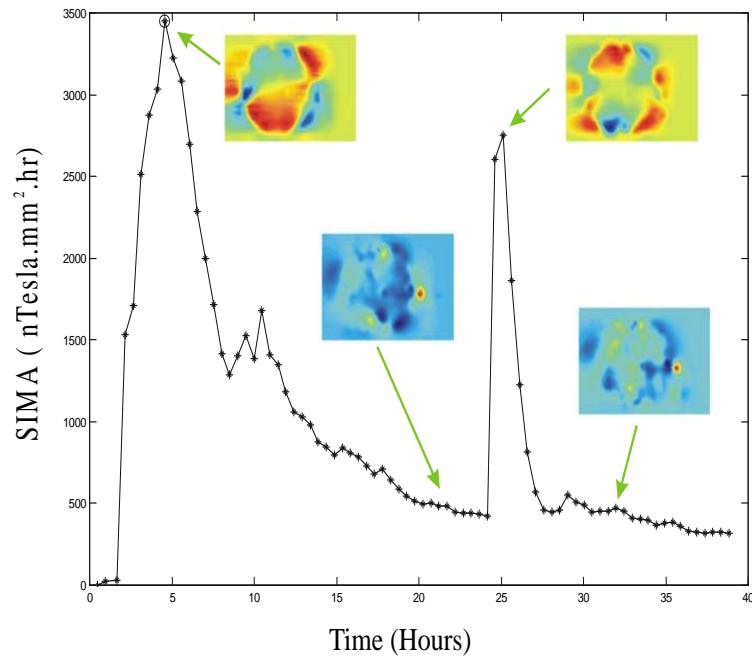
$$\text{TSSIMA} = \sum_j \text{SIMA}(t_j) \Delta t_j$$

TSSIMA correlates well with mass loss, but with a geometry-dependent calibration factor

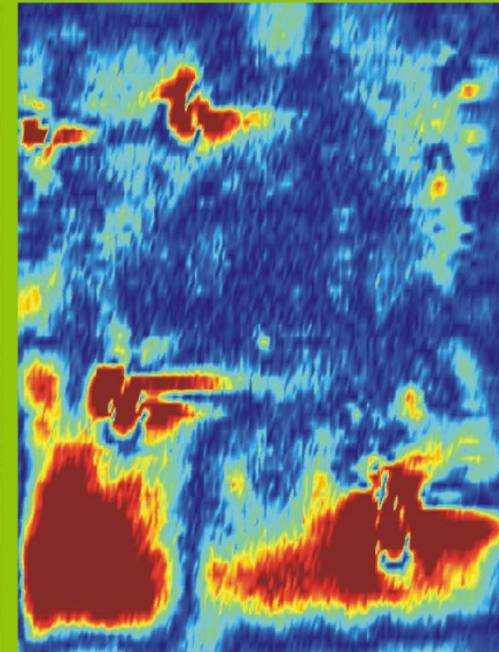
SIMA and TSMA with the UVa LJSS cocktail

SIMA

SIMA versus time for 7075-T6, 0.5M NaOH



TSMA



Old Lap Joints





Corrosion Rates in Old Lap Joints

Protocol 3 exposure sequence

Step 1: Humid Air (98% RH)

Step 2: Distilled Water

Step 3: 0.01 M Chloride

Step 4: 0.1 M Chloride

- Bake-out before each step
- Degauss after each bake-out
- Each step is repeated three times for all specimens



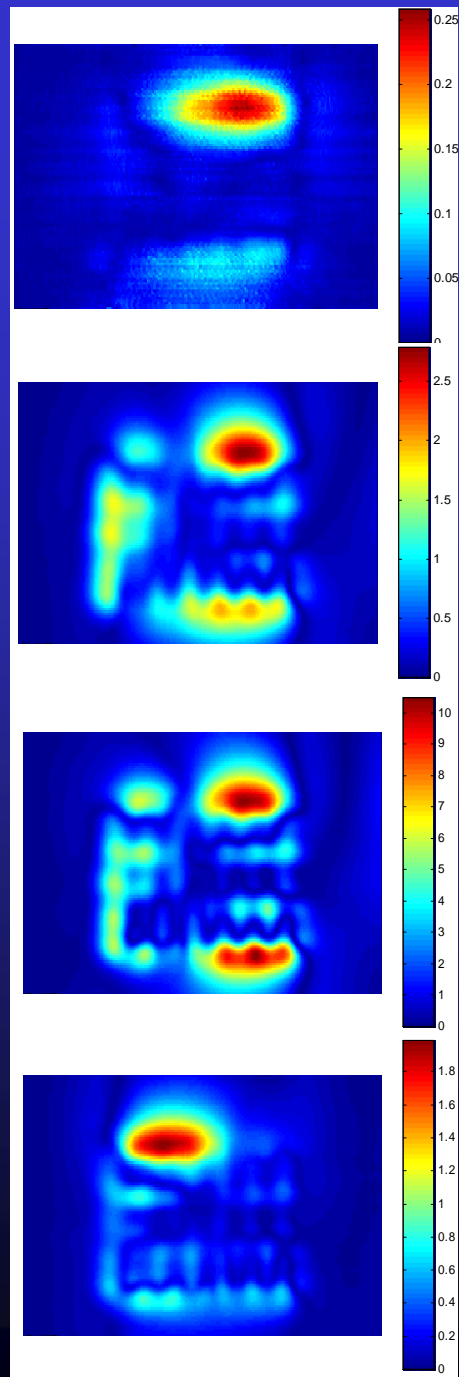
SQUID Images

Step 1: Humid Air (98% RH)

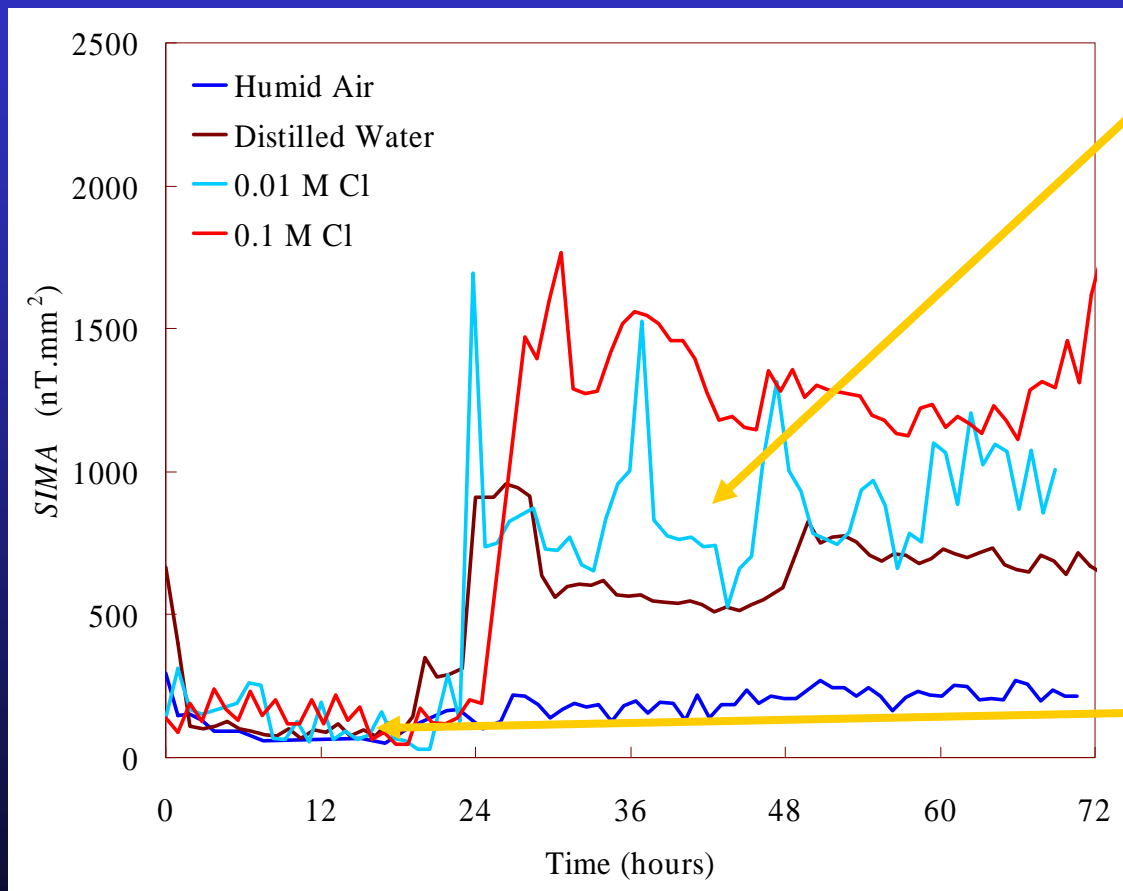
Step 2: Distilled Water

Step 3: 0.01 M Chloride

Step 4: 0.1 M Chloride



Summed Magnetic Activity Versus Time for Old Aircraft Lap Joints

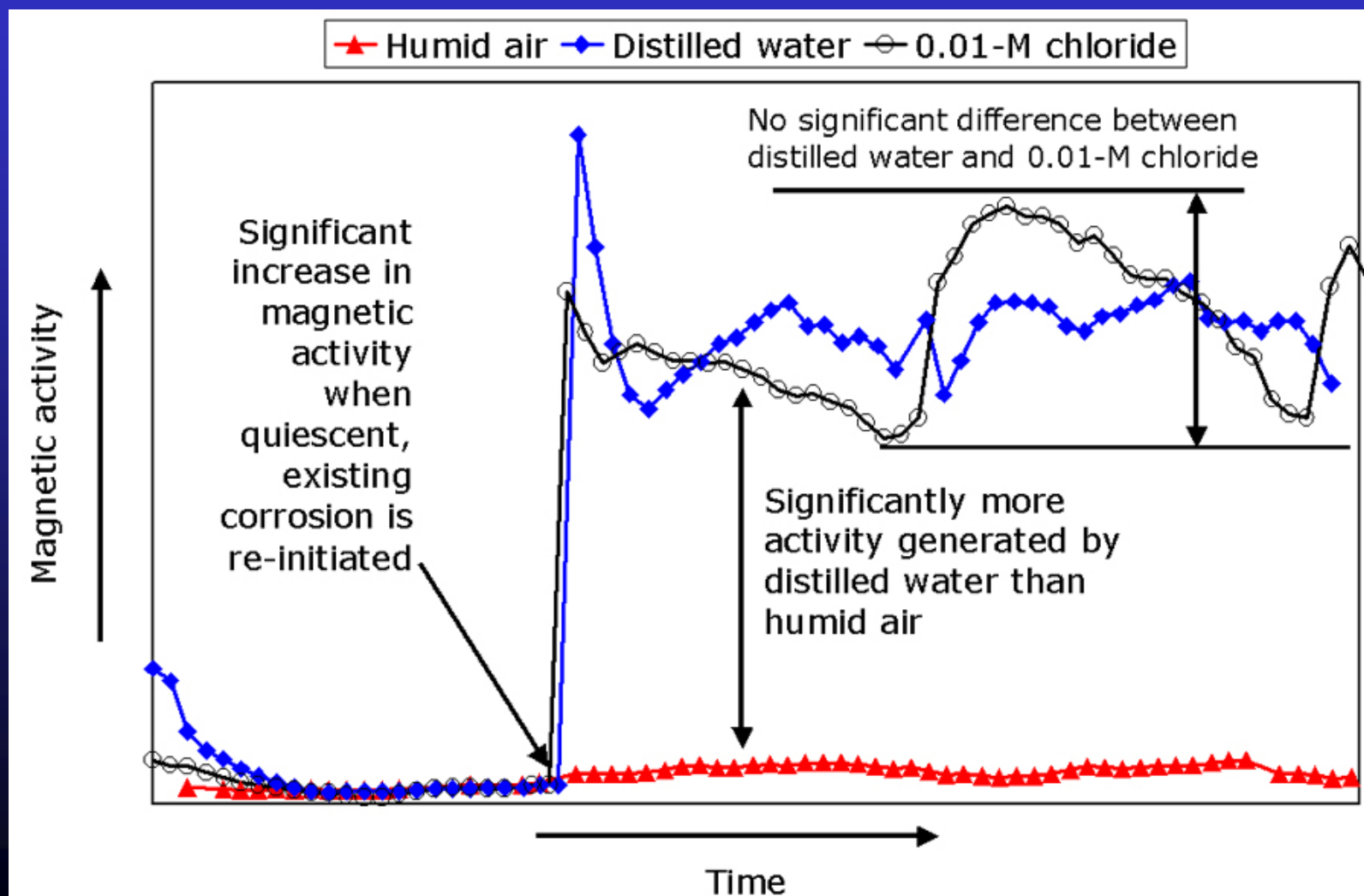


Varied activity depending upon lap joint, corrosive solution, and time

Reproducible dry background

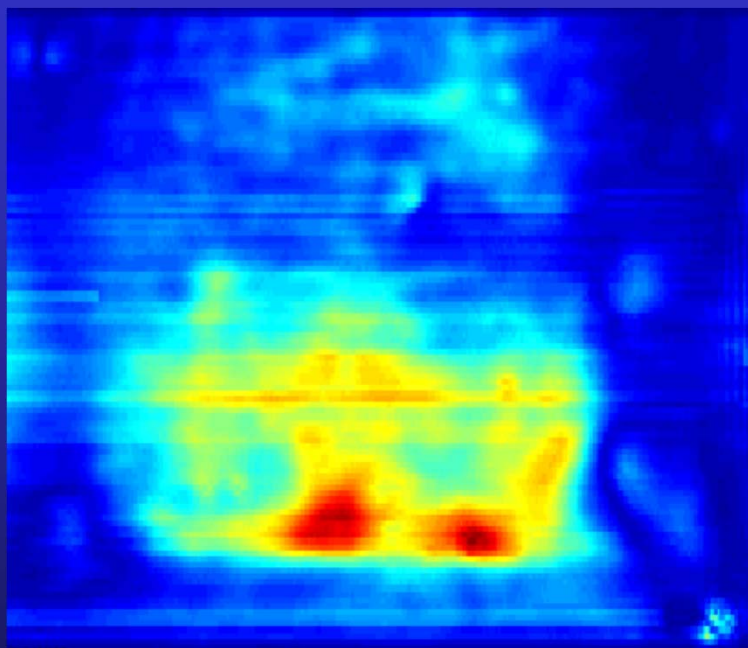


Lap Joint SIMA vs Environment

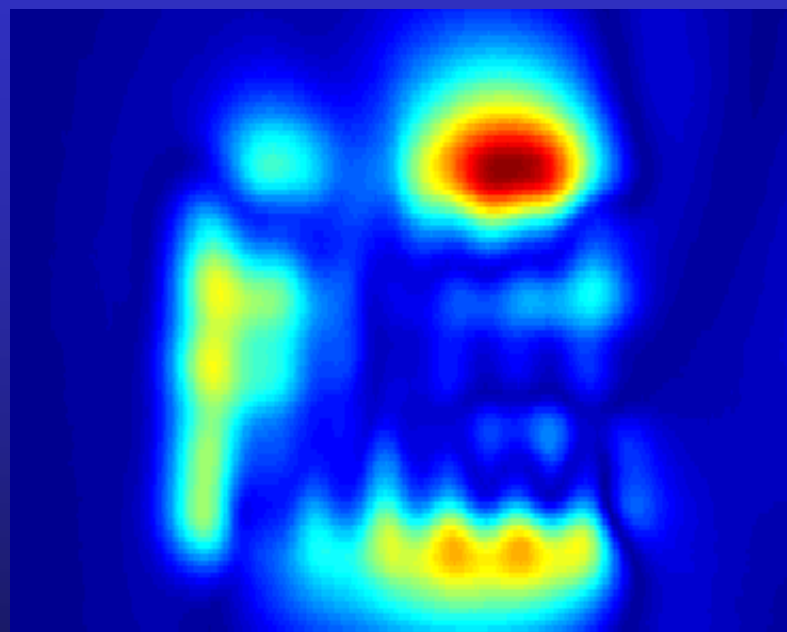




Spot-Welded Compared with Riveted: Cumulative activity map



Riveted Specimen

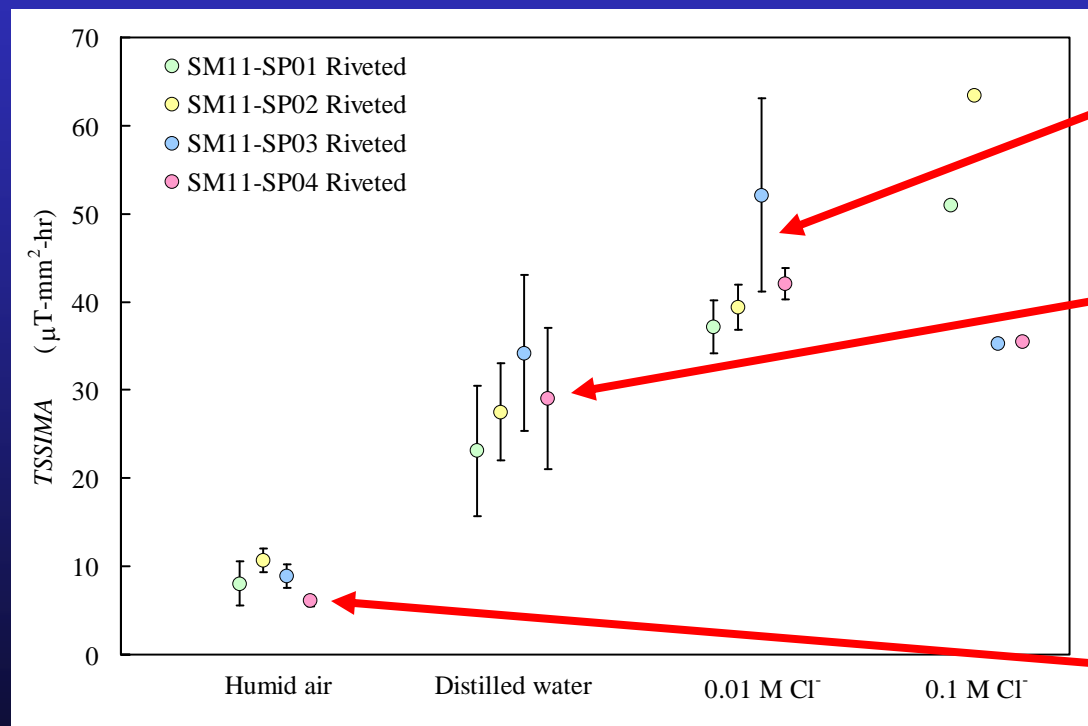


Spot-welded Specimen

Can identify internal structure apparently associated with spot welds compared with that of rivets



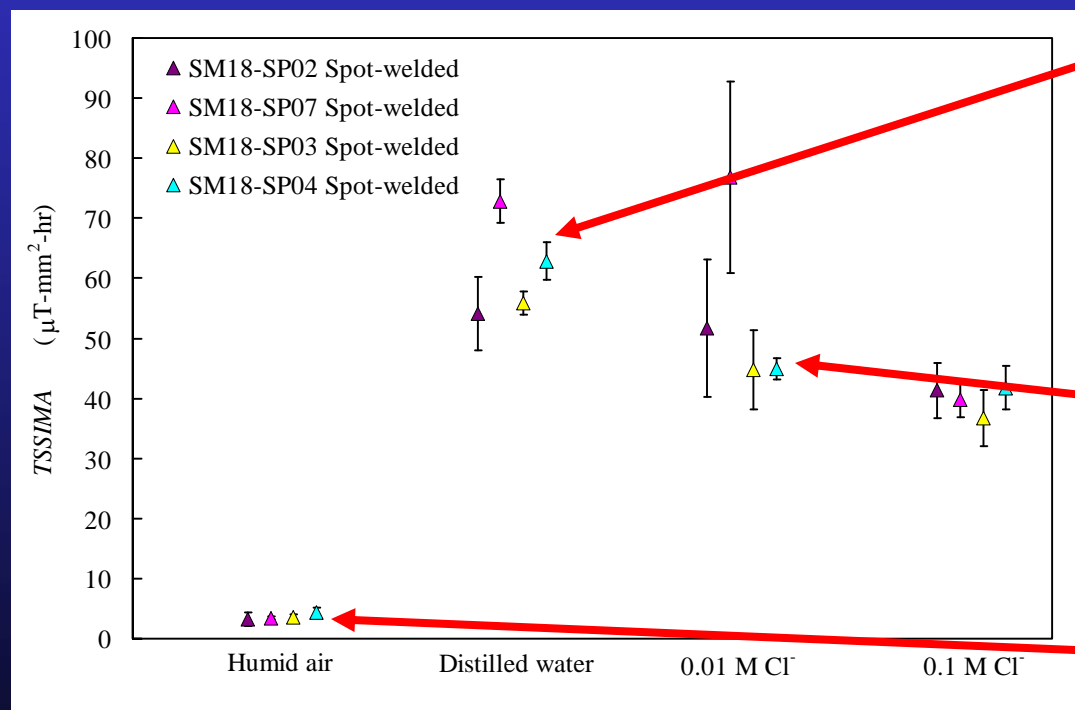
Summed Magnetic Activity Versus Time for Old Riveted Lap Joints



- 0.01 M chloride shows higher activity
- Distilled H₂O activates the chemistry within the lap joint
- Low activity in 98% relative humidity air



Summed Magnetic Activity Versus Time for Old Spot-Welded Lap Joints

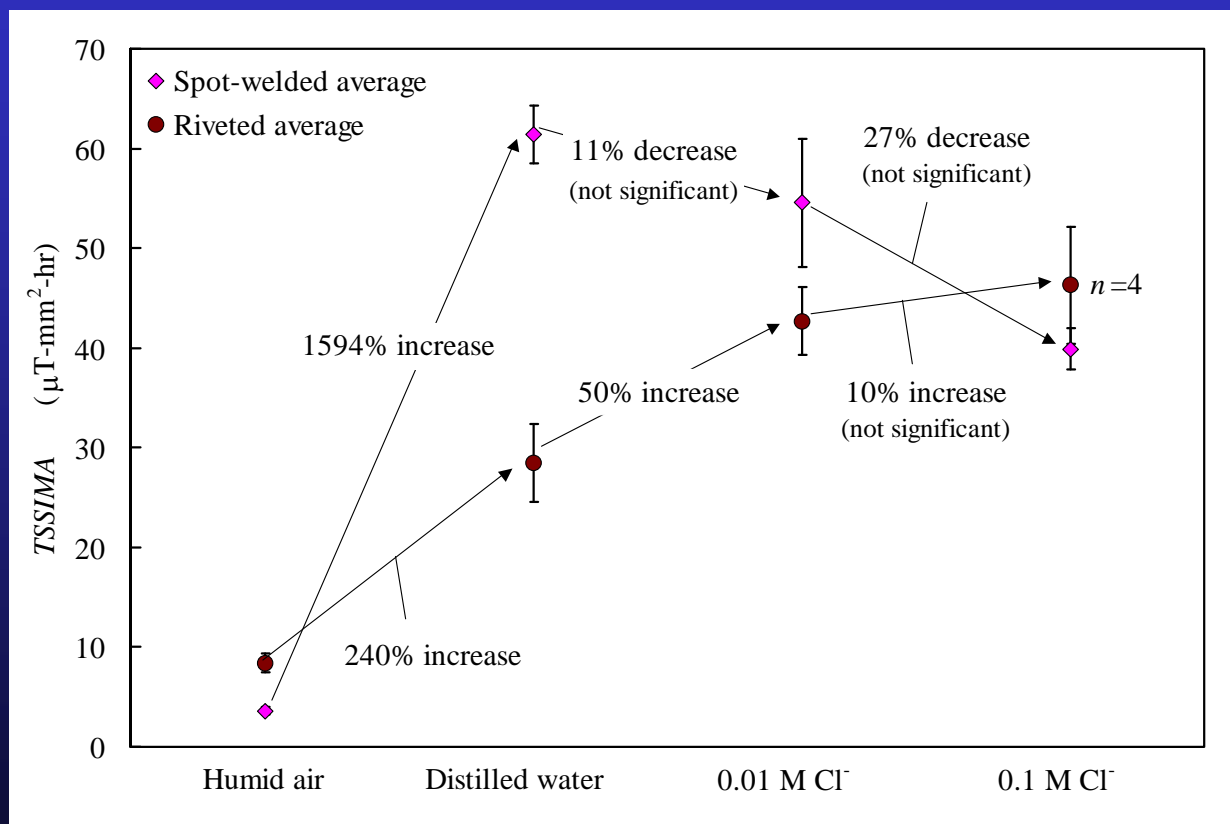


- Distilled H_2O activates the chemistry within the lap joint
- 0.01 M chloride shows lower activity than distilled water
- Low activity in 98% relative humidity air



Spot-Welded Compared With Riveted: Ratio of *TSSIMA*

- If an old lap joint is hydrated with distilled water, the chemicals already in the lap joints may be more important in the short term than what is added externally.
- **There may not be a strong dependence upon the concentration of externally-applied chloride.**





Old Lap Joint Conclusions

- SQUIDs can make useful measurements of instantaneous electrochemical activity in lap joints that are not possible with any other technique.
- These data are reproducible phenomenological representations of corrosion activity.
- We can assess the effects of moisture and NaCl on old riveted and spot-welded lap joints.
- **There may not be a strong dependence upon the concentration of externally-applied chloride.**
- **Kelly 1, SQUID 1**

Is the magnetic signal proportional
to corrosion?





Characterization Studies

Afshin Abedi and John Wikswo

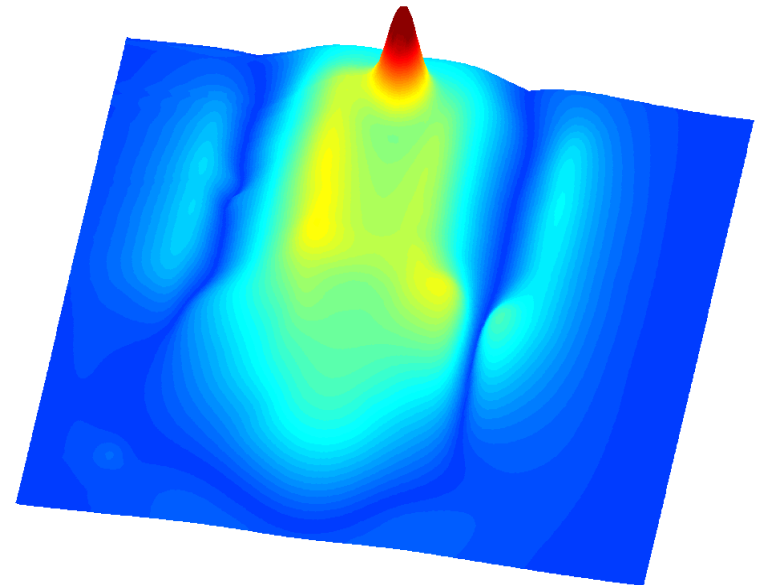
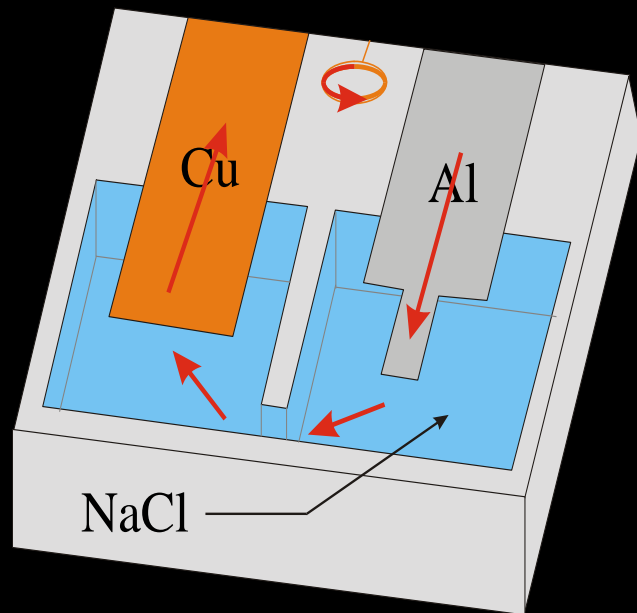
Examine the relationship between magnetic fields and the underlying corrosion activity

- Controlled **uniform corrosion**
 - 7075-T6 aluminum disks with mylar and epoxy masks
 - NaOH for uniform corrosion
 - Control corrosion area and pH

Galvanic cell with current loop

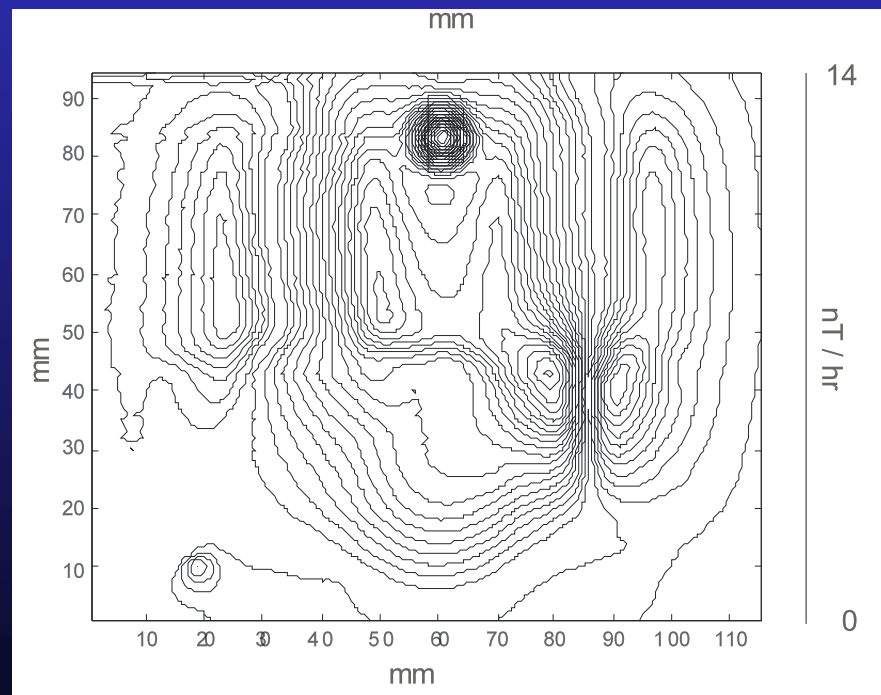
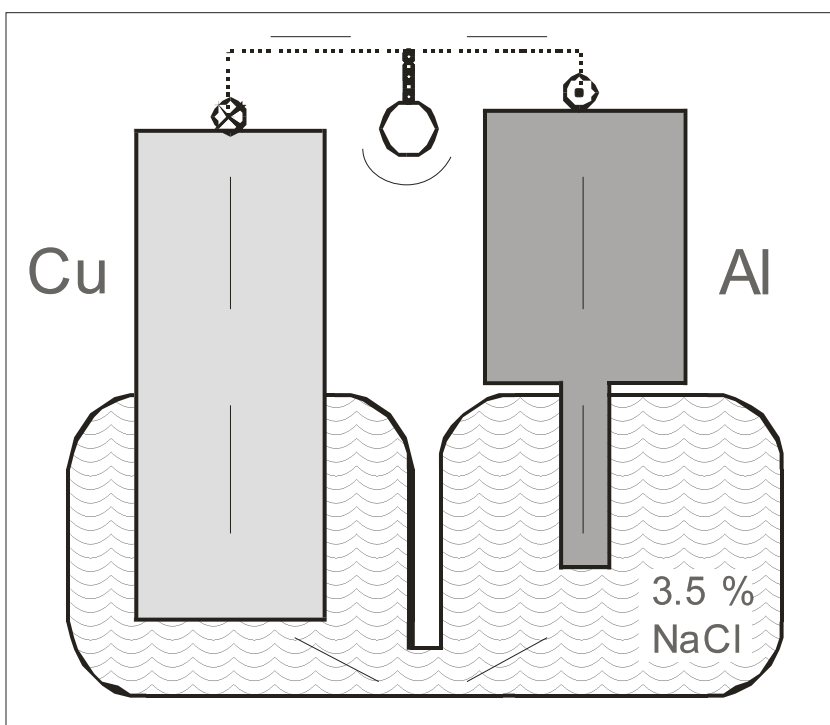
A SQUID as a Scanning, Zero-Resistance Ammeter

Afshin Abedi and John Wikswo



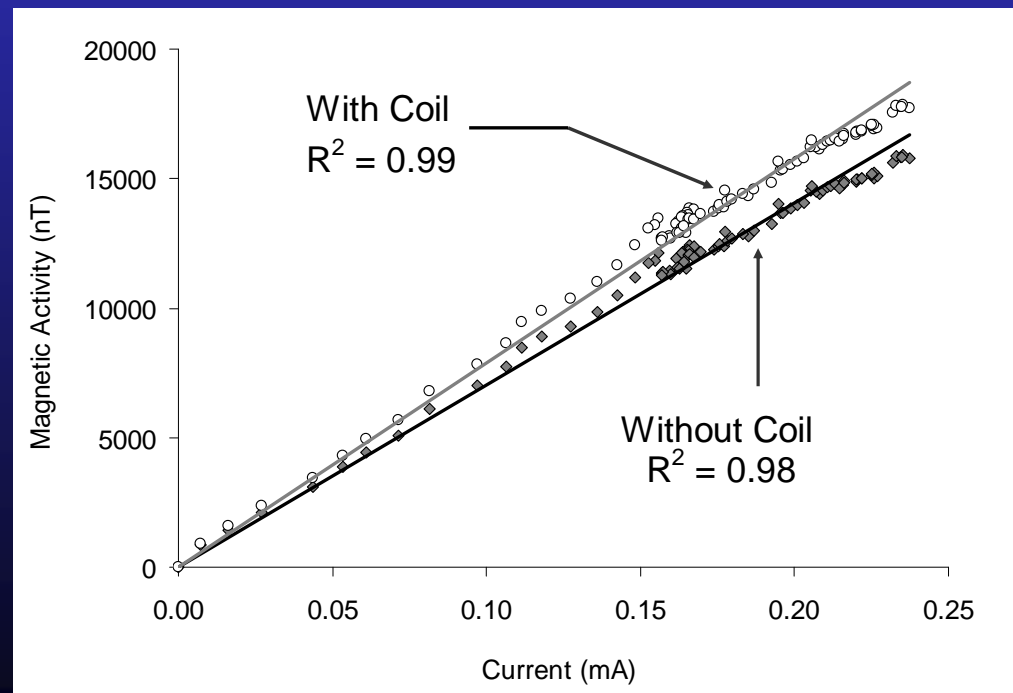
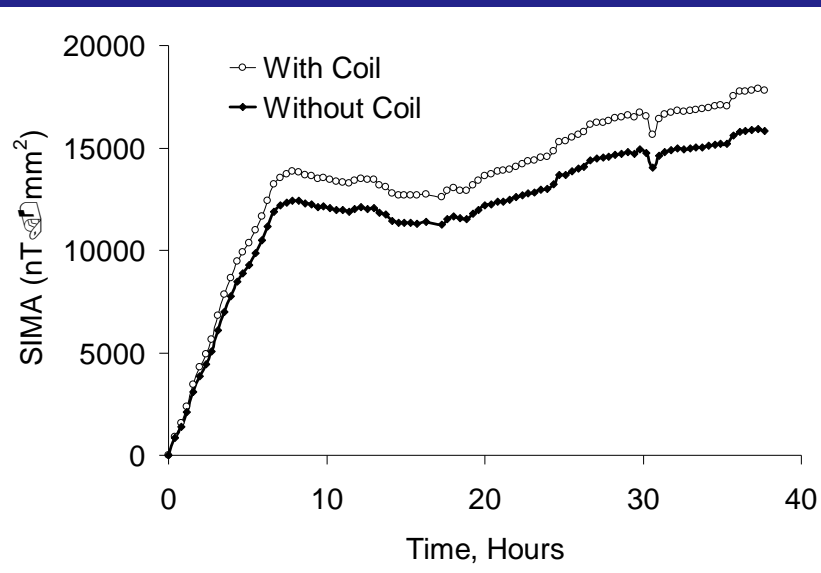


Plan View and TSMA Image



Galvanic Corrosion and a SQUID as a Scanning, Zero-Resistance Ammeter

- Spatially-integrated magnetic activity (SIMA) correlates well with the current that is required to produce the observed magnetic field in the small coil



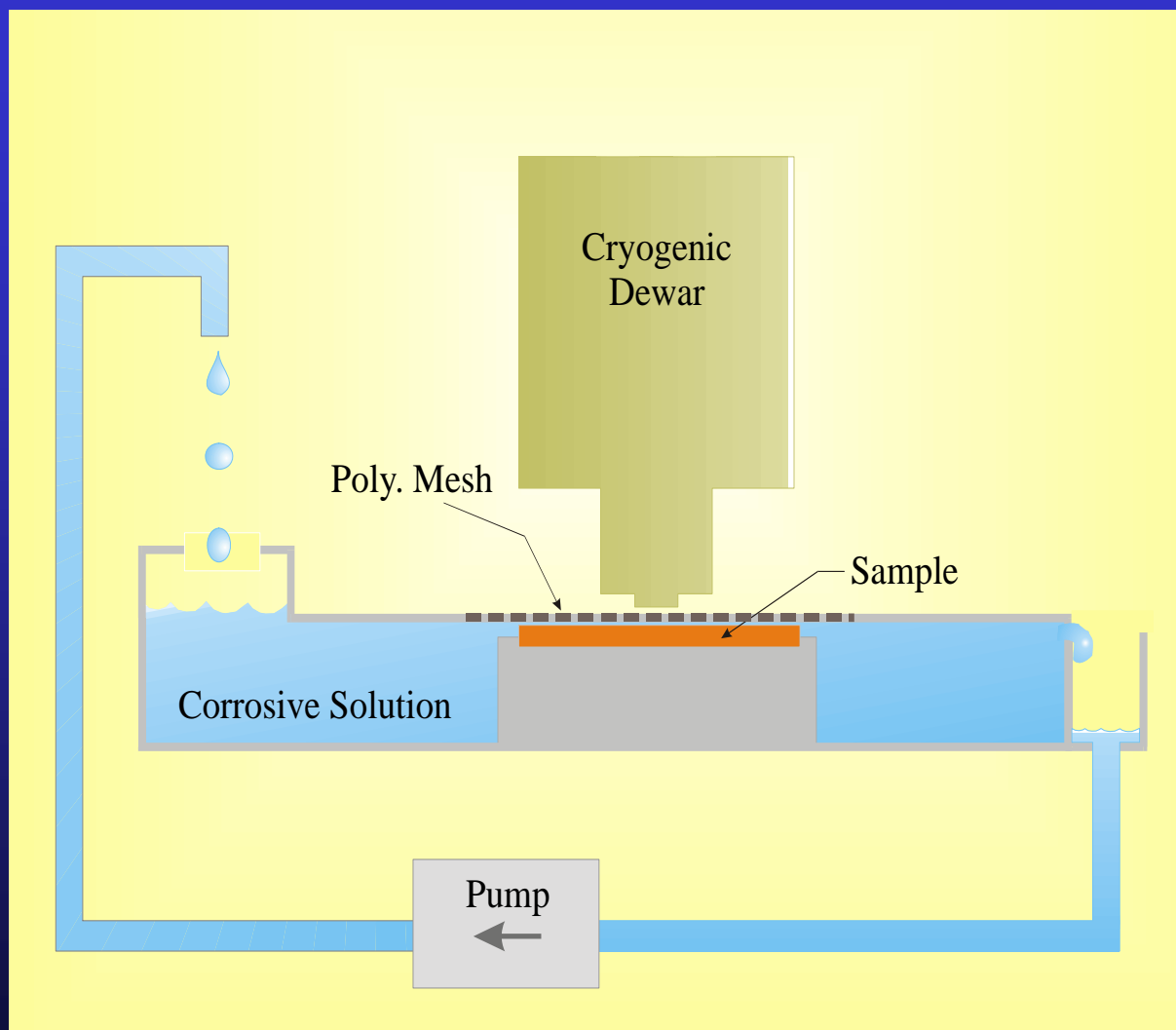
And how do the magnetic signals scale with corroding area and pH?





Uniform Corrosion Setup

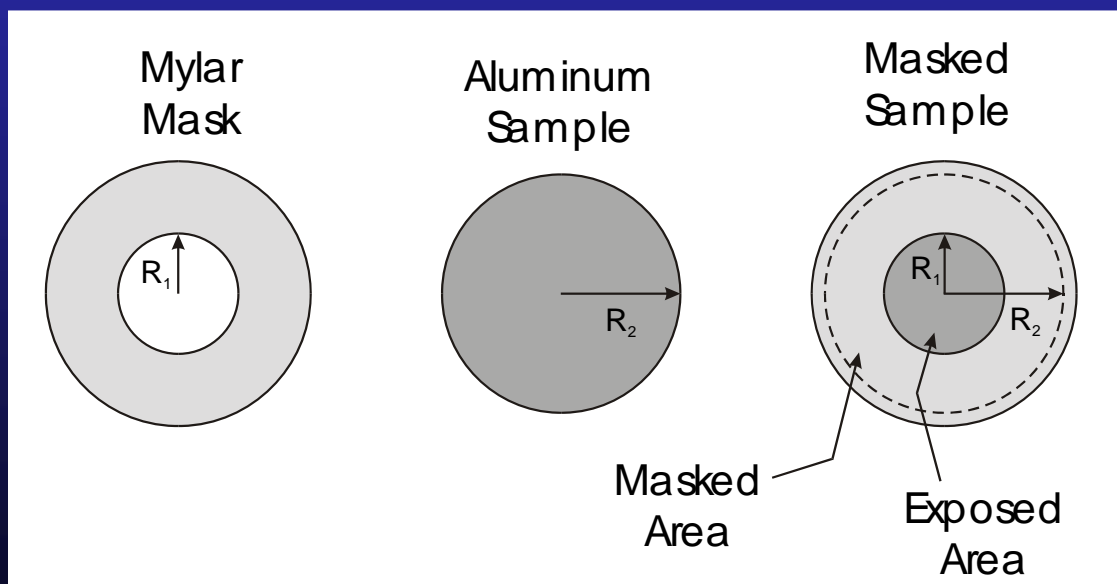
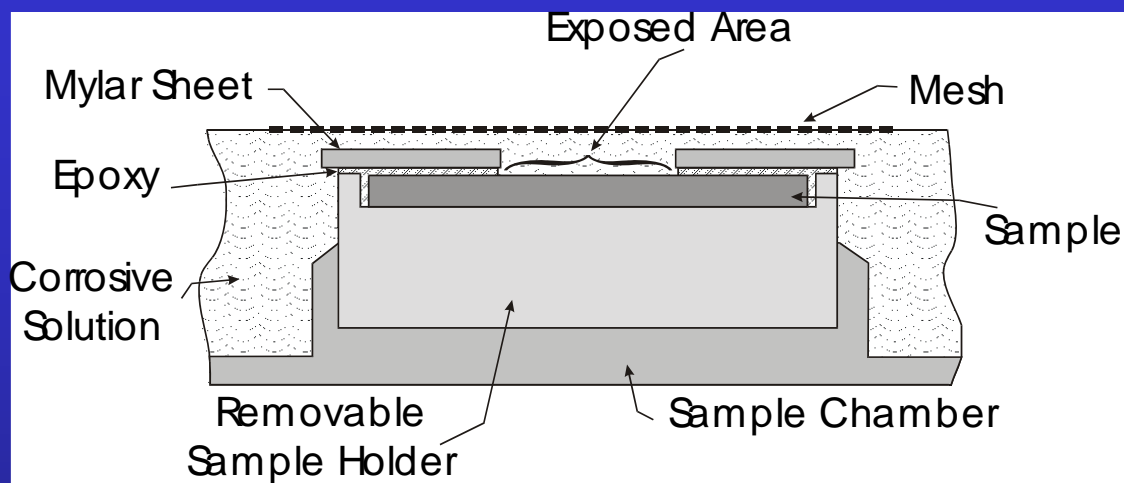
Afshin Abedi
and
John Wikswo





Uniform Corrosion Setup

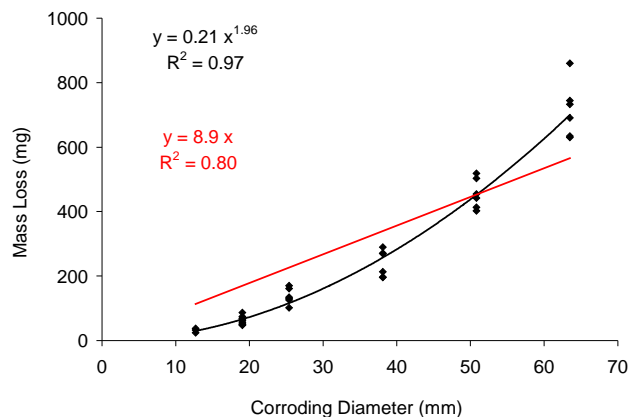
- 7075-T6 aluminum disks
- NaOH for uniform corrosion
- Control of corrosion area and pH



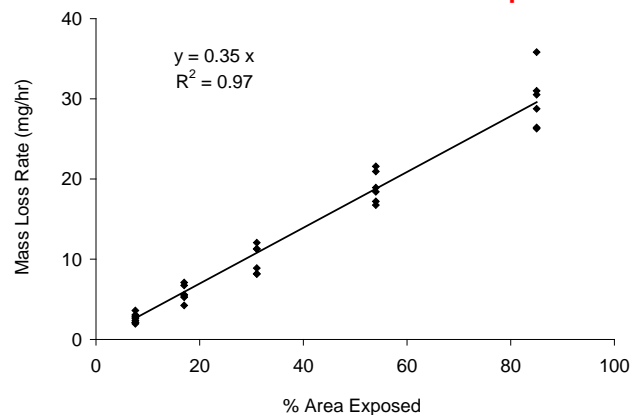


Corrosion versus Corroding Area

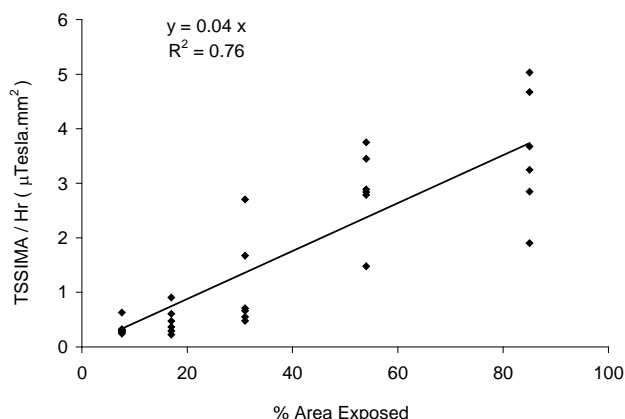
a) Mass Loss vs Diameter



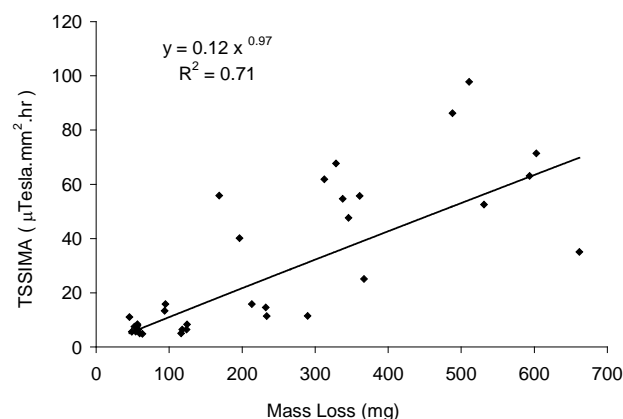
b) Mass Loss vs %Area Exposed



c) TSSIMA vs %Area Exposed



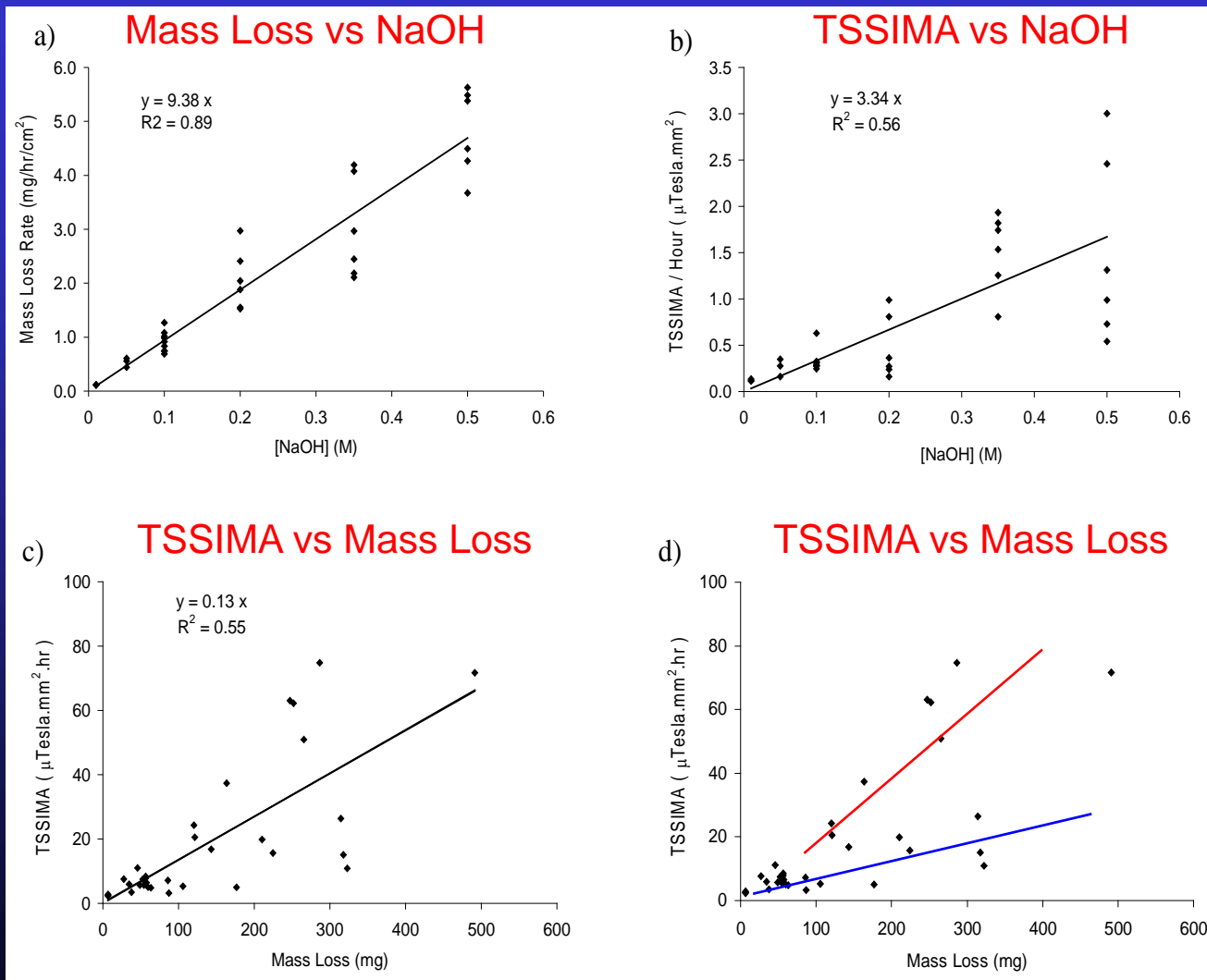
d) TSSIMA vs Mass Loss



- Mass Loss correlates with Exposed area ($R^2=0.97$)
- TSSIMA correlates with exposed area ($R^2=0.76$)
- TSSIMA correlates with mass loss ($R^2=71$)



Corrosion versus pH (19 mm spot)



- Mass Loss correlates with NaOH ($R^2=0.89$)
- TSSIMA correlates with NaOH ($R^2=0.56$)
- TSSIMA correlates with mass loss ($R^2=0.55$)
- TSSIMA greater for non-uniform corrosion (red) than for uniform corrosion (blue)

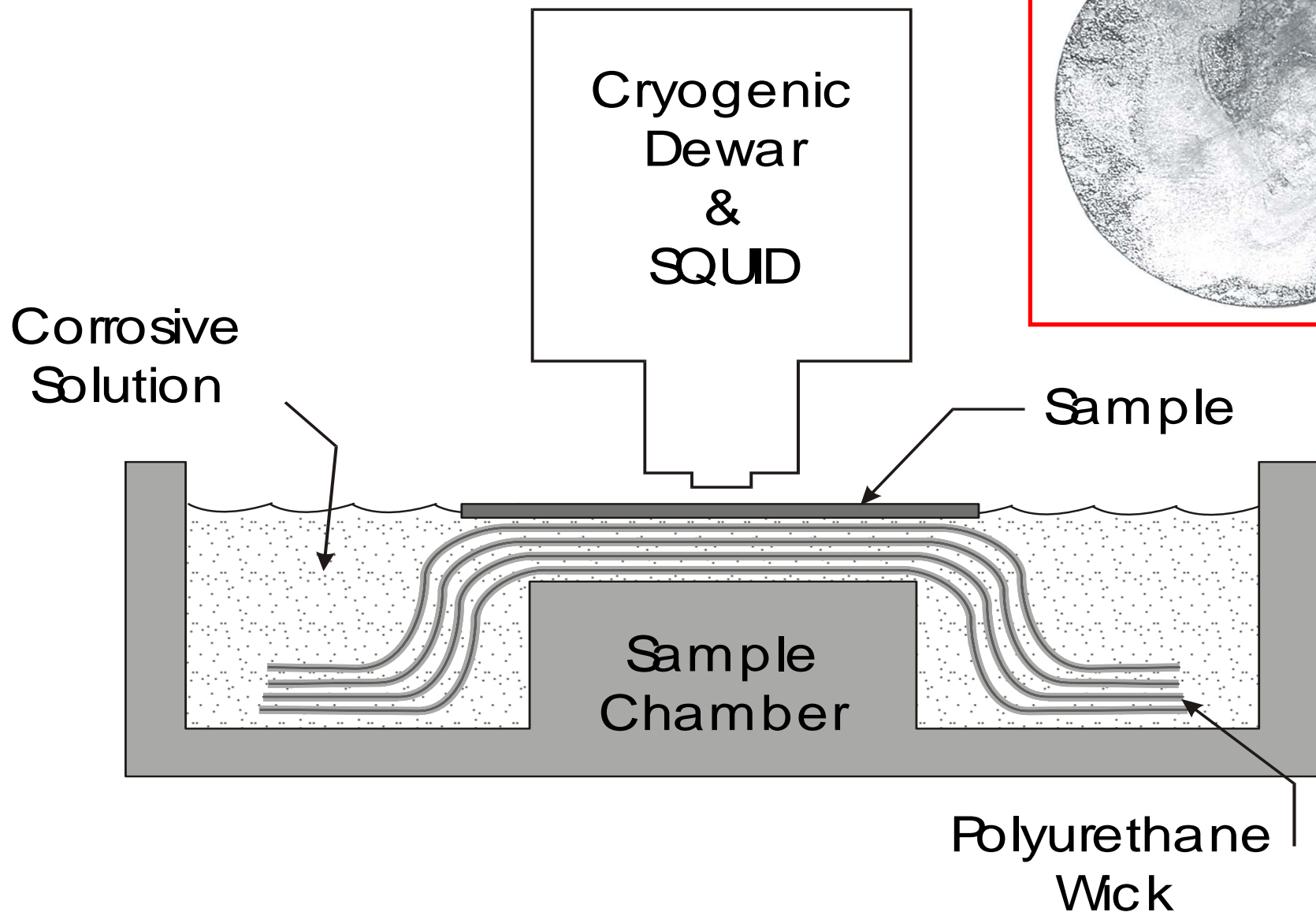
Uniform vs Heterogeneous Corrosion





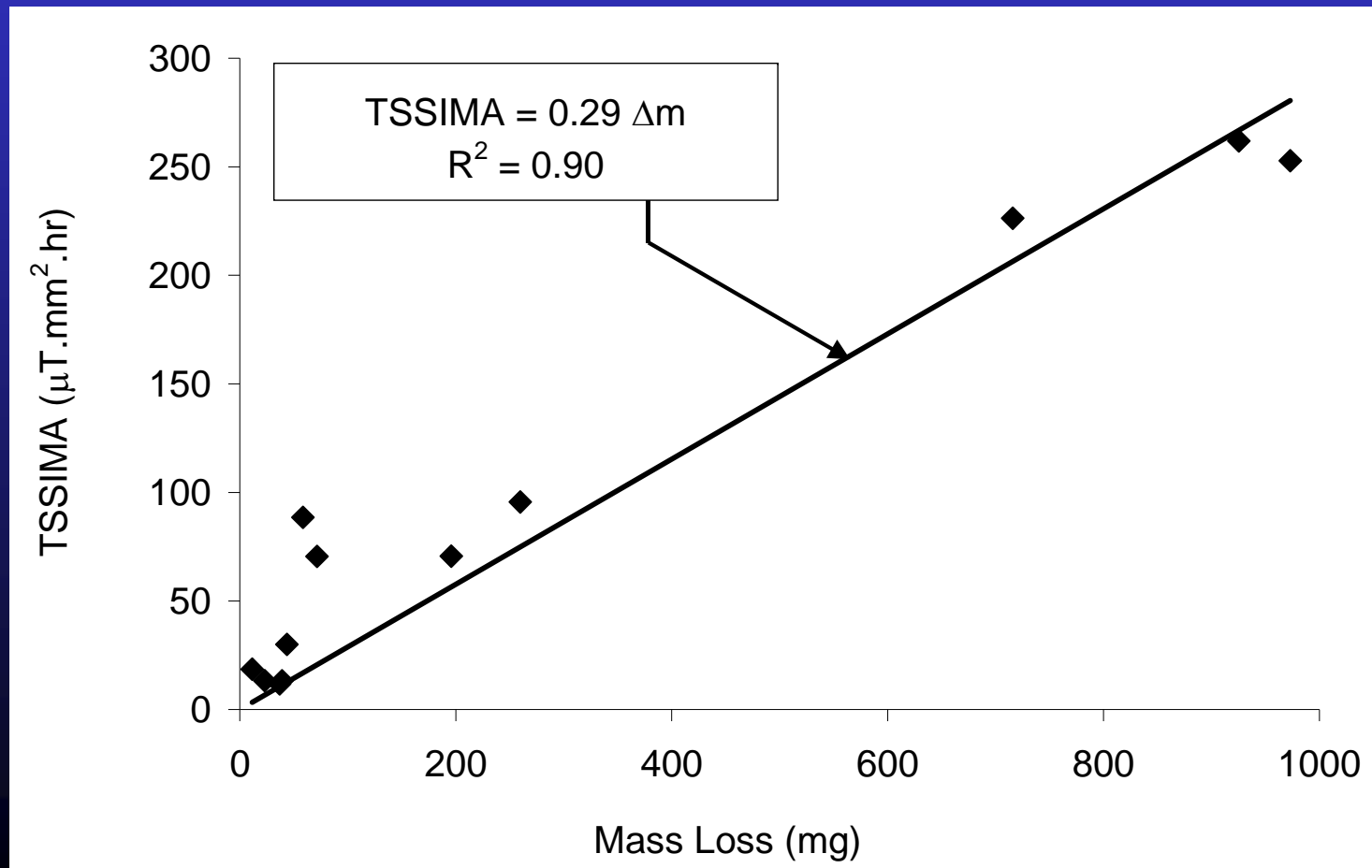
Hidden, Inhomogeneous Corrosion

0.01M NaOH



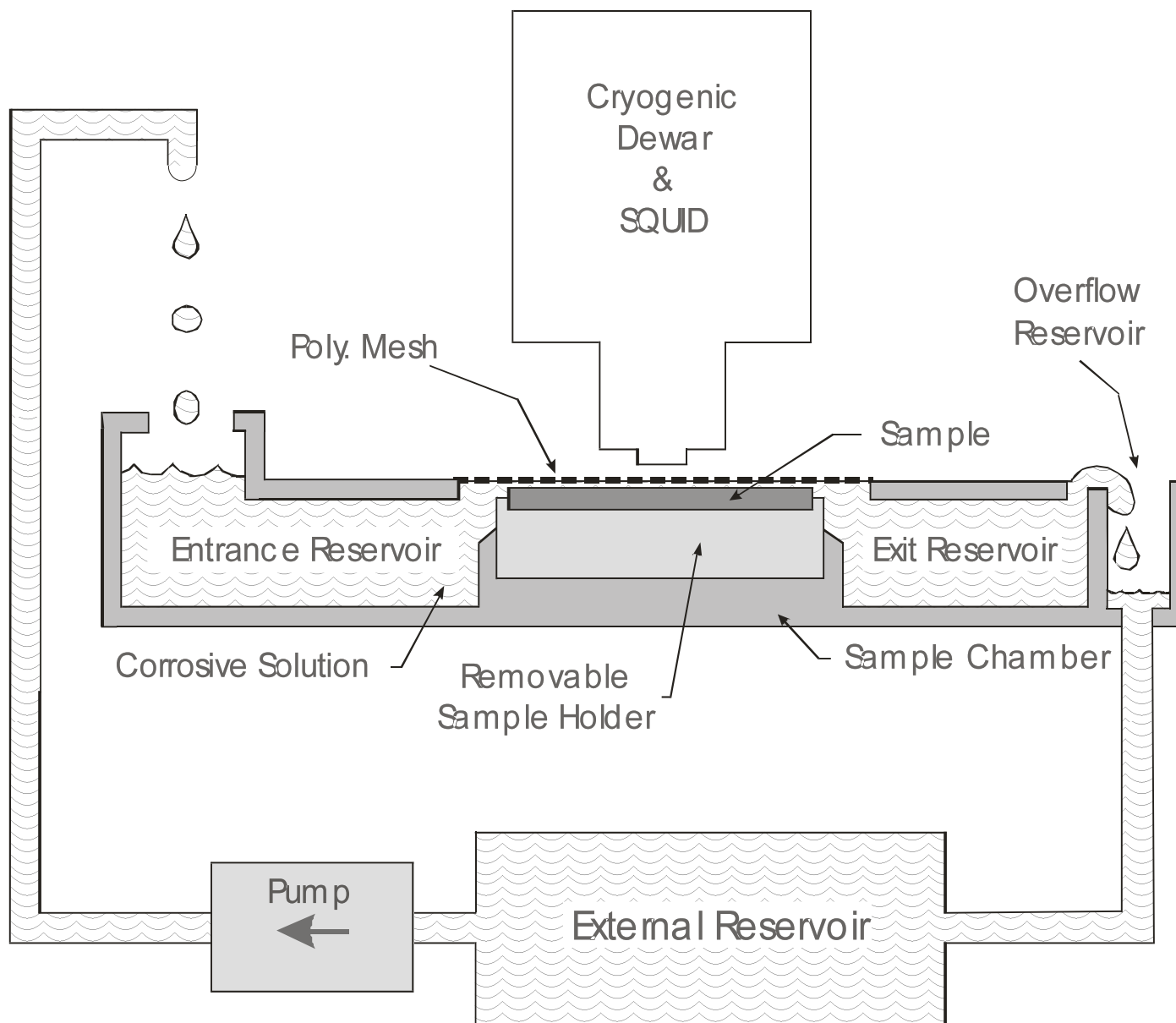
TSIMMA vs Mass Loss

- CMF in regular time intervals above a 70-mm diameter, 0.8-mm thick, 7075-T6 aluminum disk submerged in a 0.5-M NaOH solution.



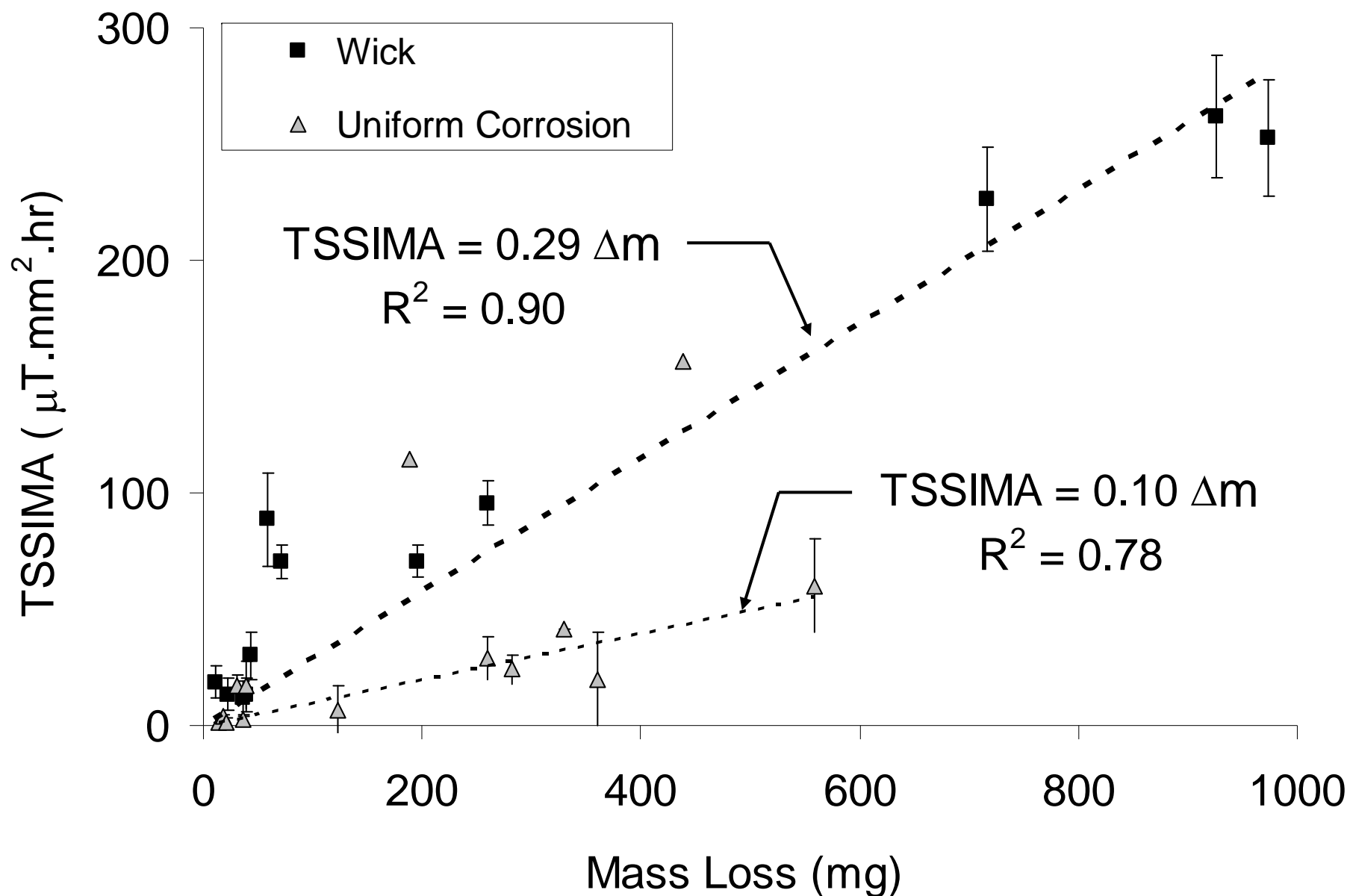


Exposed, Homogeneous Corrosion 0.1M NaOH





TSIMA vs Mass Loss



Just how sensitive are SQUIDS?



<http://www.tenhand.com/squid/squid-okeanos1.jpg>

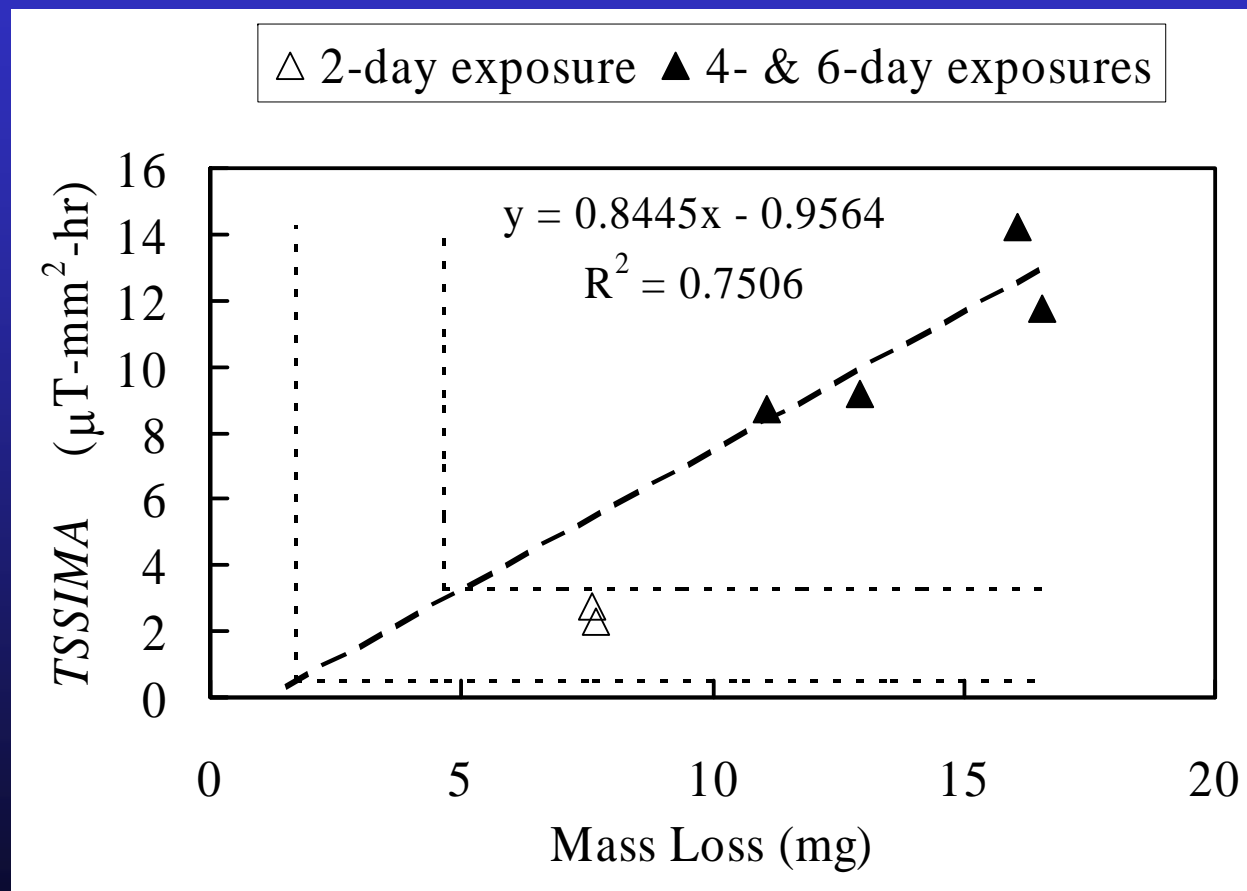
The Sensitivity of SQUID Magnetometers for Determining the Instantaneous Penetration Rate of Hidden Corrosion

**Robert G. Kelly, Grant Skennerton
Afshin Abedi, John P. Wikswo**



Simulated Lap Joints

- 76 mm x 76 mm 2024-T3 lap joint, N=6
- Weigh
- Hi-lock fasteners
- 0.1 M NaOH
- Disassemble
- Clean
- Reweight





SQUID magnetometers can detect ..

- mass loss rates as low as $0.8 \text{ mg/cm}^2\text{-y}$
- uniform penetration rate of 0.12 mils per year
- 3.5×10^{-4} mils per day ($3 \text{ }\mu\text{m/yr}$ or 9 nm/day).
- **Kelly 2, SQUID 2**

Preliminary Tests of SQUID Detection of Exfoliation Corrosion



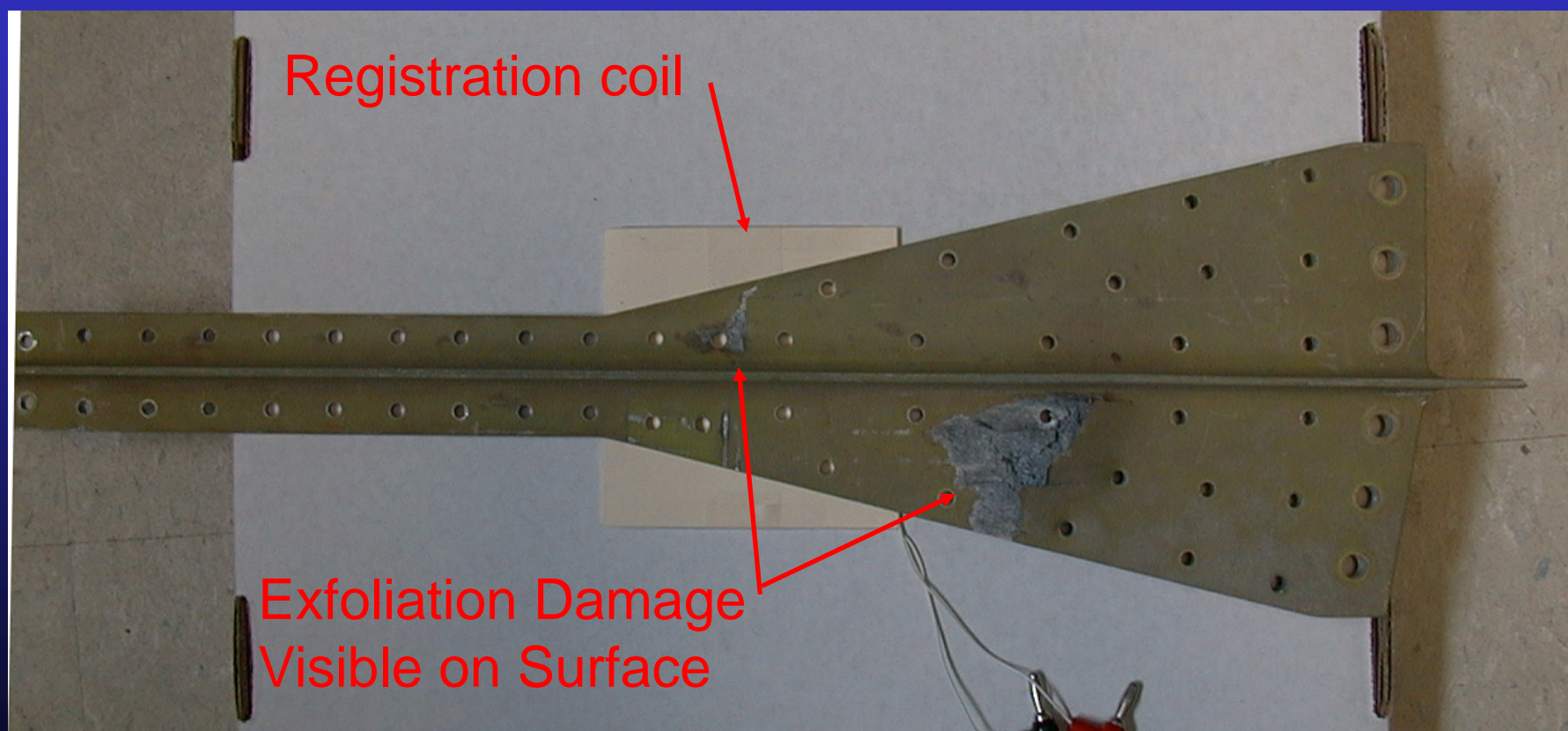
Preliminary Tests of SQUID Detection of Exfoliation Corrosion

Yu Pei Ma, John Wikswo

- Sample
 - Horizontal stabilizer carry through box stiffener
 - Aircraft MDS - KC-135
 - Material: 7075-T6 Forging
- Protocol
 - SQUID above flat side (side not shown)
 - Scan in air for baseline recording
 - Submerge distilled water and scan for one week



Horizontal Stabilizer Carry Through Box Stiffener with Square Registration Coil

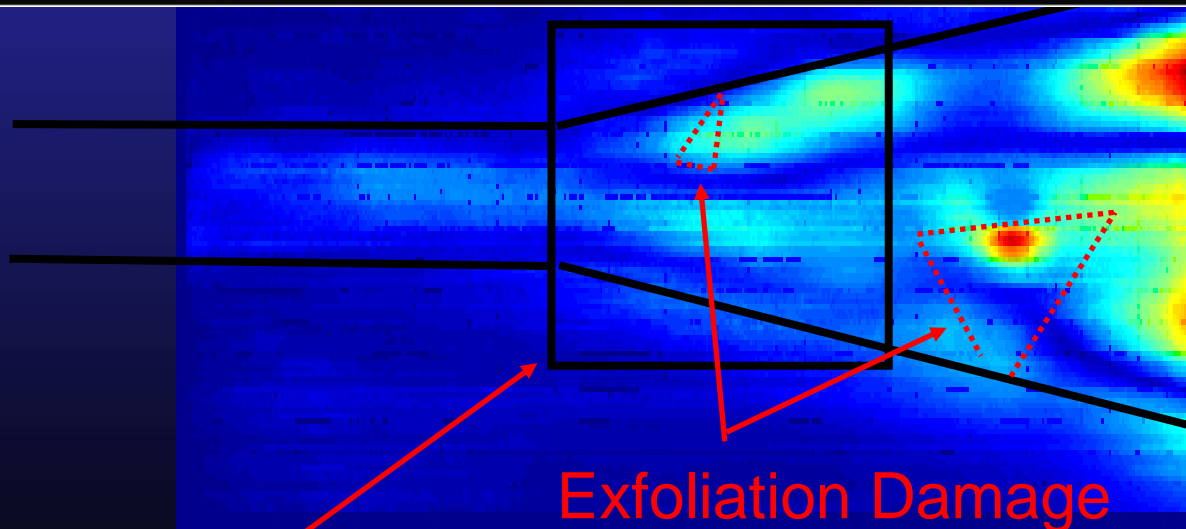
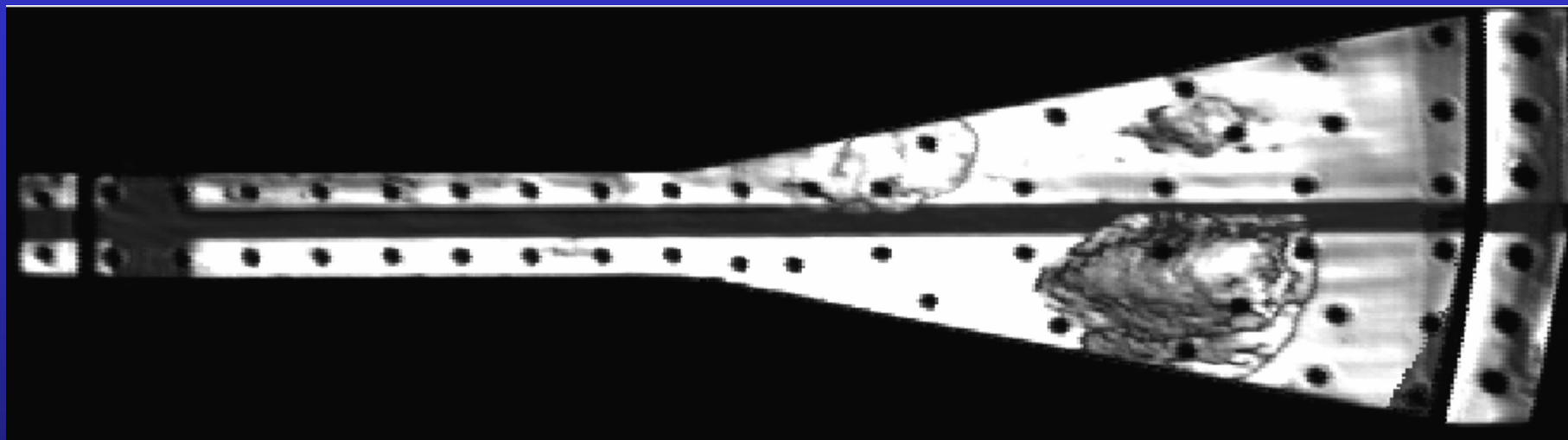




Close-Ups of Exfoliation Damage



Ultrasound and Temporally Summed Magnetic Activity (TSMA) After One Week of Exposure



Registration coil
Exfoliation Damage
Visible on Surface



Conclusions – Box Stiffener Exfoliation

- SQUIDs can readily detect exfoliation corrosion in 7075 forgings
- Deb Peeler chopped up our sample... ; (
- Needed
 - Simpler geometry
 - Correlations of SQUID with NDE and metallography

SQUID Imaging of Exfoliation and Intergranular Corrosion

John P. Wikswo and Yu Pei Ma

Vanderbilt University

Kevin Cooper, Luna Innovations, Inc.,

James Suzel, S&K Technologies

Robert Kelly, University of Virginia



Luna/S&K/VU Protocol E1

- Samples: Kaiser 0.350 7075-T6 (lot 274371) 4" wide by 10" long, grain lengthwise.
- Holes" three 3/8" holes approximately 1/8" deep and 2" apart.
- Coated twice everywhere except sides of holes with XP-2000 sealant; 0.040-0.050 bare aluminum on hole sides
- Holes filled with ANCIT solution
- Anticipate 1-3 mm of penetration in 48-96 hours.



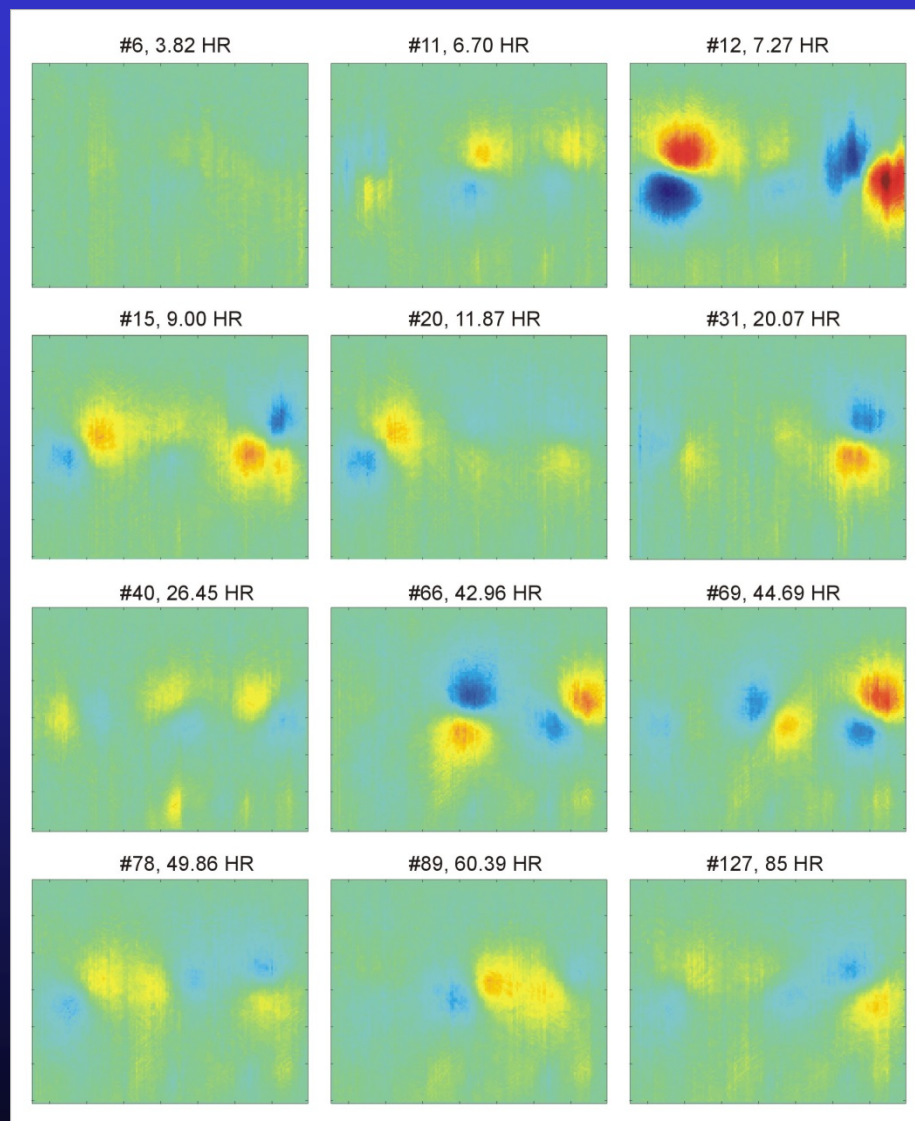
Sample VEX001





SQUID Images During Exfoliation Corrosion Development

- Corrosion activity visible within 5 hours of exposure and reach maximum about 7.5 hours
- Time-dependence of corrosion differs from hole to hole over short time intervals (Frame #12 vs #15, and #66 vs #69)





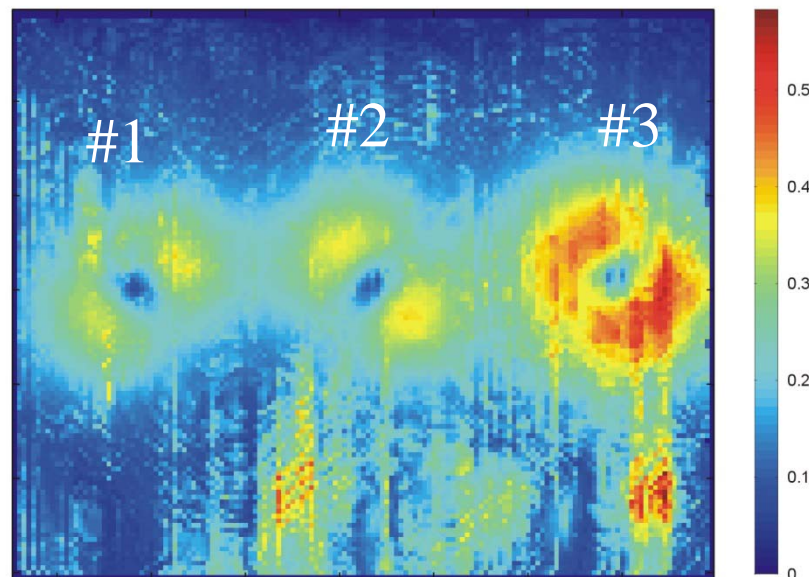
Exfoliation Solution

- **ANCIT Solution:**
 - 4 M NaCl
 - 0.6 M KNO₃
 - 0.022 M AlCl₃ (as AlCl₃ · 6H₂O)
 - natural pH ~ 3 to 3.3
- ASTM G34-90, in *Annual Book of ASTM Standards - Metal Test Methods and Analytical Procedures*, Vol. 03.02 Wear and Erosion; Metal Corrosion, ASTM, Philadelphia, PA, 119-124 (1990).

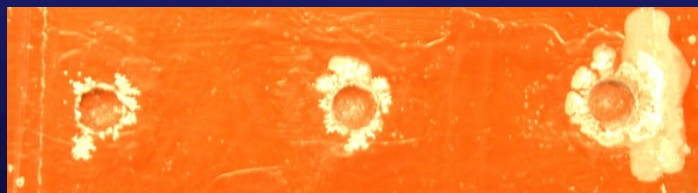


TSMA for three holes (VEX001)

TSMA - THREE HOLES

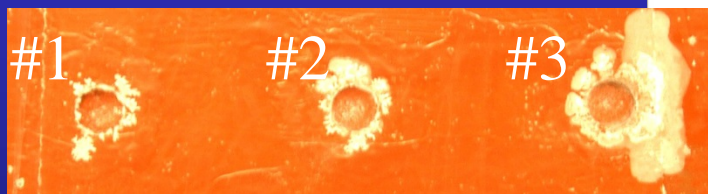


Holes after exposing
to solution

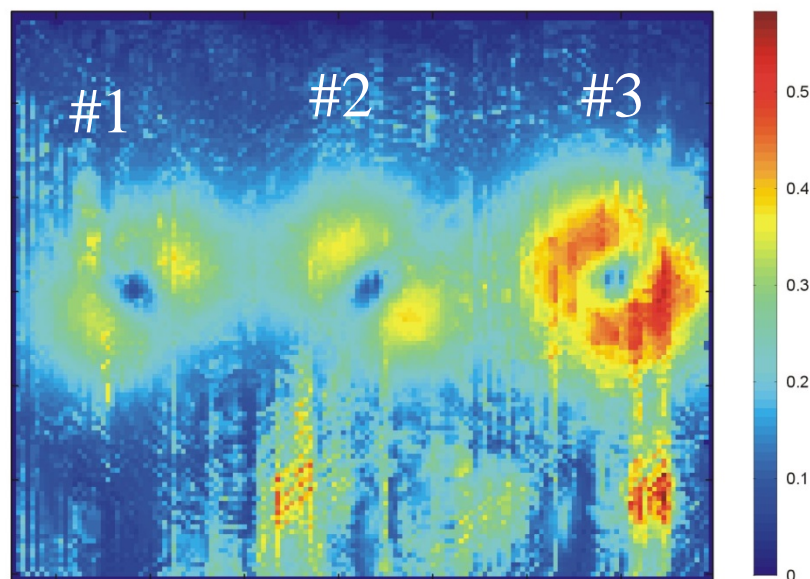




TSMA for three holes (VEX001)

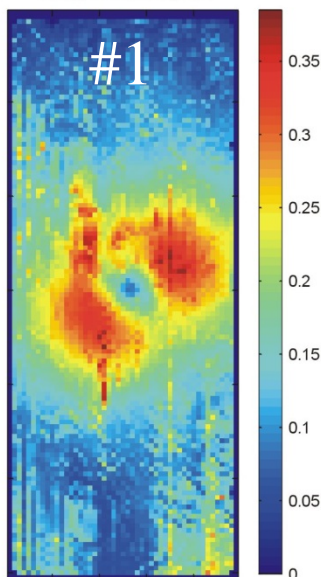


TSMA - THREE HOLES

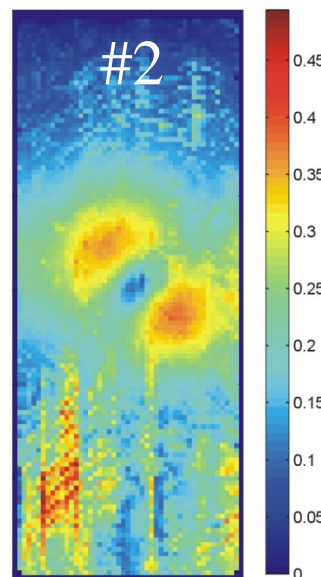


Holes after exposing
to solution

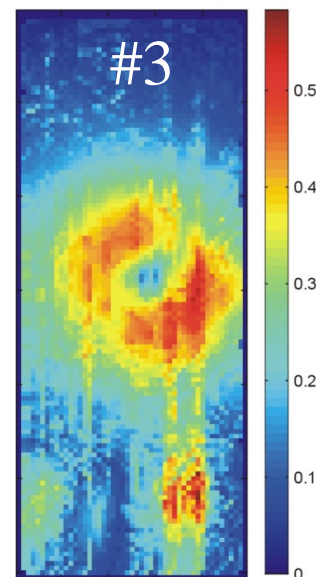
TSMA-1STHOLE



TSMA - 2NDHOLE



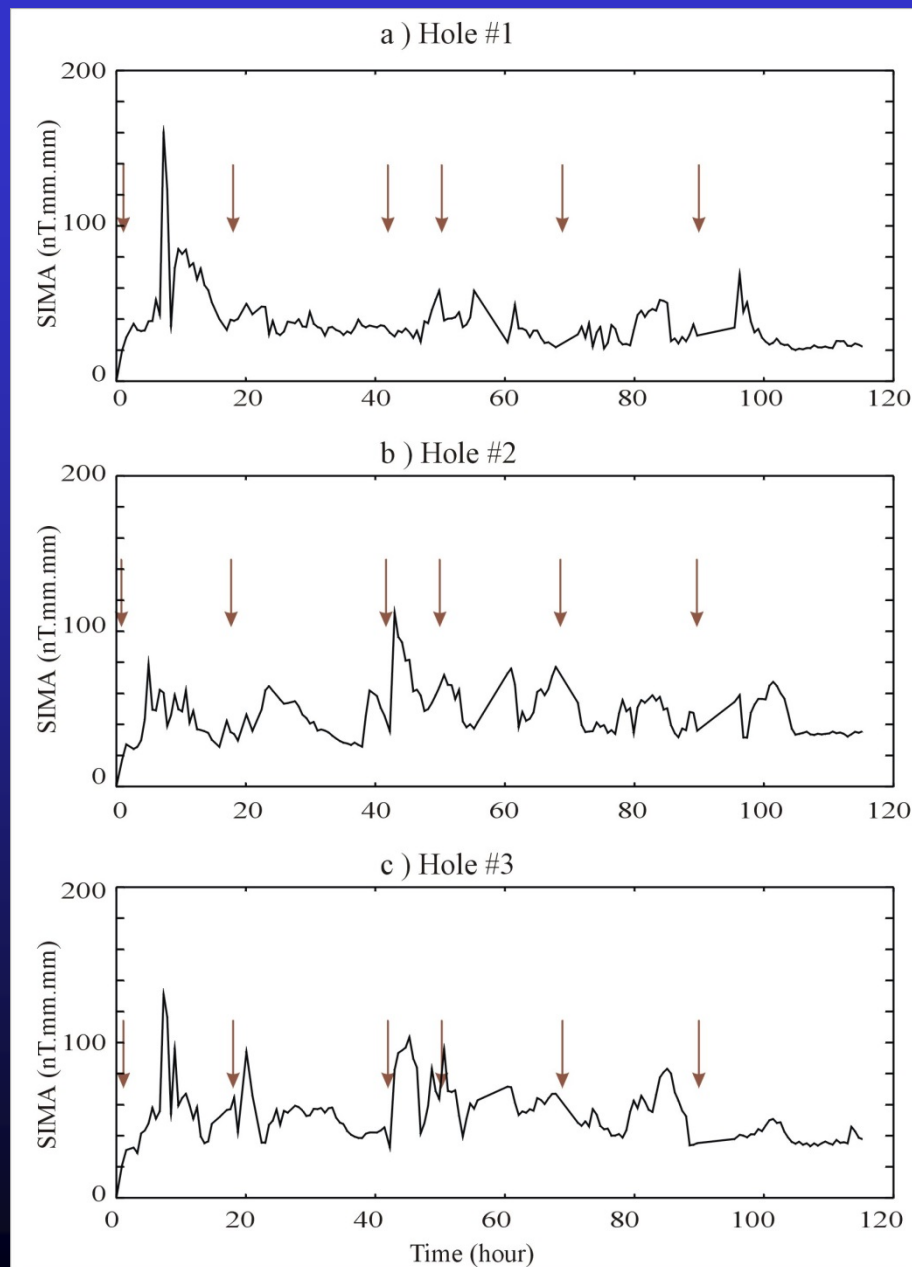
TSMA - 3RD HOLE





SIMA for three holes (VEX001)

- 35 minutes/data point
- Maximum signals after 7-8 hours
- Arrows indicate time of adding solution to the holes

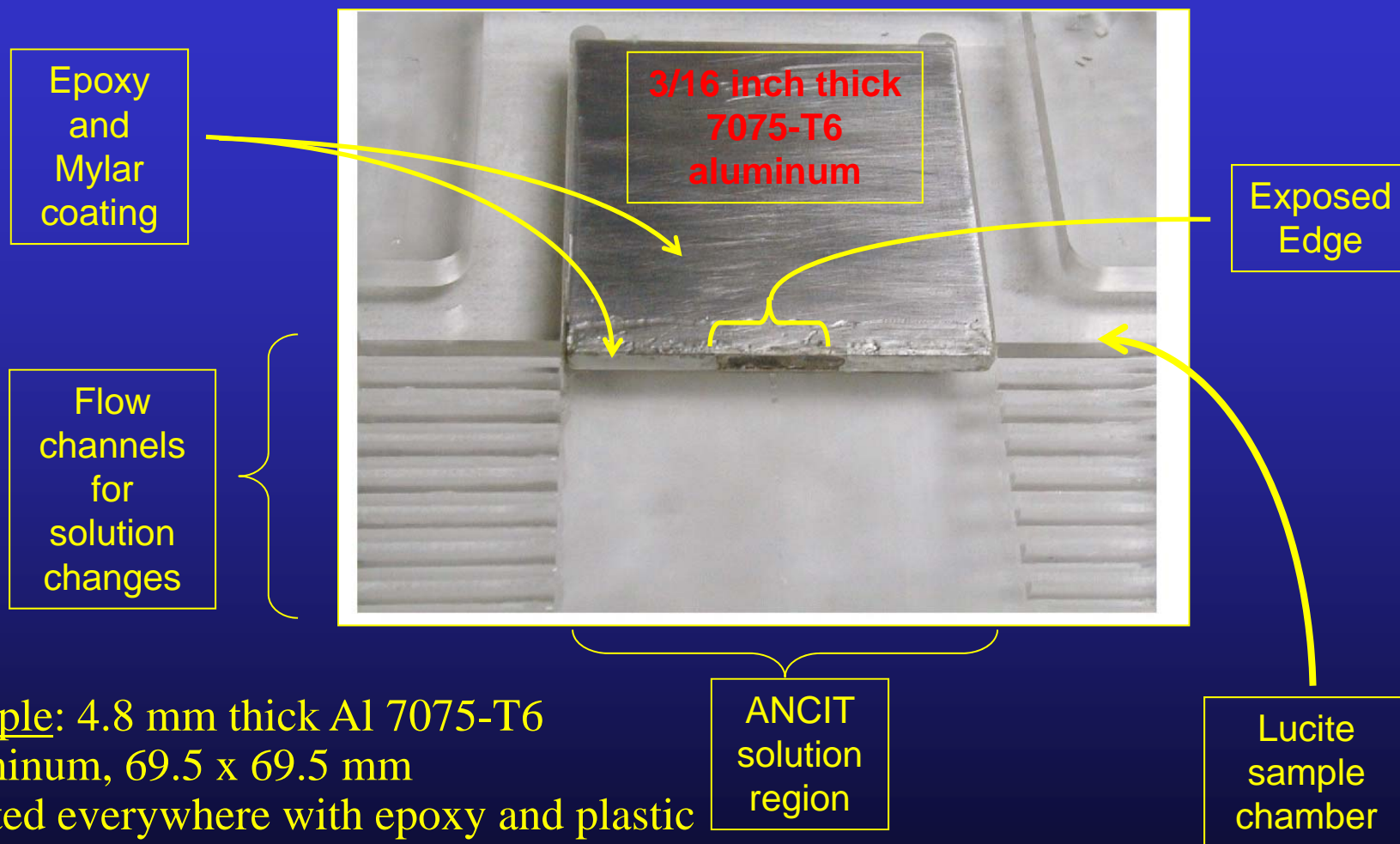


Protocol E1 Conclusions

- SQUID can see corrosion in well sample, with clear time-dependence over 96 hours
- TSMA images of individual hole is consistent with the corrosion activity
- The evaporation of solution in holes cause crystal accumulate around and inside the holes which may block the reaction.
- **NDE of damage does not show intergranular corrosion.**
- **Kelly 2, SQUID 3**



VU Protocol E2 - Edge Test



Sample: 4.8 mm thick Al 7075-T6 aluminum, 69.5 x 69.5 mm

Coated everywhere with epoxy and plastic film except 1/3 of one edge

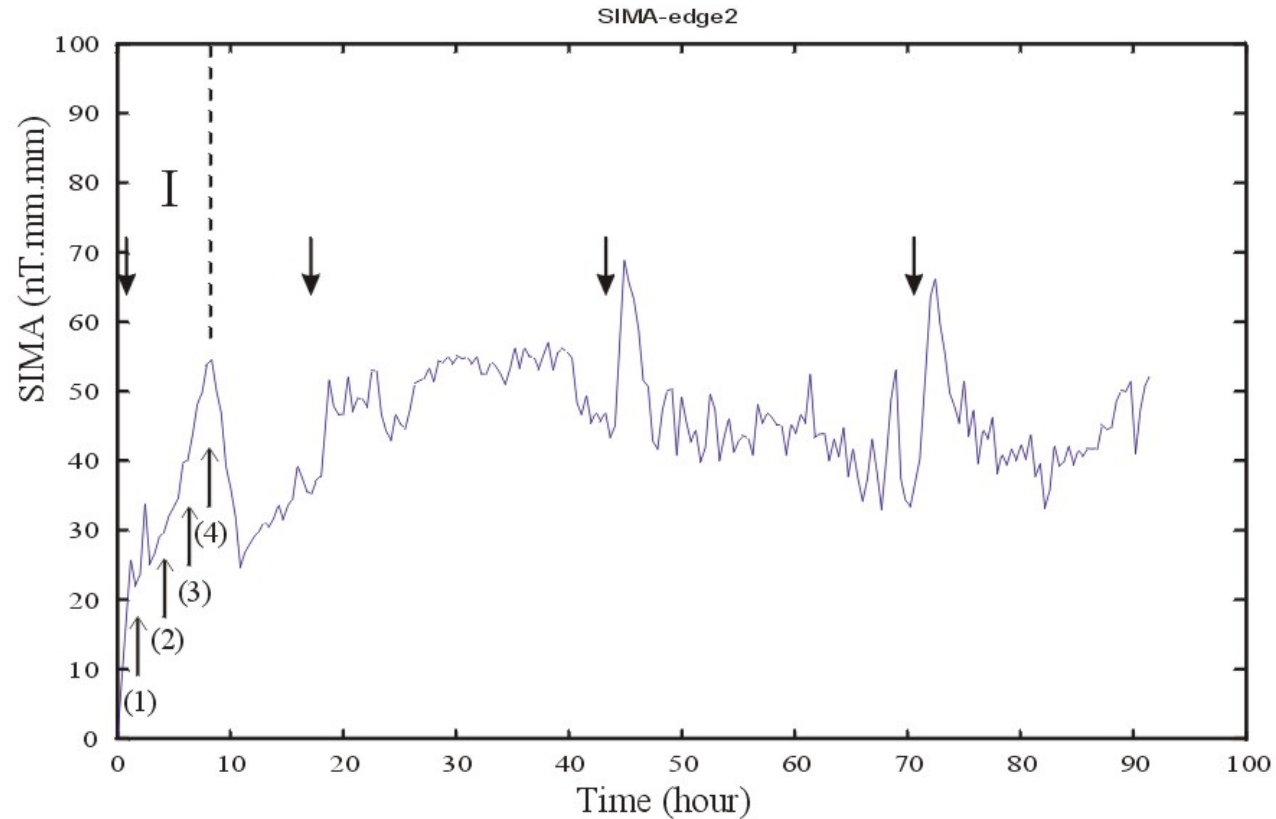
Uncoated edge is exposed to ANCIT solution for 96 hours



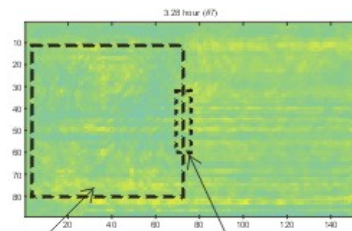
Exfoliation Solution

- ANCIT Solution:
 - 4 M NaCl
 - 0.6 M KNO₃
 - 0.022 M AlCl₃ (as AlCl₃ · 6H₂O)
 - natural pH ~ 3 to 3.3
- ASTM G34-90, in *Annual Book of ASTM Standards - Metal Test Methods and Analytical Procedures*, Vol. 03.02 Wear and Erosion; Metal Corrosion, ASTM, Philadelphia, PA, 119-124 (1990).

I. Initiating - eight hours after introducing solution

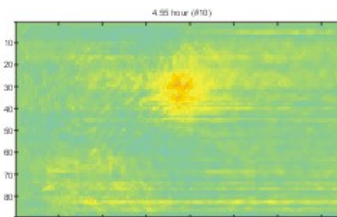


(1) 3.28 hour (#7)

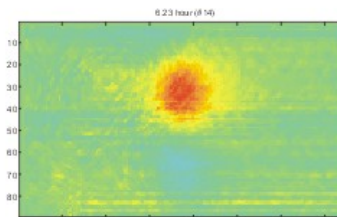


Al plate Exposed edge

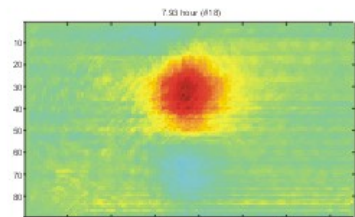
(2) 4.55 hour (#10)



(3) 6.23 hour (#14)

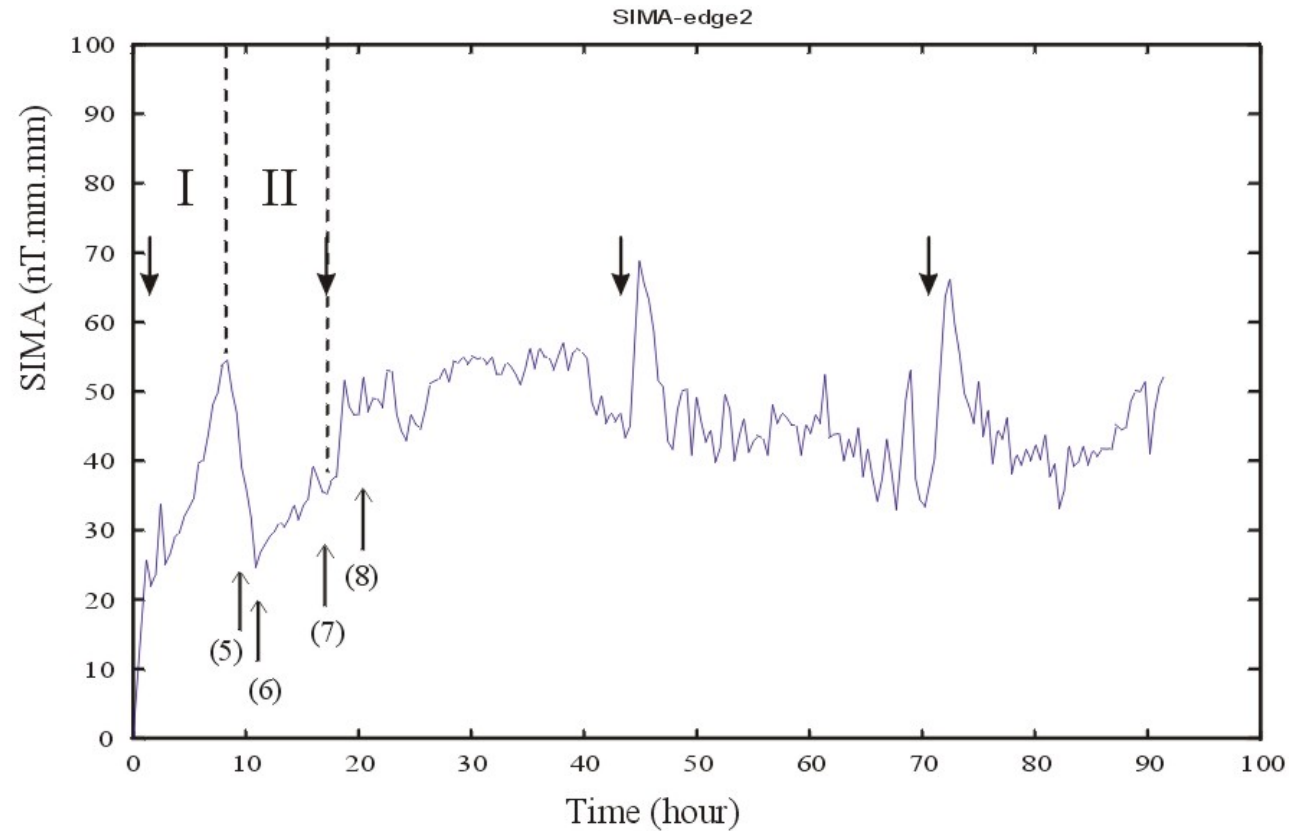


(4) 7.93 hour (#18)

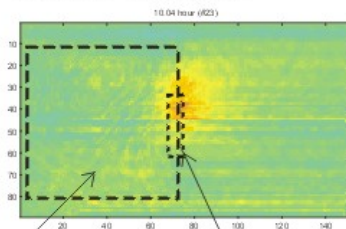


Upward arrows with (numbers) indicate times of each magnetic image

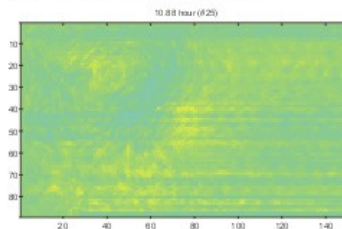
II. Developing --- changing polarity



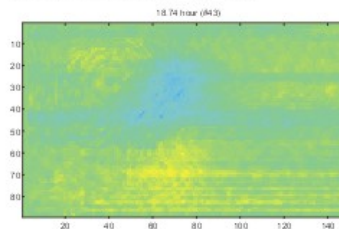
(5) 10.04 hour (#23)



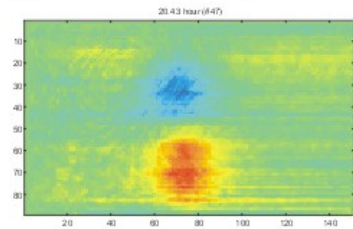
(6) 10.88 hour (#25)



(7) 18.74 hour (#43)

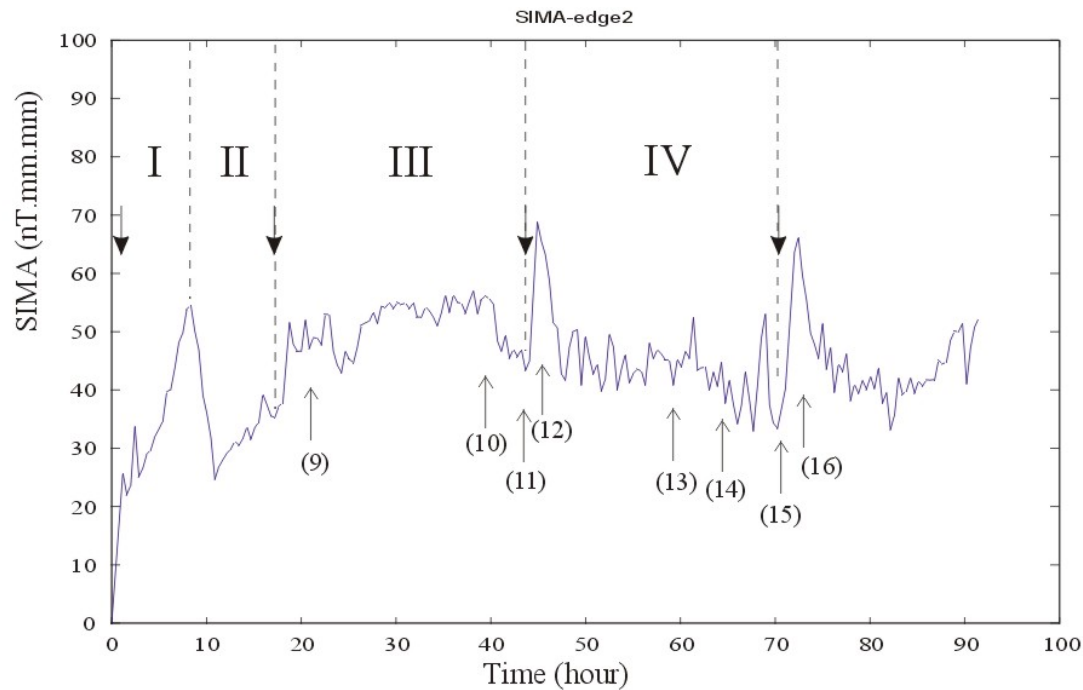


(8) 20.43 hour (#47)

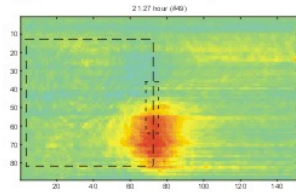


Upward arrows with (numbers) indicate times of each magnetic image

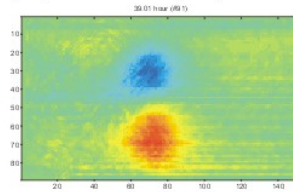
III & IV. Adding new solution accelerates corrosion



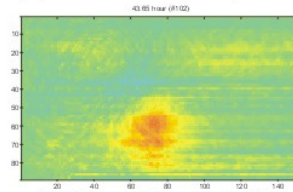
(9) 21.27 hour (#49)



(10) 39.01 hour (#91)

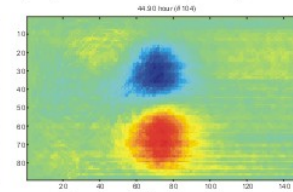


(11) 43.65 hour (#102)



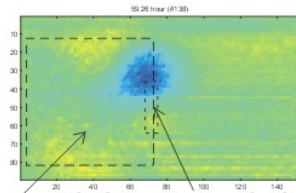
Before adding solution

(12) 44.90 hour (#104)



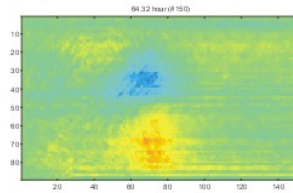
After adding solution

(13) 59.26 hour (#138)

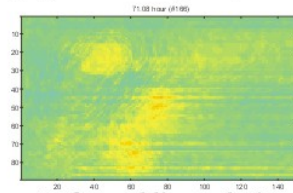


Al plate Exposed edge

(14) 64.32 hour (#150)

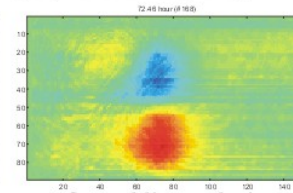


(15) 71.08 hour (#166)



Before adding solution

(16) 72.46 hour (#168)



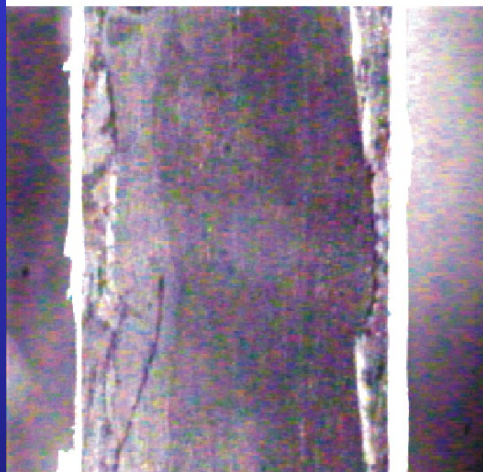
After adding solution

Upward arrows with (numbers) indicate times of each magnetic image

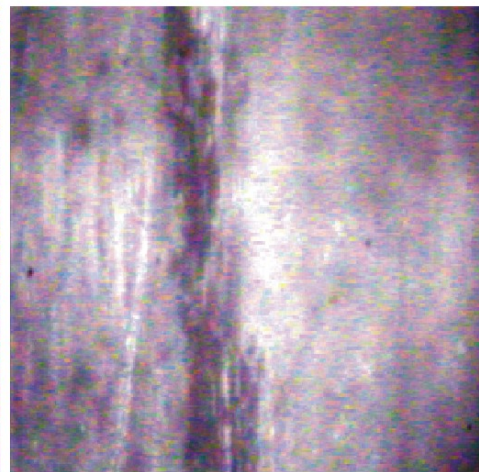


Microscopic Photo

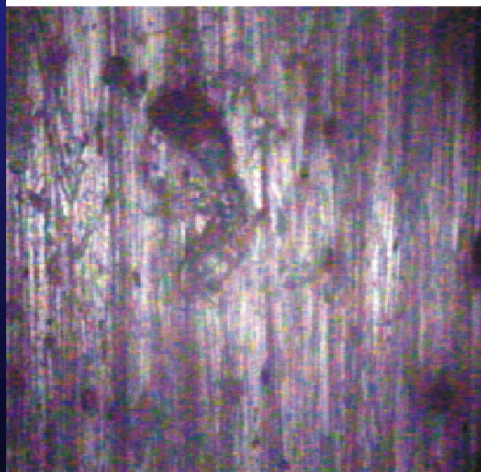
(a) Surface of the edge



(b) Possible exfoliation

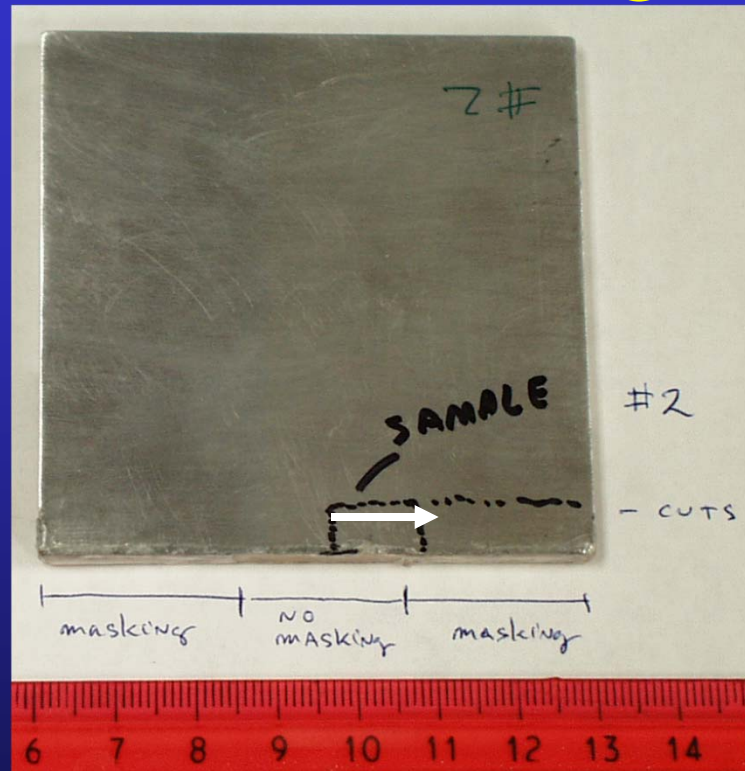


(c) Pitting corrosion





Metallographic Examination



The exposed edge (non-masked) was the S-T plane.



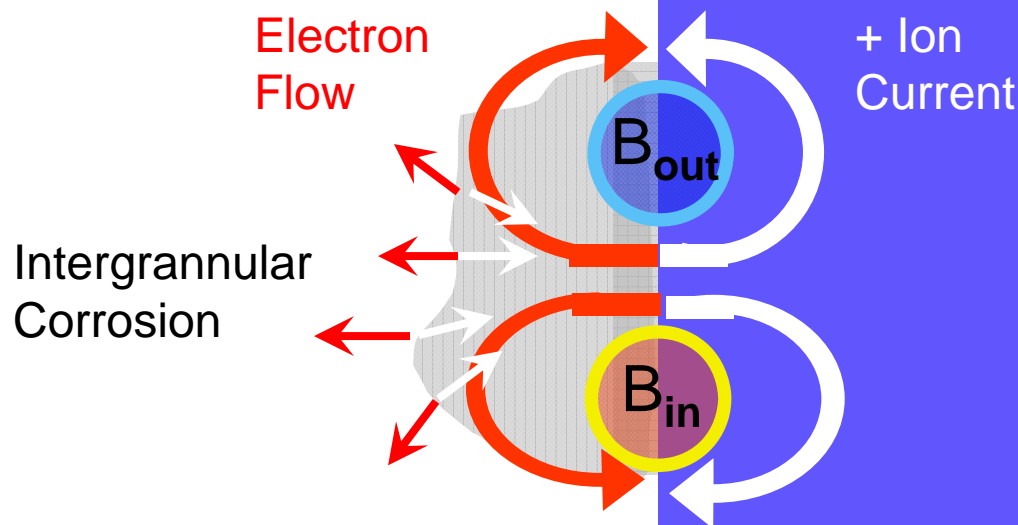
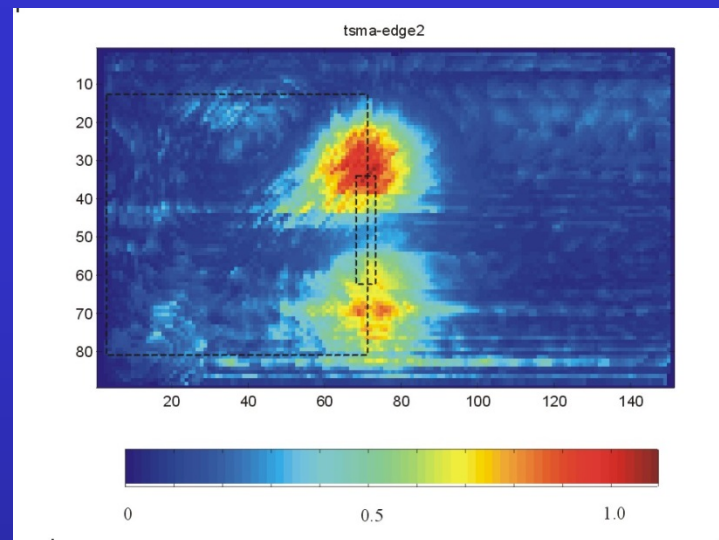
Arrow indicates direction of observation for metallography samples.

There is not a significant amount of attack of either of the two specimens. Low magnification visual observations are suggestive of only slight surface attack of the exposed region (*i.e.*, non-epoxyed area). Cross-sectional metallographic examination also did not reveal visible attack, despite successive grinding, polishing and examination. Neither exfoliation nor intergranular corrosion was observed. Samples were wet polished to 1200 grit.



TSMA for Al 7075

Current Flow in Exfoliation Test E2??



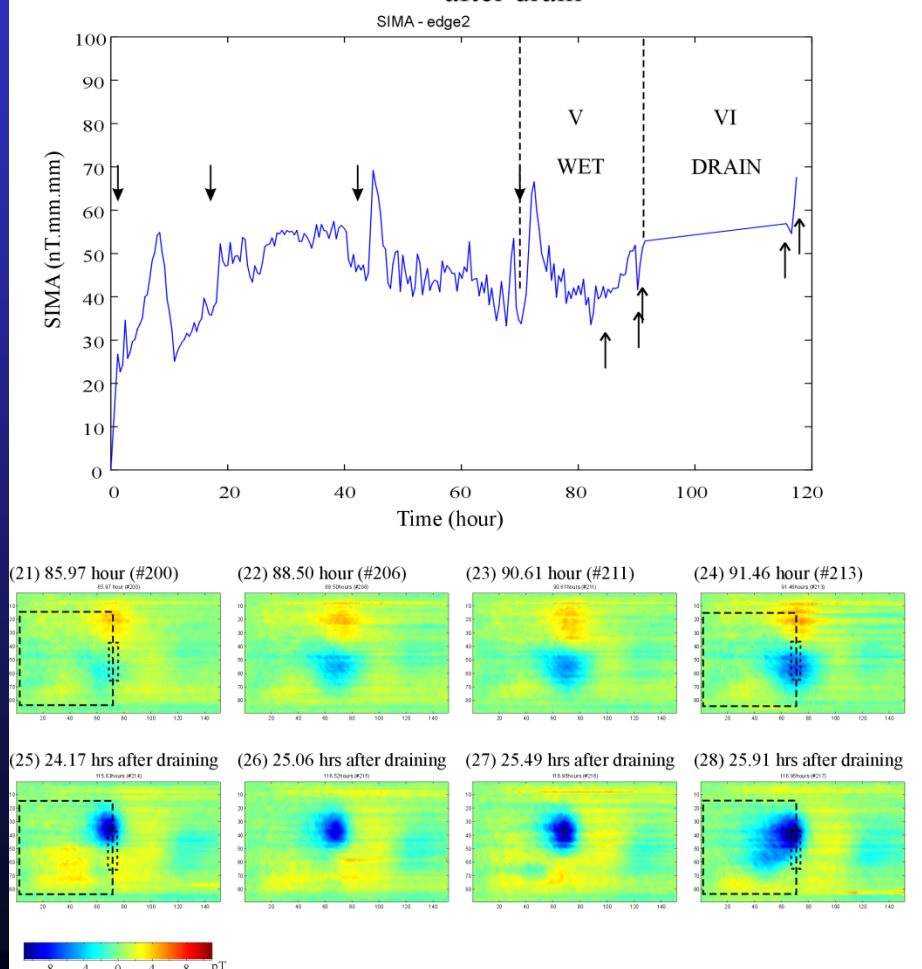
Aluminum

Electrolyte



Magnetic Activities After Draining

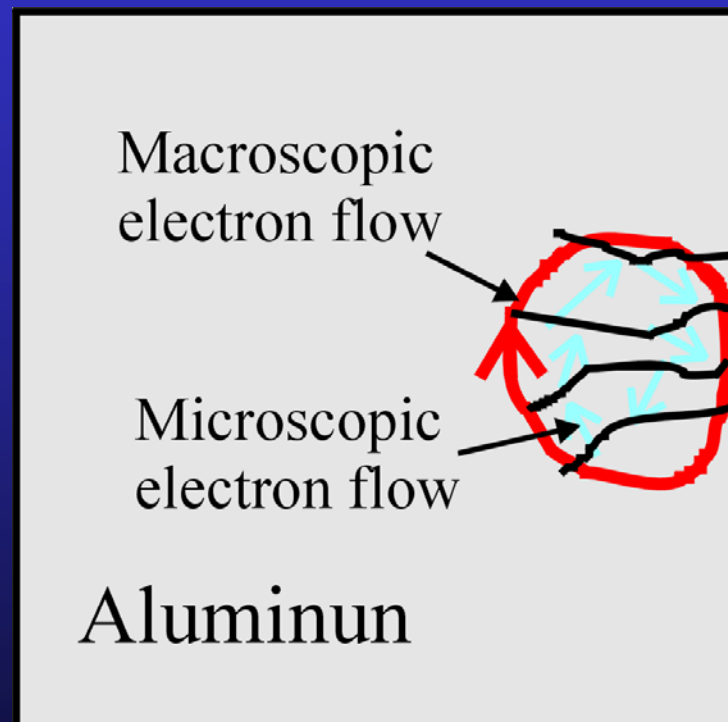
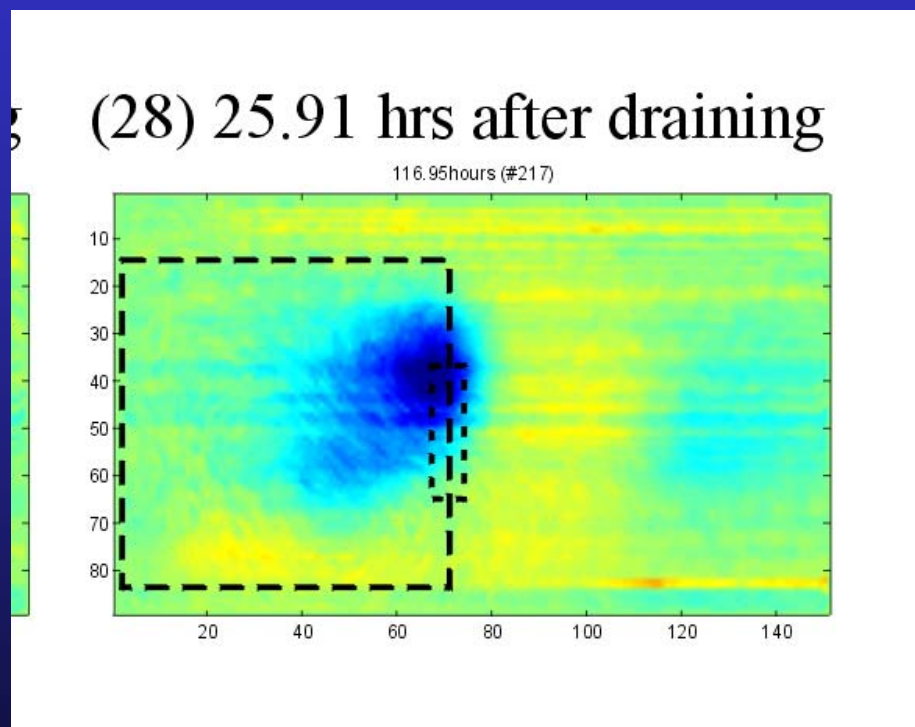
V & VI. Before and after drain: active exfoliation corrosion develops after drain



1. Before draining the corrosion at the side surface produce a dipolar magnetic signal.
2. 24 hours after draining the corrosion continues and the signal becomes a monopole.
3. The corrosion develops into the metal, even the cross-sectional metallographic examination did not reveal visible attack.



Possible Intergranular Corrosion Activity After Draining



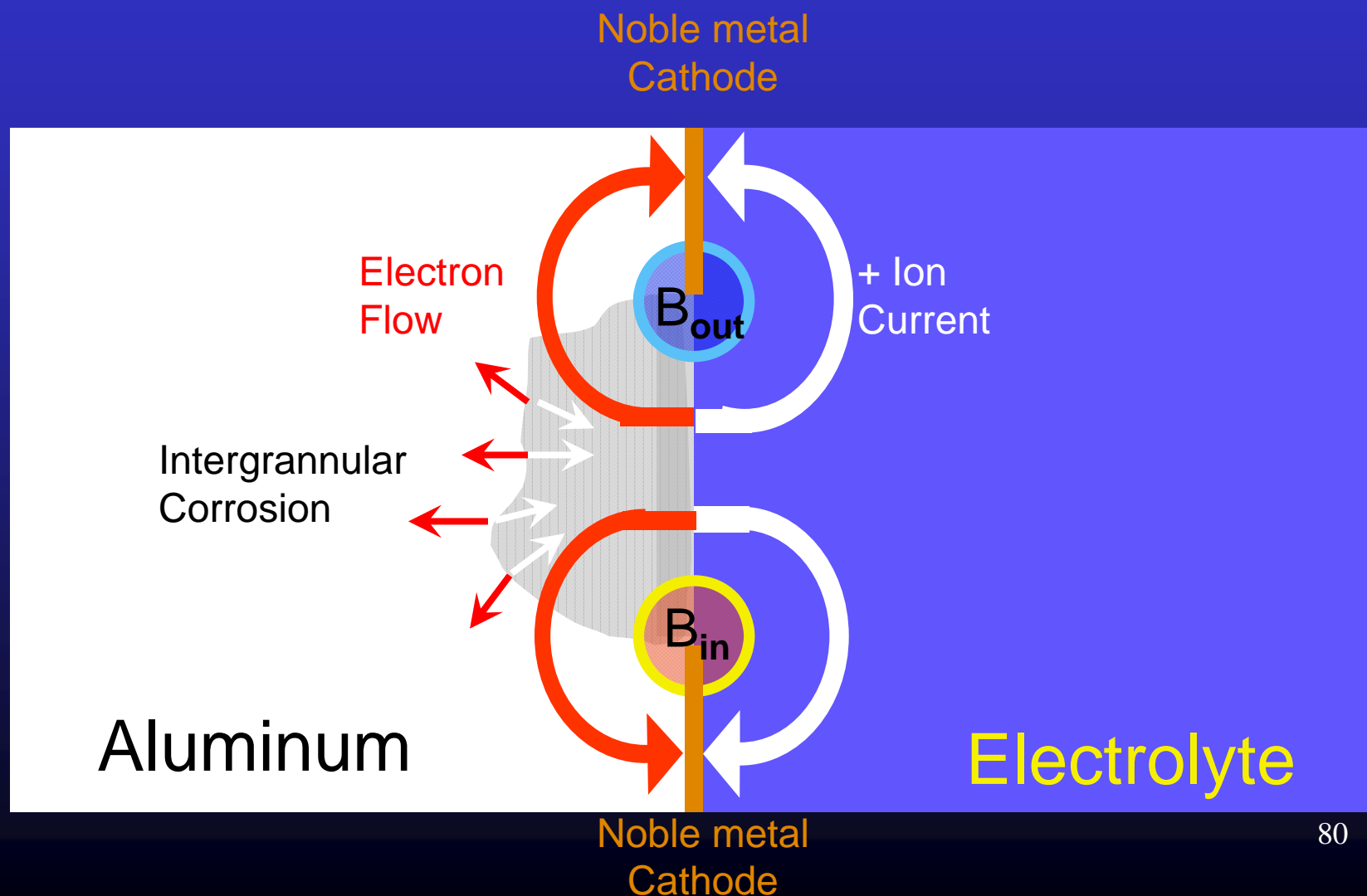


Protocol E2 Conclusions

- Edge-exposed square sample with square fluid reservoir
 - Simple geometry will allow quantitative analysis
 - Readily accessible corrosion face for damage characterization
 - Epoxy-Mylar coating more effective than red paint
 - Mechanically robust
 - Blocks corrosion
 - Can be removed chemically
- Distinct magnetic signature from corrosion
 - Field distribution correlates with exposed corrosion edge
 - Temporal fluctuations in activity correlate with addition of solution
 - Corrosion activity reaches steady state in approximately 24 hours
 - Ideal for tracking response to environmental change
- **Neither exfoliation nor intergranular corrosion was observed**
- **Kelly 2, SQUID 4**



Proposed SQUID Exfoliation E3 Geometry





There are, however, some loose ends..





$1 \text{ cm}^2 \text{ KC-135}$
 $4 \text{ mils of metal lost}$
 10^{-2} cm^3
 2.7 gm/cm^3
 27 gm/mole
 $10^{-3} \text{ mole} \Rightarrow I =$
 $\frac{3 \times 10^{-3} \times 10^{-3} \times 10^{-3} + 1.6 \times 10^{-19}}{80 \text{ years}} = 100 \text{ nA}$
 $3 \times 10^7 \text{ s/yr}$
 $B = \frac{\mu_0 I}{2\pi r}$
 $= 20 \text{ pT}$
 $200:1 \text{ SNR} \leftarrow !$

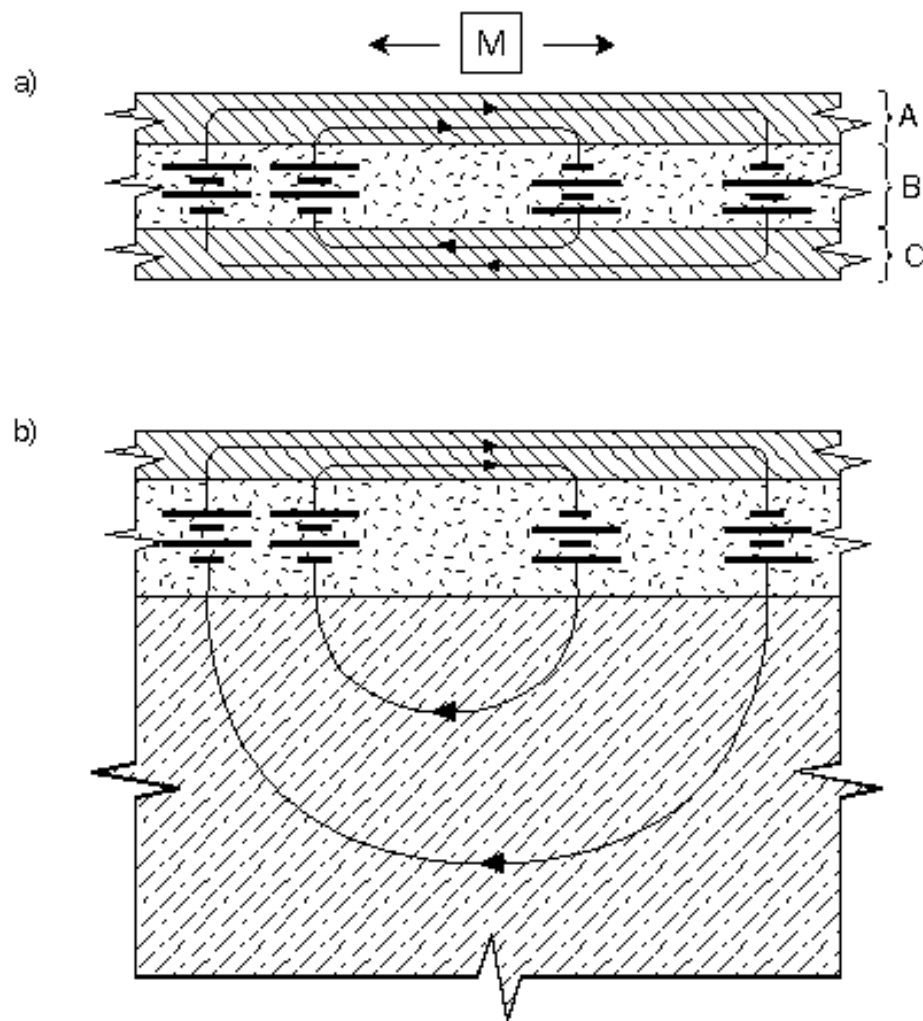
The measured magnetic fields are one to two orders of magnitude smaller due to cancellation of fields from small-scale corrosion circuits.

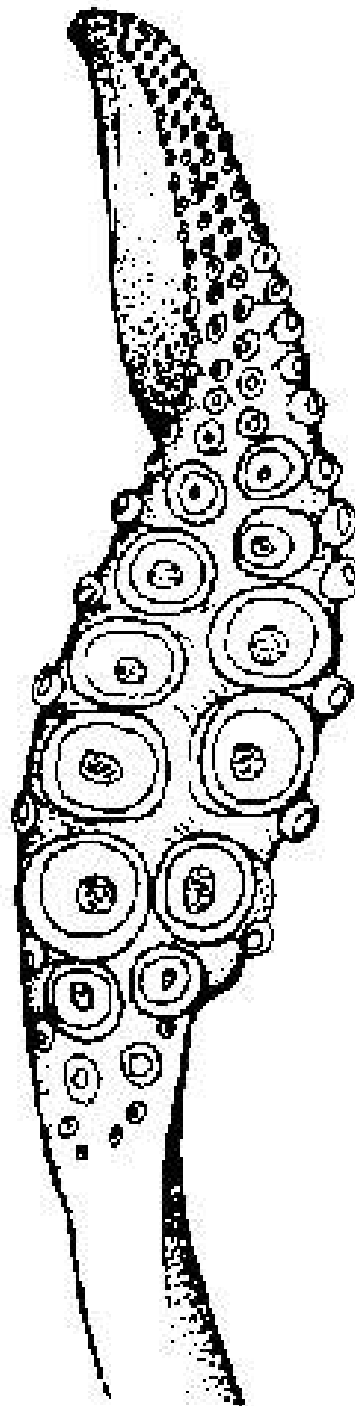


Problem

- A zeroth-order model of corrosion has a thin electrochemical layer between two metallic sheets, with the electrochemical current perpendicular to the layer

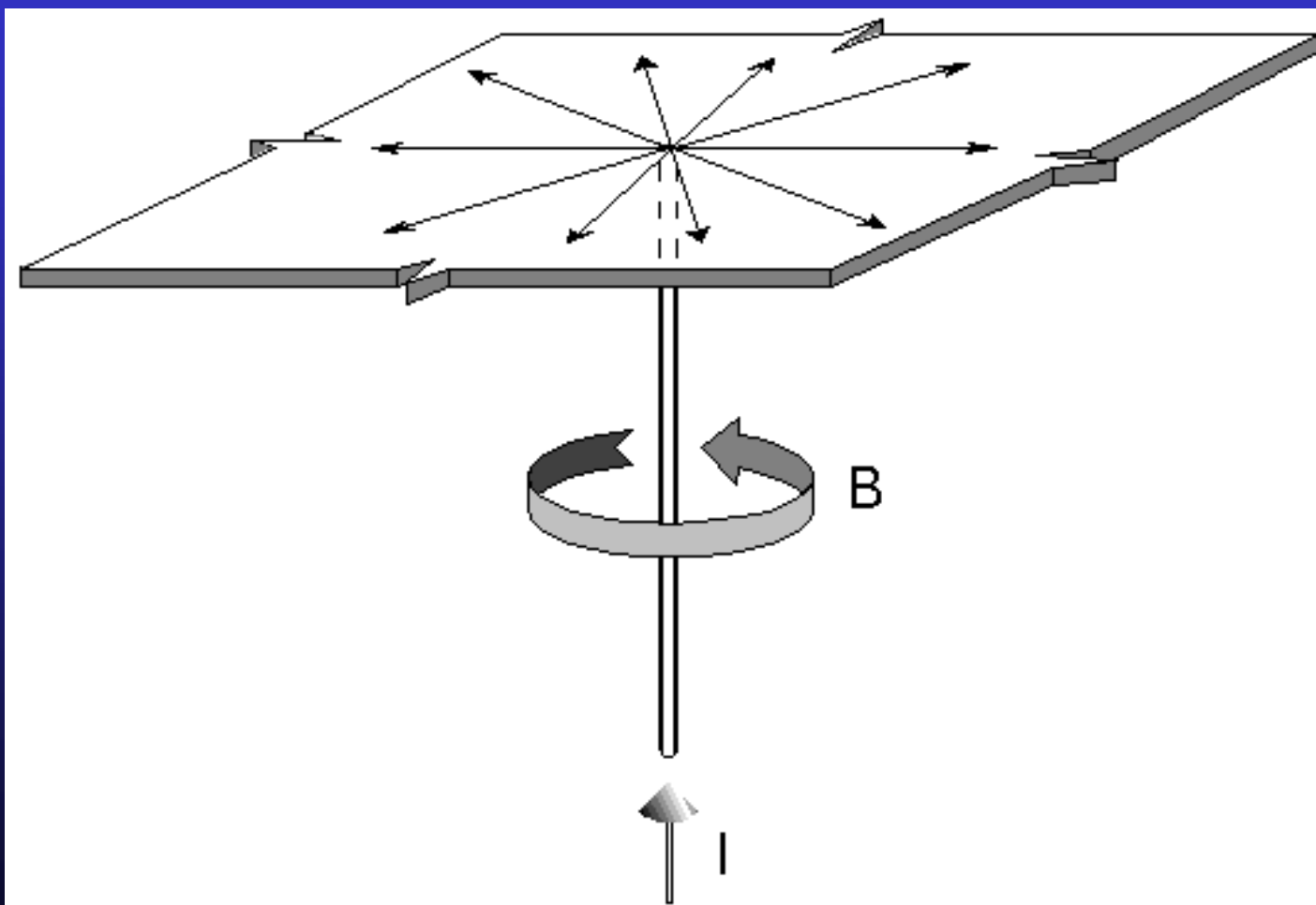
THE SIMPLEST MODEL





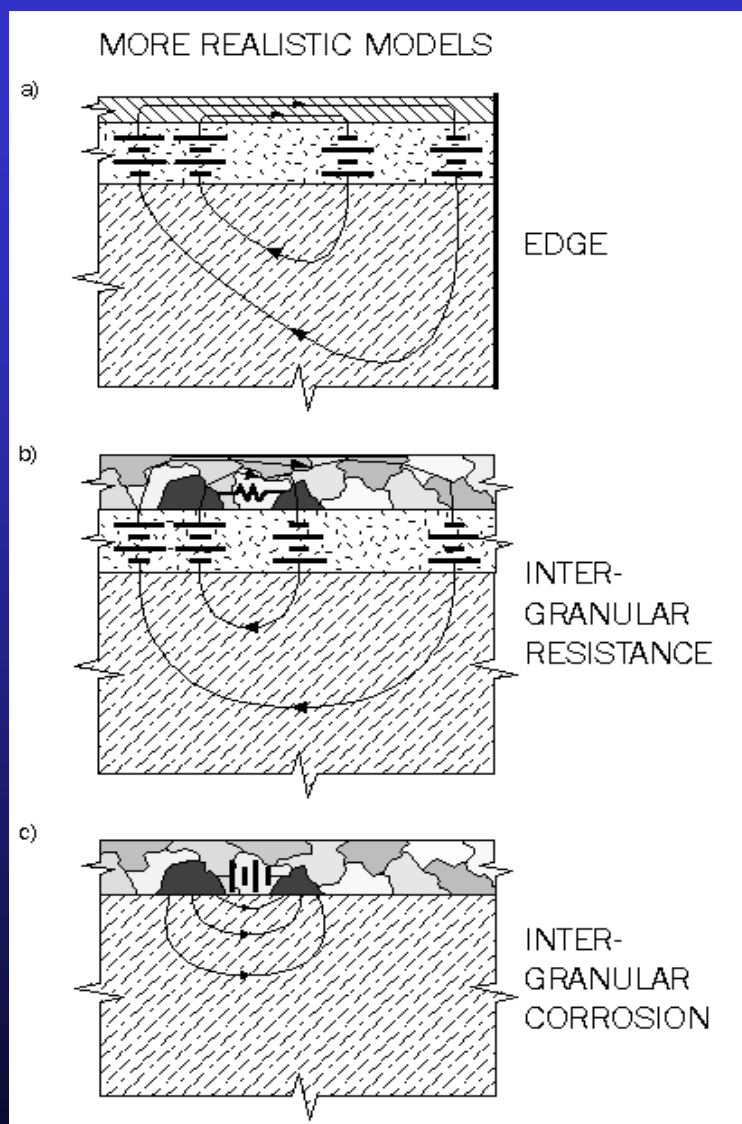


Simple vector calculus shows...





Three models for corrosion currents that produce magnetic fields



A very small demonstration...



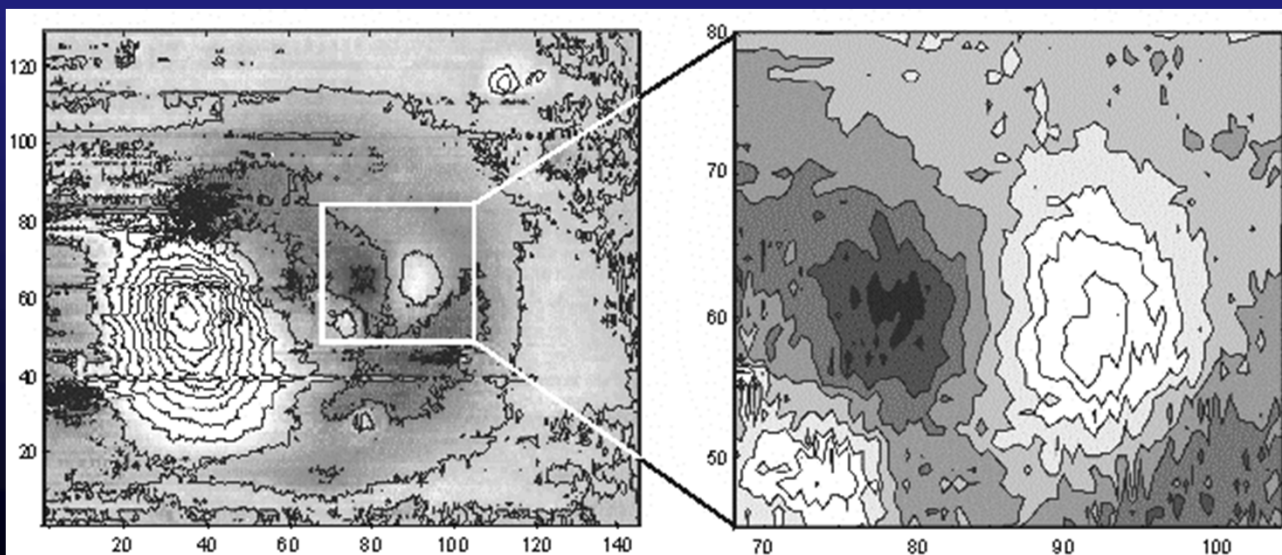
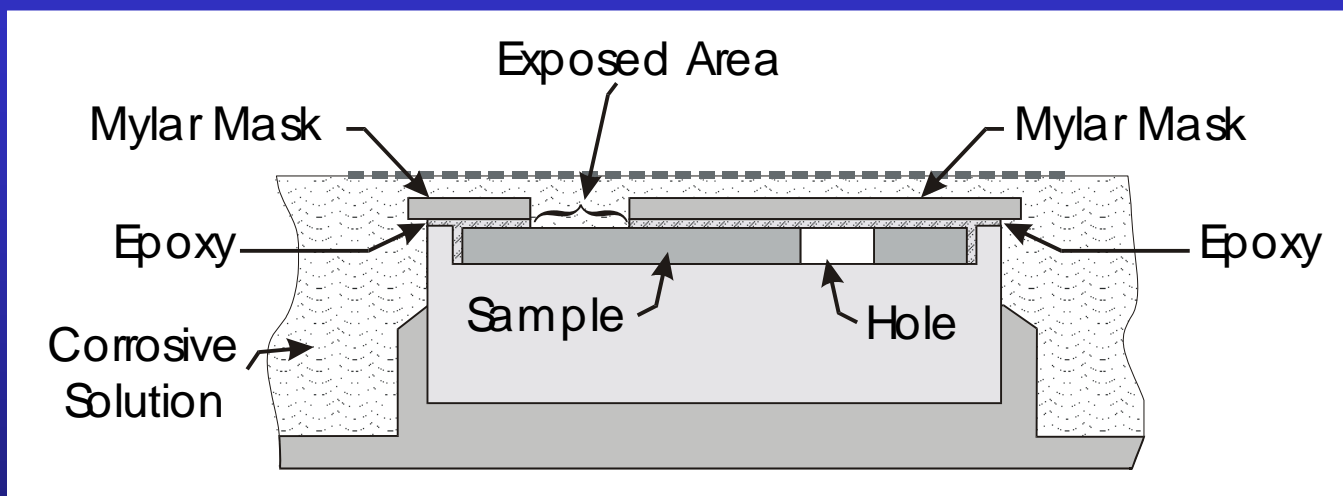


Uniform Corrosion Test, Insulated, Off-Center Hole, E26-S39





SQUIDs see current flowing through the metal, and are sensitive to when the current is deflected by internal conducting boundaries



Effects of Geometry on Magnetic Signals Due to Corrosion



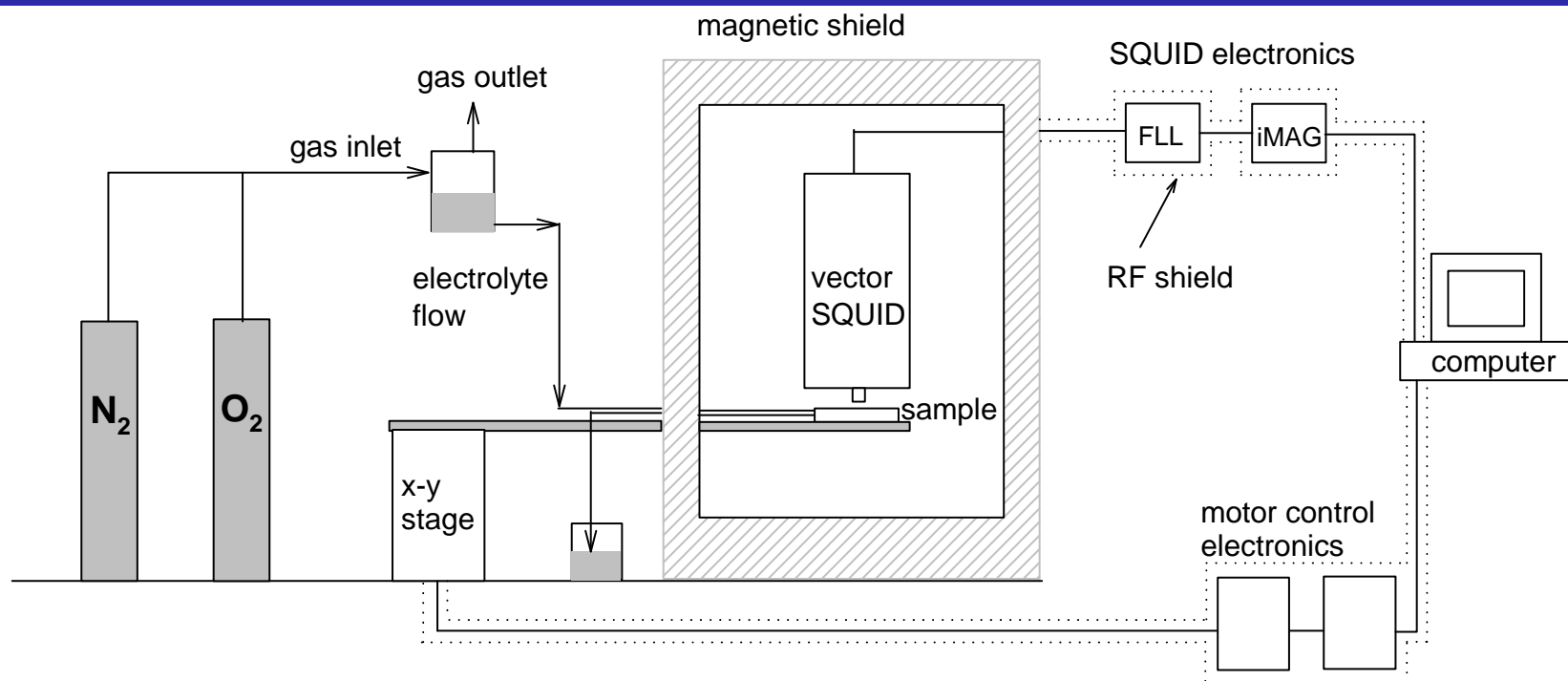


Effects of Geometry on Magnetic Signals Due to Corrosion

Eimutis Juzeliunas,
John P. Wikswo, and Yu Pei Ma
Institute of Chemistry, Lithuania
Vanderbilt University



The experimental setup of the magnetic imaging system for static or electrolyte flow conditions in 2024 Al





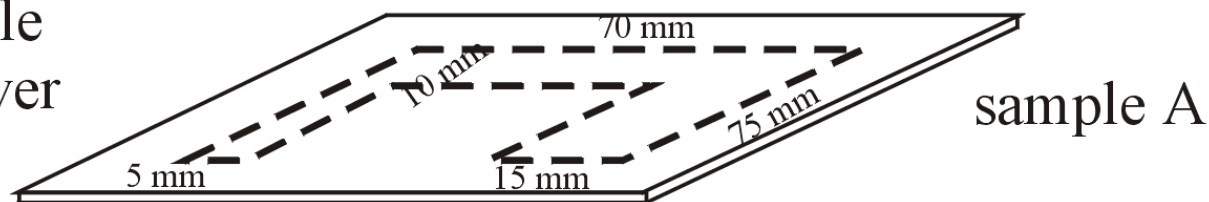
Experimental Set Up



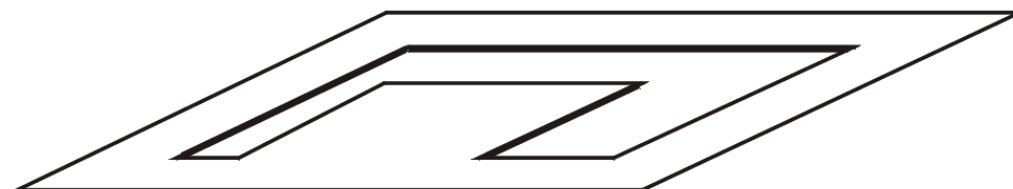


Measurement Cell

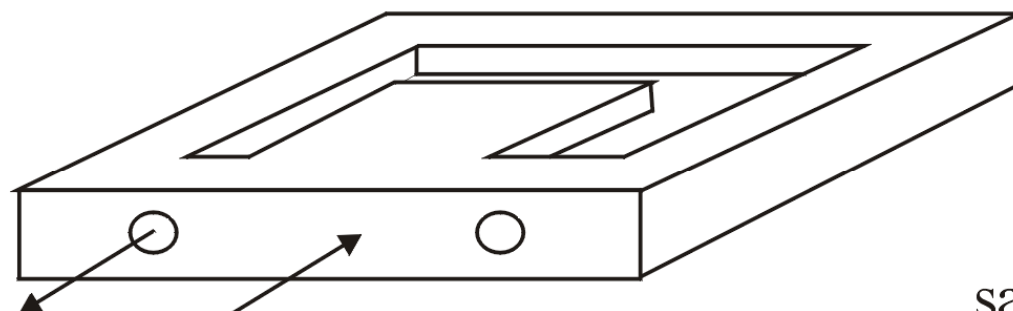
AA 2024 sample
under G-10 cover



rubber seal



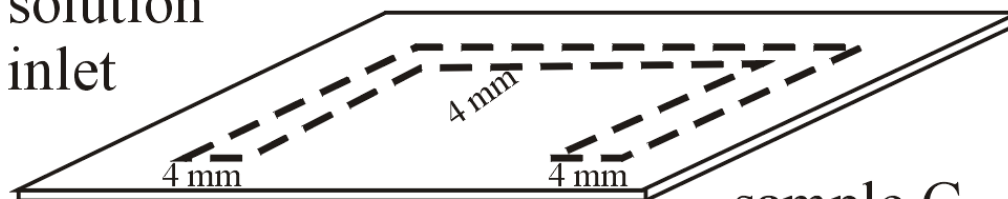
corrosion cell
(plexiglass)



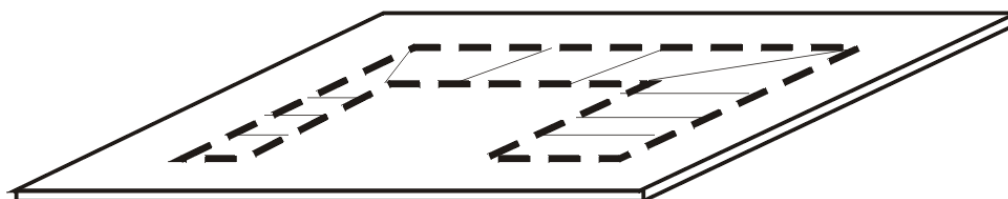
solution
outlet

solution
inlet

sample B

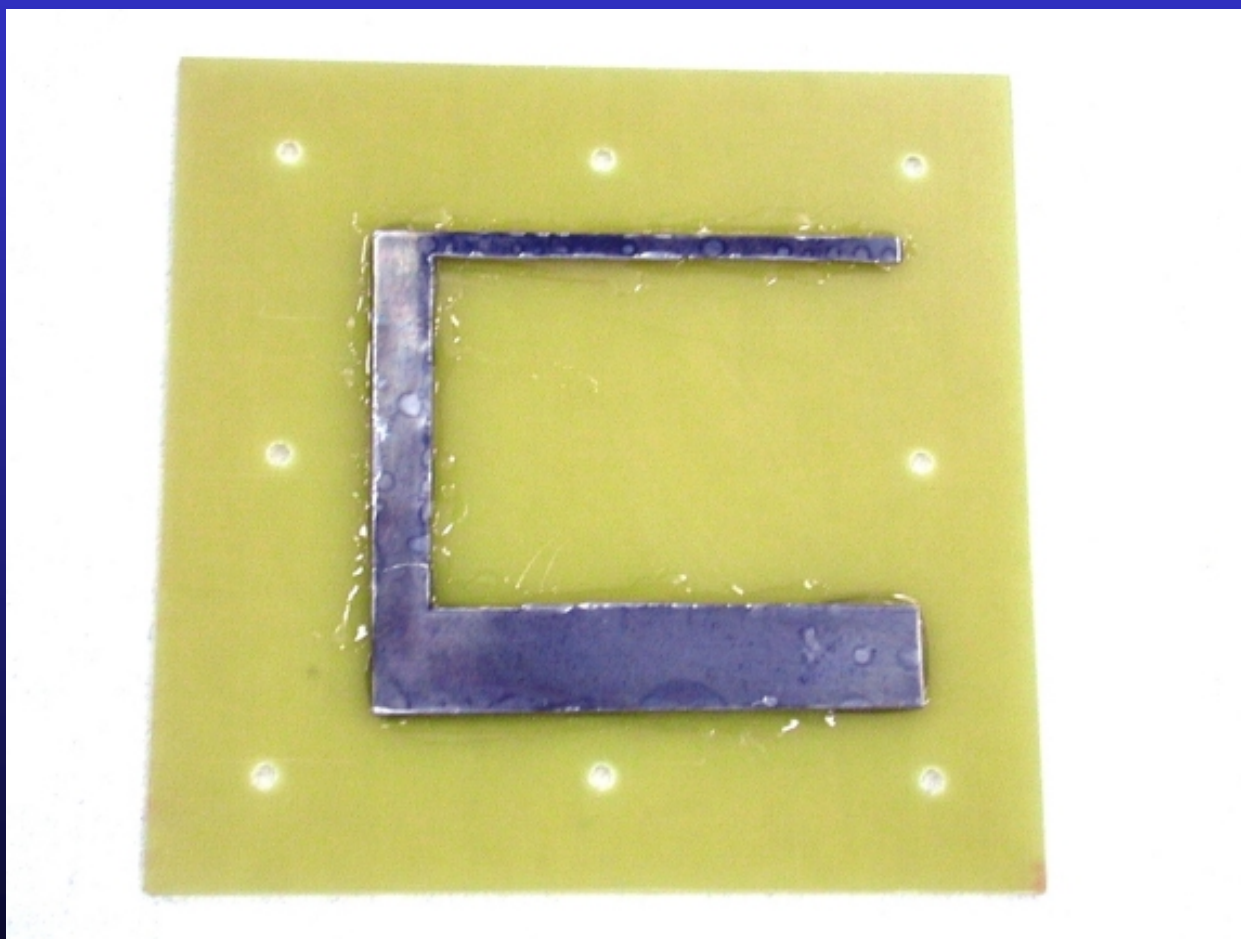


sample C





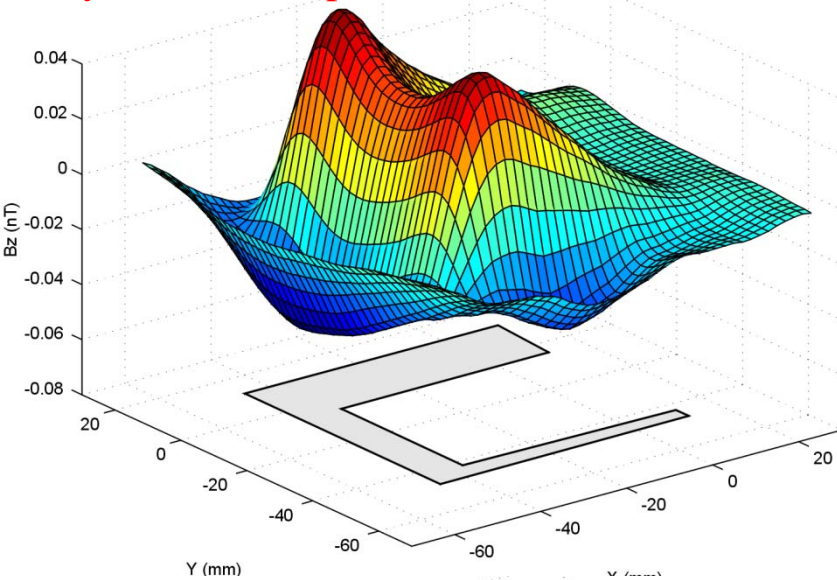
Sample



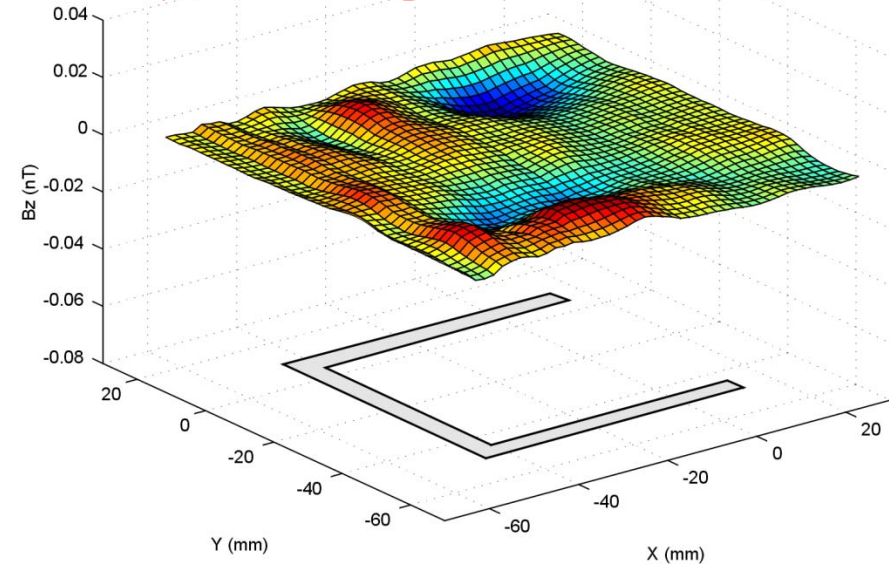
AA2024 sample in naturally aerated 3.5% NaCl solution

Four Measurements

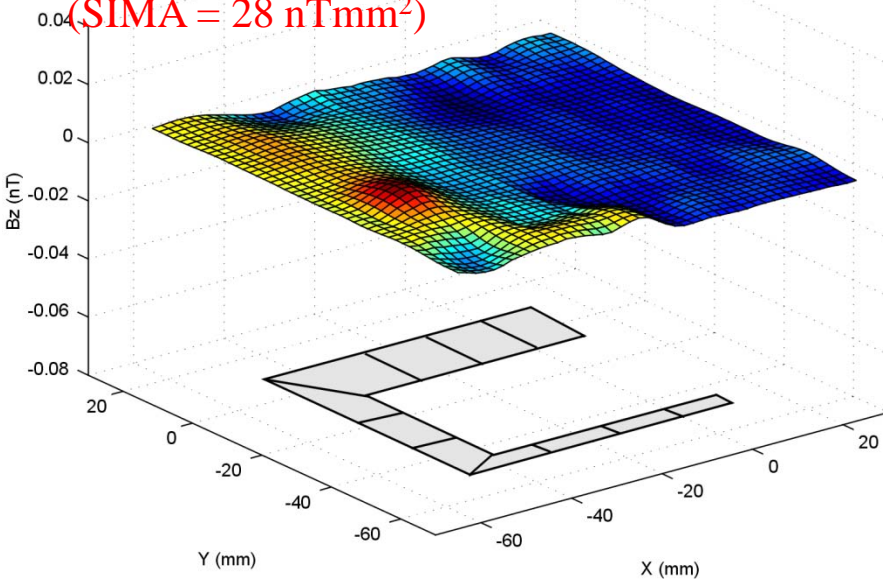
Asymmetric sample A ($\text{SIMA} = 147 \text{ nTmm}^2$)



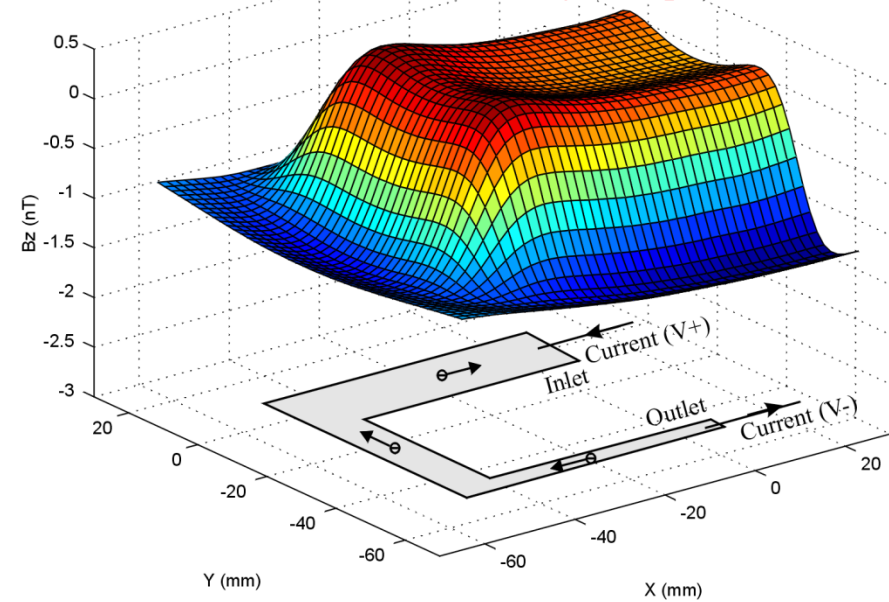
Symmetric sample B ($\text{SIMA} = 34 \text{ nTmm}^2$)



Sectioned asymmetric sample C
($\text{SIMA} = 28 \text{ nTmm}^2$)



0.1 mA current through sample A



Geometry Conclusions

- Asymmetric sample A produces a larger magnetic signal than does symmetric sample B. The ratio is 147/34 (4/1) .
- Due to the dish geometry, the magnetic signal is mainly due to the electron flow inside the AA 2024 sample.
- The corrosion reaction causes a potential difference inside the Al sample associated with the electron flow inside the sample.
- The sectioned sample shows localized current flow which produces much less magnetic signal
- ?? Does the corrosion rate depend upon geometry? Mass loss measurements are required!

Summing it up...



Ron O'Dor

What have we learned?

- We can image the magnetic fields from hidden corrosion
- Since we are measuring current, we have an instantaneous indication of corrosion activity
- Magnetic fields can detect corrosion currents associated with nA/cm voltage gradients in metals
- We are very sensitive to metal/insulator edges
- For a given geometry, the summed magnetic activity is proportional to mass loss
- There are cancellation and symmetry effects that deserve careful attention in experimental design and data analysis/interpretation

Future Studies

- Noble metal cathode to drive exfoliation/intergranular corrosion
- Correlate SQUID data with corrosion damage
- Examine factors that affect intergranular corrosion rate
 - Solution chemistry
 - Corrosion prevention compounds
 - Alloy preparation
 - Sample thickness and rolling direction
 - Temperature
- Examine samples with long-term corrosion
 - Signals from deep penetration
 - Dependence on deep corrosion rate on external environment and baking
 - Spatial correlation between TSMA and extended corrosion damage
- Current imaging instead of TSMA
- Higher spatial resolution SQUID images with SQUID microscope



Future Studies, Con't

- Magnetic fields from streaming currents/potentials
- Dependence of corrosion on oxygen and flow
- Biofilms
- Corrosion inside copper pipes



Acknowledgements

Support

- AFOSR URI (Harold Weinstock)
- AFCPO (Dick Kinzie)
- AFRL (Deb Peeler)
- NCI (Garth Brooks)
- S&KT (John Wensits)

Collaborators

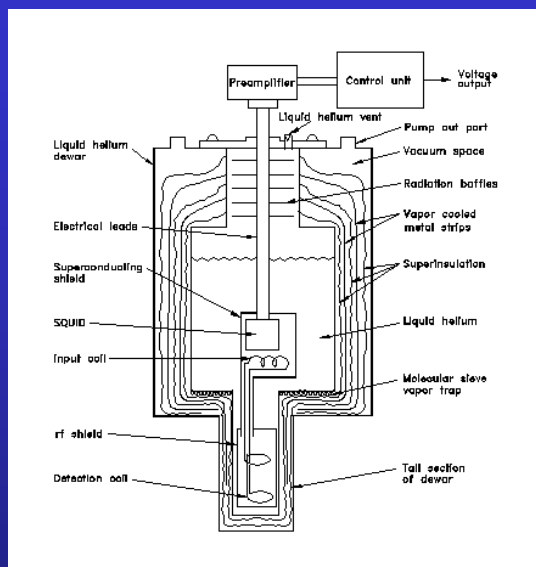
- Rob Kelly (UVa)
- Jim Suzel (NCI & S&KT)
- Kevin Cooper (Luna Innovations)
- Eimutis Juzeliunas (Institute of Chemistry, Vilnius, Lithuania)
- **Yu Pei Ma**, Afshin Abedi, Grant Skennerton, Delin Li (Vanderbilt)



M.-A. Bray

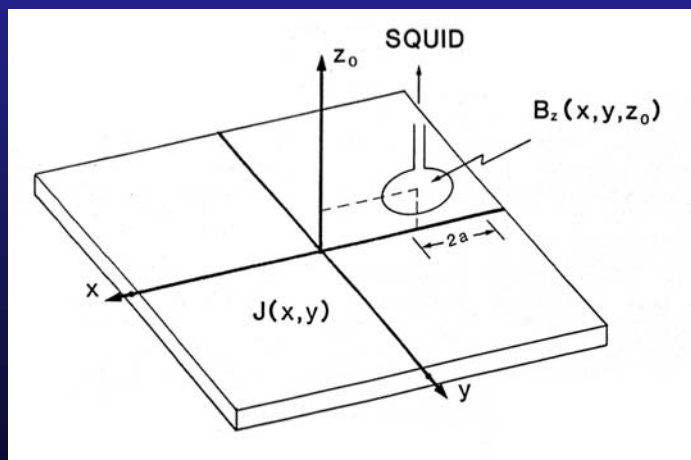
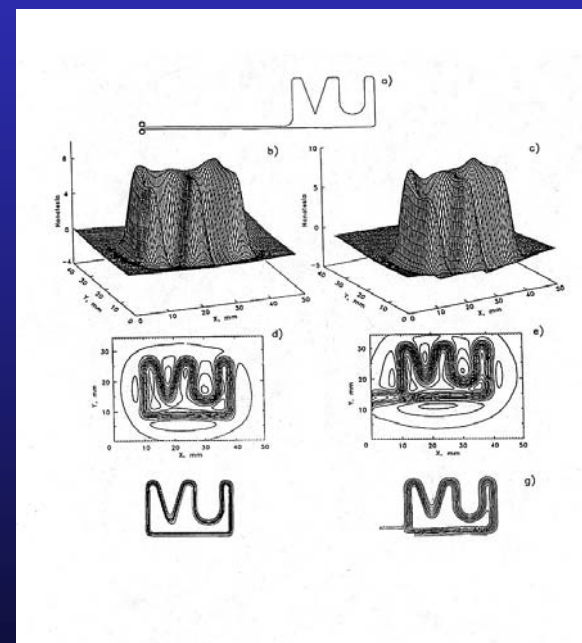
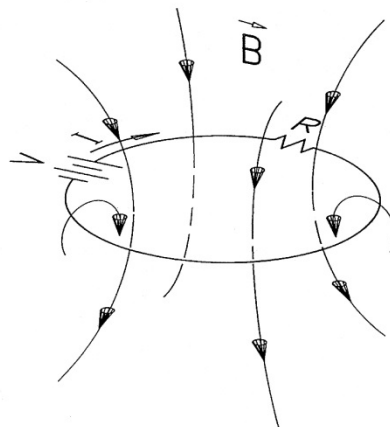


The Mathematics of Magnetic Imaging



The Law of Biot and Savart

$$\vec{B}(\vec{r}) = \frac{\mu_0 I}{4\pi} \oint \frac{d\vec{\ell}' \times (\vec{r} - \vec{r}')}{|\vec{r} - \vec{r}'|^3}$$





SPATIAL FILTERING AND CURRENT IMAGING THEORETICAL APPROACH

