

Chemistry 236 -- Practice Quiz 5
October 8, 2003 — Thermal Expansivity

1. A hypothetical gas follows the equation of state, $PV = nRT (1 + aP)$, where a is a constant. What is for this gas?
 - $nR (1/P + a)$
 - $\frac{nR}{V} (1 + a)$
 - $\frac{nR}{PV} (1 + aP)$
 - $1/T$
 - none of these
 - There are multiple correct answers.
2. The volume of a certain fluid is observed to increase by 1.00% when it is heated from 35.0°C to 43.0°C. What is its average over this temperature range?
 - 0.00125 K
 - 0.0100 K⁻¹
 - 0.125 K⁻¹
 - none of these
 - This cannot be determined without additional information.
3. What would be the best temperature to record for the estimate of in the previous problem?
 - 35°C
 - 39°C
 - 40°C
 - 43°C
 - none of these
4. A 1.00-cm³ cube of titanium ($\rho = 4.51 \text{ g/cm}^3$) is suspended in water at 20.0°C from a sensitive spring scale. What does it "weigh"?
 - 3.51 g
 - 4.51 g
 - 5.51g
 - none of these
5. The weighing is repeated using an analytical balance equipped with brass weights ($\rho = 8.4 \text{ g/cm}^3$), and the entire balance is immersed in a tub of water at 20.0°C. What does the Ti weigh now?
 - 3.51 g
 - 4.51 g
 - 5.51g
 - none of these
6. If $x \ll 1$, what is the approximate value of $(1 + x)^5 - 1$?
 - x^5
 - $5 x^5$
 - $3 x$
 - none of these