

Chemistry 236 -- Practice Quiz 5  
October 8, 2003 — Thermal Expansivity

- A hypothetical gas follows the equation of state,  $PV = nRT (1 + aP)$ , where  $a$  is a constant. What is for this gas?
  - $nR (1/P + a)$
  - $\frac{nR}{V} (1 + a)$
  - $\frac{nR}{PV} (1 + aP)$
  - $1/T$
  - none of these
  - There are multiple correct answers.
- The volume of a certain fluid is observed to increase by 1.00% when it is heated from 35.0°C to 43.0°C. What is its average over this temperature range?
  - 0.00125 K
  - 0.0100 K<sup>-1</sup>
  - 0.125 K<sup>-1</sup>
  - none of these
  - This cannot be determined without additional information.
- What would be the best temperature to record for the estimate of in the previous problem?
  - 35°C
  - 39°C
  - 40°C
  - 43°C
  - none of these
- A 1.00-cm<sup>3</sup> cube of titanium ( $\rho = 4.51 \text{ g/cm}^3$ ) is suspended in water at 20.0°C from a sensitive spring scale. What does it "weigh"?
  - 3.51 g
  - 4.51 g
  - 5.51g
  - none of these
- The weighing is repeated using an analytical balance equipped with brass weights ( $\rho = 8.4 \text{ g/cm}^3$ ), and the entire balance is immersed in a tub of water at 20.0°C. What does the Ti weigh now?
  - 3.51 g
  - 4.51 g
  - 5.51g
  - none of these
- If  $x \ll 1$ , what is the approximate value of  $(1 + x)^5 - 1$ ?
  - $x^5$
  - $5 x^5$
  - $3 x$
  - none of these