

**Chemistry 236**  
**Freezing Point Depression Study Problems -- Answers**

1. d
2. c
3. a
4. b
5. c
6. e (0.625 M, since H<sub>2</sub>SO<sub>4</sub> has two neutralizable H's)
7. d
8. a
9. d
10. b
13.  $\gamma = 0.217$  and  $T = -0.451$  K.
14. (a) Expected  $T$  (simple treatment) =  $-1.030$  K.  
(b)  $\gamma = 0.9599$  (eq 7);  $a_A = 0.99047$  (eq 6);  $\Delta T = 1.00035$  (eq 4).
15. molality of HA =  $0.245 \text{ mol kg}^{-1}$ .