

Chemistry 236 -- Quiz 1

January 31, 2012 — Statistics and KaleidaGraph Basics

Pledge and signature:

Note: If you want your paper returned folded (i.e., score concealed), please print your name on the back.

1. (4) State each of the following numbers with unambiguous precision and with the minimal number of digits needed to achieve 0.3 % precision: (a) 1.2345 (b) 8767500

2. (5) Marge Inovera measures a quantity 25 times and obtains an average and a sum of squared residuals. If the latter is 189.155,
 - a. Give Marge's estimated variance, standard deviation, and standard deviation in the mean. (Give precision commensurate with the provided information.)

 - b. Marge also finds that the sum of her measured values is 204.715. Using the 10% rule, properly state her average and its uncertainty.

3. (2) If y is uncertain by 3.2%, what is the uncertainty in $z = \ln(3y)$?

4. (3) A quantity x is uncertain by 3.0% and y is uncertain by 1.0%. Give the % uncertainties for z in each of the following cases:
 - a. $z = 5/y$
 - c. $z = 5 x/y^2$

 - b. $z = 11x^{1/3}$

